

## Supplementary material to:

A. Munyaneza, M.D. Bala and N.J. Coville, *S. Afr. J. Chem.*, 2009, **62**, 14–19.

**Table S1** Melting temperatures for  $\text{CpRu}(\text{CO})_2\text{I} + \text{PR}_3$  (1:1) mixtures.

Mixture	Melting point/°C
$\text{CpRu}(\text{CO})_2\text{I} + \text{PPh}_3$	70
$\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-OCH}_3\text{C}_6\text{H}_4)_3$	97
$\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{m-CH}_3\text{C}_6\text{H}_4)_3$	88
$\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-FC}_6\text{H}_4)_3$	70
$\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-ClC}_6\text{H}_4)_3$	87

**Table S2** Reactions of  $\text{CpRu}(\text{CO})_2\text{I}$  and  $\text{PR}_3$  ligands at 100 °C.

Ligand (L)	Time/h	Uncatalyzed		Catalyzed	
		2	3	2	3
$\text{PPh}_3$	0.5	29	13	58	34
	4	50	19	35	65
$\text{P}(\text{p-OCH}_3\text{C}_6\text{H}_4)_3$	0.5	18	0	66	22
	4	44	8	0	100
$\text{P}(\text{m-CH}_3\text{C}_6\text{H}_4)_3$	0.5	0	0	5	95
	4	27	8	0	100
$\text{P}(\text{p-FC}_6\text{H}_4)_3$	0.5	0	15	0	100
	4	0	100	0	100
$\text{P}(\text{p-ClC}_6\text{H}_4)_3$	0.5	0	9	31	69
	4	10	17	0	100

**Table S3** Melting temperatures of  $\text{MeCpRu}(\text{CO})_2\text{I} + \text{PR}_3$  mixtures (1:1 ratio).

Mixture	Melting point/°C
$\text{CH}_3\text{CpRu}(\text{CO})_2\text{I} + \text{PPh}_3$	36
$\text{CH}_3\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-MeC}_6\text{H}_4)_3$	42
$\text{CH}_3\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{m-MeC}_6\text{H}_4)_3$	38
$\text{CH}_3\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-FC}_6\text{H}_4)_3$	36
$\text{CH}_3\text{CpRu}(\text{CO})_2\text{I} + \text{P}(\text{p-ClC}_6\text{H}_4)_3$	40

**Table S4** Reactions of  $\text{MeCpRu}(\text{CO})_2\text{I}$  and  $\text{PR}_3$  ligands at 70 °C.

Ligand (L)	Time/h	Uncatalyzed		Catalyzed	
		2	3	2	3
$\text{PPh}_3$	0.5	0	0	30	0
	4	0	0	100	0
$\text{P}(\text{p-OCH}_3\text{C}_6\text{H}_4)_3$	0.5	0	0	0	0
	4	0	0	10	0
$\text{P}(\text{m-CH}_3\text{C}_6\text{H}_4)_3$	0.5	0	0	0	0
	4	0	0	19	0
$\text{P}(\text{p-FC}_6\text{H}_4)_3$	0.5	0	0	0	0
	4	0	0	0	0
$\text{P}(\text{p-ClC}_6\text{H}_4)_3$	0.5	0	0	0	0
	4	0	0	0	0