

## Original Research Article

# Preparation, characterization and antibacterial effects of eco-friendly gold nanorods

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### Abstract

**Purpose:** To synthesize and characterize eco-friendly gold nanorods (Au-NRs) and to assess their effects against two bacterial strains.

**Methods:** Synthesis of eco-friendly gold nanorods was done from an aqueous solution of chloroauric acid and cetyltrimethylammonium bromide by mixing *Olea europaea* fruit and *Acacia nilotica* husk extracts with the latter as a reducing agent. The synthesis was monitored by ultraviolet-visible (UV) spectrophotometry and a zetasizer, while the morphology of the resulting nanorods was assessed by transmission electron microscopy (TEM) and scanning electron microscopy (SEM) coupled with an energy-dispersive spectrophotometer (EDS). The effect of the prepared eco-nanorods on growth of *Escherichia coli* and *Staphylococcus aureus* bacteria were investigated by agar diffusion method.

**Results:** The synthesized Au-NRs were mostly rod-shaped, with mean size of 96 nm. They exhibited a potent antibacterial activity against Gram-positive and Gram-negative microorganisms (*Escherichia coli* and *Staphylococcus aureus*).

**Conclusion:** The findings suggest that the nanoformulation of the biomaterial mix of *Olea europaea* fruit and *Acacia nilotica* husk extracts is a cost-effective, eco-friendly, and non-toxic synthesis of Au-NRs which has therapeutic potential.

**Keywords:** *Olea europaea*, *Acacia nilotica*, Gold nanorods, Eco-friendly synthesis, Gram-positive and Gram-negative microorganisms

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## INTRODUCTION

Nanotechnology is a science that involves the manipulation of matter at the atomic and molecular levels. Nanobiotechnology is an integral and emerging tool for engineering eco-friendly nanoscale materials using natural biological sources [1]. The integration of inorganic nanoparticles (NPs) and biological structures is one of the most challenging areas of research. Biological synthesis elucidates the importance of metal-microbe interaction in

several biotechnological applications, including bioremediation, biomineralization, bioleaching, and microbial corrosion [2].

NPs are unique in having a high surface area-to-volume ratio and high percentage of atoms at the grain boundaries [3,4]. Moreover, biological processes do occur at the nanoscale, which accounts for the ease of application of NPs in the field of medicine because of their amenability to biological functionalization [5]. NPs are broadly classified as organic and inorganic NPs. The

latter have gained significant importance because of their ability to resist the processing conditions [6].

NPs are potent antibacterial and antifungal agents because of their ability to interact with microorganisms [7,8]. The antimicrobial activity of NPs is a function of the surface area in contact with the microbes. The small size and the high surface area-to-volume ratio (i.e., large surface area) of NPs helps them to interact well with microbes which affects their antimicrobial ability. The antimicrobial potential of metal-based NPs has led to their incorporation in consumer, health-related, and industrial products [9].

The synthesis and applications of stable gold nanoparticles (Au-NPs) have been a subject of great interest because of the remarkable physical and chemical properties of Au-NPs. Au-NPs are uniformly structured and are highly stable. In addition their size-related electronic, magnetic, and optical properties are judiciously exploited in fields such as catalysis, biosensing, drug delivery, energy-related applications, and biological applications [10,11,12].

Metal NPs that exhibit antimicrobial activity when embedded and coated onto surfaces are widely used in water treatment, synthetic textiles, biomedical and surgical devices, and food processing and packaging [13]. Furthermore, antibiotics often disturb the bacterial flora of the digestive tract, leading to the development of multiple drug-resistant isolates; hence, novel methods for formulating biocide materials is imperative [9,12,14,15]. Thus, there is a need for a novel bactericidal that offers least resistance coupled with good bactericidal properties. Au-NPs exhibit potent bactericidal effects on a wide range of microorganisms; however, the bactericidal effect is a function of the size and the shape of the structured nanoparticles [16].

With this premise, the present study aimed at the development of a novel method for the synthesis of gold nanorods (Au-NRs) by the reduction of aqueous  $\text{AuCl}_4$  and cetyltrimethylammonium bromide (CTAB), adopting an eco-friendly approach. The natural products used in the synthesis, we is a mix (1:3) of *Olea europaea* fruit extract and *Acacia nilotica* husk extract.

The antibacterial activity of Au-NRs against various strains of bacteria (*Escherichia coli*, *Staphylococcus aureus*) were tested. To our knowledge, method of preparation used in the study has not been reported in previous literature.

## EXPERIMENTAL

### Reagents and chemicals

For the synthesis of Au-NRs, all the chemicals used were of analytical grade. Chloroauric acid was purchased from Loba Chemie, India, and cetyltrimethylammonium bromide was purchased from BDH Laboratory supplies pool, BH151 TD, England. *Olea europaea* fruit and *Acacia nilotica* husk were obtained from Al-Jouf, Saudi Arabia and Sudan, respectively.

### Preparation of aqueous extract solution

A 1 % by volume (g/v) solution of *Olea europaea* fruit was washed well, ground in deionized water, and filtered. The extract was stored for later use. For *Acacia nilotica* husk extract, 1 % by volume was soaked in deionized water overnight and filtered, and the extract was stored for later use. A mix of the *O. europaea* fruit extract and the *A. nilotica* husk extract with a volume ratio of 1:3 constituted the final organic solution. The solution contained flavonoids, phenols, and oleic acid and pentacyclic triterpenoid compounds, as revealed by thin-layer chromatography.

### Preparation of Au-NRs

A novel approach was employed to synthesize eco-friendly Au-NRs. The nanorods (NRs) were synthesized by adding an aqueous solution comprising a mix (1:3) of 1 % *O. europaea* fruit extract and 1 % *A. nilotica* extract to an aqueous solution of 0.0009 mol/L colloidal chloroauric acid ( $\text{HAuCl}_4$ ) and 0.35 mol/L CTAB under vigorous stirring at a temperature of 45 °C for 13 min until an orange color was observed in the mixture. When the extract was added, the mixture changed from orange to transparent and then became colorless. The colorless endpoint indicated the formation of NRs. The presence of Au-NRs were confirmed by the color changes and qualitatively characterized using an ultraviolet-visible (UV-vis) spectrophotometer. The microstructure was observed using a transmission electron microscope (TEM) and a scanning electron microscope (SEM).

### Instruments and apparatus

The green synthesis of the Au-NRs was characterized spectrophotometrically. The UV-vis spectra of the synthesized NRs were recorded with respect to time at room temperature using a UV-vis spectrometer (Perkin Elmer). The size of the synthesized Au-NRs was determined using a Zetasizer system (Nano Series, HT Laser, ZEN3600, Malvern

Instruments, UK). SEM was employed to characterize the shape and morphologies of the ecofriendly synthesized NRs (JEOL-FE-SEM), and an energy-dispersive X-ray spectroscopy (EDS) analysis was performed for the confirmation of elemental gold using (INCA x-act, Oxford Instruments) equipped for SEM. A transmission electron microscope (JEM-1011, JEOL, Japan) was employed to characterize the size, shape, and morphologies of the ecofriendly synthesized the Au-NRs.

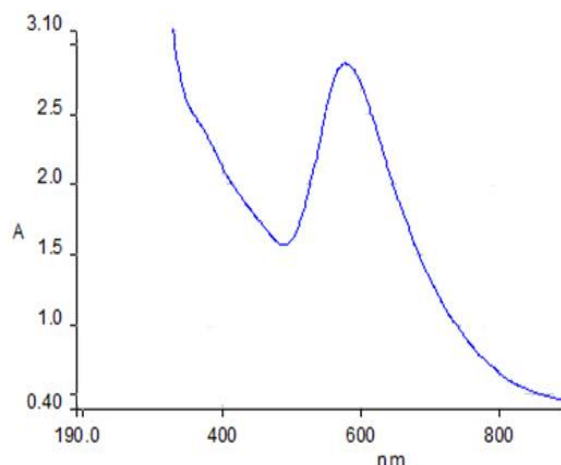
### Antibacterial activity assay

Pure cultures of *Escherichia coli* and *Staphylococcus aureus* bacteria were used. The antibacterial activity of the biosynthesized Au-NRs were evaluated by the disk diffusion method. Nutrient agar medium plates were prepared, sterilized, and solidified. Thereafter, the bacterial cultures were swabbed on the plates. The sterile discs were dipped in a Au-NR solution (5 or 10 µg/ml), placed on the nutrient agar plate, and incubated at 37 °C for 24 h. Upon inhibitory activity, a zone of clearing around the wells was observed. The diameter of the clearing zones was measured in millimeters using a ruler. The experiment was repeated three times, and the mean value of the zone diameter was recorded [17].

## RESULTS

In our novel approach, green synthesis of Au-NRs using CTAB as a surfactant and a mix (1:3) of *Olea europaea* fruit extract and *Acacia nilotica* husk extract was performed. The consistency of Au-NRs was successful, as indicated by the initial changes in color, and the Au-NRs exhibited a plasmon peak in the visible region (Figure 1).

An absorbance peak at ~577 nm was observed in the spectrum.



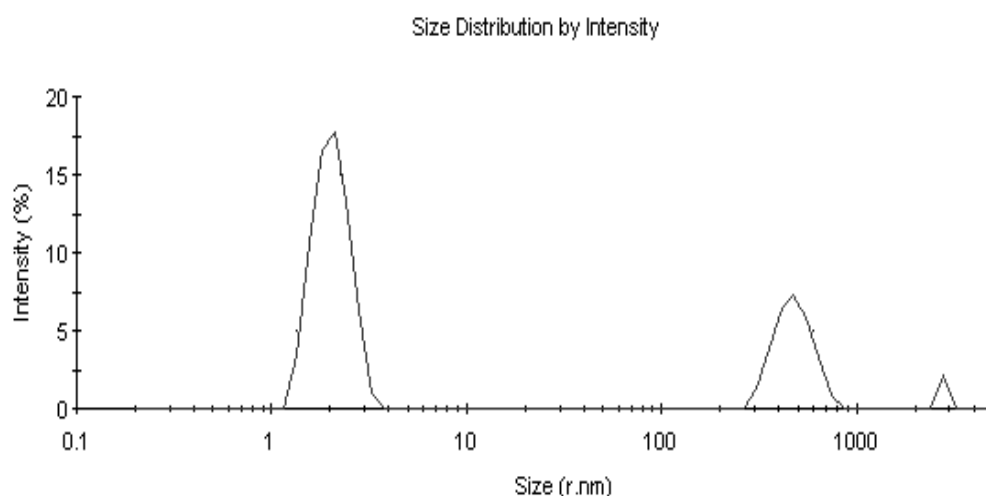
**Figure 1:** UV-vis spectra of Au-NRs synthesized with 0.09 mol/L HAuCl<sub>4</sub>

The mean size of the synthesized AuNRs was measured using a zetasizer system and was ~96 nm, as shown in Figure 2.

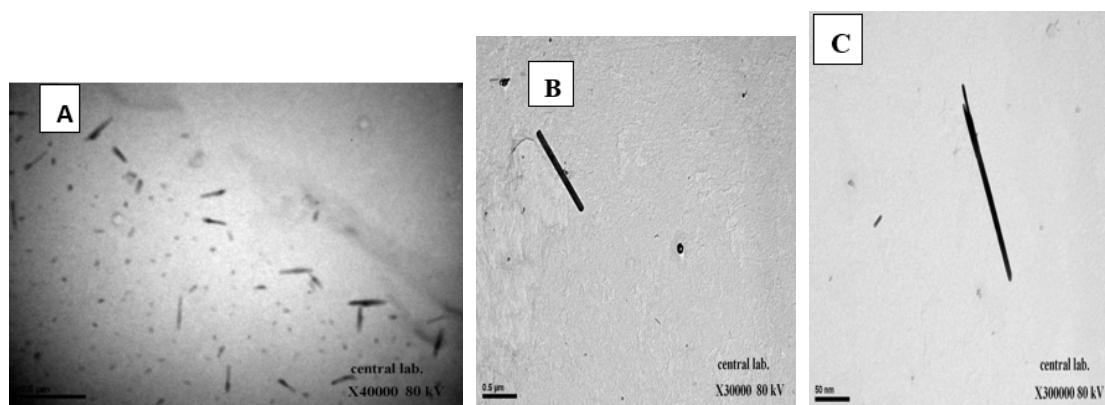
TEM was used to determine the shape and size of the Au-NRs, as shown in Figures 3A, 3B, and 3C. Short nanorods are observed in Figure 3A, but long nanorods appear in Figures 3 B and 3C.

Figure 4 shows the SEM image of the eco Au-NRs synthesized by a mixture of *Olea europaea* fruit extract and *Acacia nilotica* husk extract, showing Au-NRs possessing rods shapes with monodispersity.

Figure 5 presents the EDS analysis for verifying the presence of gold in the suspension of NPs. Our results show a peak and confirm that gold was present in the suspension.



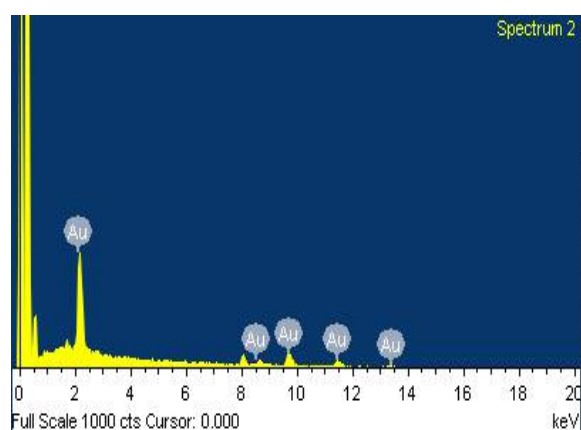
**Figure 2:** Zetasizer results for the nanoparticles



**Figure 3:** (A, B, C) TEM images of the Au-NRs stabilized with the surfactant, CTAB, and a mix of *Olea europaea* fruit extract and *Acacia nilotica* extract



**Figure 4:** SEM image of Au-NRs synthesized by an aqueous mix of *Olea europaea* fruit extract and *Acacia nilotica* husk extract



**Figure 5:** EDS results showing the content (%) of elemental gold in the resulting suspension and the spectrum of the Au-NRs

### Antibacterial activity

Using our novel approach, we determined the antibacterial activity of eco-friendly Au-NRs with an average size of 96 nm at different concentrations (5 or 10 μg/ml) via agar plate-based growth studies on Gram-positive and Gram-negative microorganisms (*Escherichia coli*, *Staphylococcus aureus*). The results of the bacterial growth studies showed that the Au-NRs at both varying concentrations (5, 10 μg/ml) exhibited a significantly higher toxicity against *Staphylococcus aureus* in comparison to the *Escherichia coli*. This was evident from the diameter of the zone of inhibition which was 24mm and 19 mm for *S. aureus* at 10 and 5 μg/ml, respectively, while for *E. coli*, the diameter

of the zone of inhibition was 16 and 15 mm at 10 and 5 µg/ml, respectively.

## DISCUSSION

The mechanism of formation of rod-shaped NPs in aqueous surfactant media remains unclear. In the present study, a novel ecofriendly synthesis of Au-NRs was successfully performed. The consistency of Au-NRs was evident, by the initial changes in color, and the Au-NRs exhibited a plasmon peak in the visible region. The formation of Au-NRs was confirmed by recording the UV–vis spectrum of the colloidal solution. An absorbance peak at ~577 nm was observed in the spectrum. The sharpness of the peak indicated that the nanorods were monodispersed.

The average size of the synthesized AuNRs was measured using a zetasizer system and was ~96 nm with monodispersity. These results of the Zetasizer clearly indicated that the particles were fairly stable with uniform sizes, this was in line with the results of the UV–vis spectrum.

The shape and size distribution for the stable Au-NRs were characterized by particle characterization and TEM analysis. Short and long nanorods appear in Figures 3 B and 3C. Our results are similar to those of another study performed on long NRs prepared in the absence of silver nitrate [18,19]. The TEM images show that Au-NPs have various sizes, which agrees with the results obtained from the analysis of the UV–vis spectrum and the Zetasizer readings.

SEM images exhibited Au-NRs were rod shaped with monodispersity, and EDS analysis confirmed the presence of gold in the suspension.

Gold particles in particular extensively exploited in biology owing to their biocompatibility [19]. Au-NPs are generally considered to be biologically inert but can be structured to possess chemical or photothermal functionality. Under near-infrared (NIR) irradiation, gold-based nanomaterials, nanospheres, nanocages, and Au-NRs with characteristic NIR absorption can destroy cancer cells and bacteria via photothermal heating. Au-NRs conjugated with photosensitizers can kill methicillin-resistant *Staphylococcus aureus* via photodynamic antimicrobial chemotherapy and NIR photothermal radiation [20,21].

Using the novel approach of the present study, we evaluated the antibacterial activity of ecofriendly Au-NRs, with an average size of 96 nm, at different concentrations (5 and 10 µg/ml) via

agar plate-based growth studies on Gram-positive and Gram-negative microorganisms (*Escherichia coli*, *Staphylococcus aureus*). The resulting Au-NRs at both concentrations exhibited a very high toxicity against *S. aureus* than *E. coli*. Thus the study reports a successful ecofriendly synthesis of Au-NRs with antibacterial activity against both Gram-positive and Gram-negative bacteria hence, Au-NRs synthesized by the proposed method could prove to be effective in antimicrobial applications. The biological method employed saves a large amount of energy and is eco-friendly, and economical. Moreover, it does not involve the use of any toxic chemicals.

## CONCLUSION

The key findings of the present study include an eco-friendly synthesis of Au-NRs using CTAB as a surfactant and a mixture (1:3) of *Olea europaea* fruit extract and *Acacia nilotica* husk extract. can be achieved, The technique used is also, simple and cost-effective, and does not involve the use of toxic chemicals. *Olea europaea* fruit extract and *Acacia nilotica* husk extract play an important role in the reduction and stabilization of gold. The method employed in this study synthesis may also be suitable for the preparation of nanomaterials for use in the pharmaceutical and biomedical fields as well as in other industries.

## DECLARATIONS

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### Conflict of Interest

No conflict of interest associated with this work.

### Contribution of Authors

The authors declare that this work was done by the authors named in this article and all liabilities pertaining to claims relating to the content of this article will be borne by them.

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