

SUPPLEMENTARY MATERIAL

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TABLE S1. A SELECTION OF BIMOLECULAR COMPLEXES FORMED BETWEEN BORON TRIFLUORIDE AND OTHER SMALL MOLECULES IN THE GAS PHASE

Base	Technique	References	Base	Technique	References
Ne	infrared	6,7	H ₂ O	UV-PES ^b	14,15
Ar	infrared	6,7	H ₂ O	UV-PES ^b , EELS ^c	20
Ar	MBER ^a	8	H ₂ S	UV-PES ^b	14,15
Kr	infrared	6,7	H ₂ S	UV-PES ^b , EELS ^c	21
N ₂	MBER ^a	8	SO ₂	microwave	22
HF	microwave	9	NCCN	microwave	19
HF	infrared	10	NH ₃	microwave	17,23,24
HCl	MBER ^a	11	PH ₃	microwave	25
CO	MBER ^a	8	HCCCN	infrared	18
CO	infrared	12,13	CH ₃ F	infrared	26,27
NO	UV-PES ^b	14,15	CH ₃ CN	UV-PES ^b	14,15
HCN	microwave	16,17	CH ₃ CN	microwave	28
HCN	infrared	18	CH ₃ OH	UV-PES ^b	14,15
CO ₂	MBER ^a	19	C ₅ H ₅ N	UV-PES ^b	29
CS ₂	UV-PES ^b	15	(CH ₃) ₃ N	microwave	17,30
N ₂ O	microwave	19	(C ₂ H ₅) ₂ O	UV-PES ^b	14,15

^a MBER – molecular beam electric resonance spectroscopy.

^b UV-PES – ultraviolet photoelectron spectroscopy.

^c EELS – electron energy loss spectroscopy.

TABLE S2. A SELECTION OF BIMOLECULAR COMPLEXES FORMED BETWEEN BORON TRIFLUORIDE AND OTHER SMALL MOLECULES DETECTED BY MATRIX ISOLATION SPECTROSCOPY

Base	Matrix	References	Base	Matrix	References
N ₂	Ar	31	CH ₃ F	Ar	26
CO	Ar	31	CH ₃ CN	Ar	46
CO	Ar, N ₂	32,33	CH ₃ CN	Ar,Xe,N ₂	47
CO ₂	Ar, N ₂	34	CH ₃ CN	Ne,N ₂	48
N ₂ O	Ar, N ₂	35	CD ₃ CN	Ar	49
H ₂ O	Ar	36	CH ₃ NC	Ar	50
H ₂ O,D ₂ O	Ar	37	CH ₂ CHF	Ar	51
H ₂ O	N ₂	38	(CH ₃) ₂ O	Ar	36
H ₂ O	Ne	39	(CH ₃) ₂ O	Ar,N ₂	32,52,53
H ₂ S	Ar	36	(CH ₃) ₂ S	Ar,N ₂	52,53
SO ₂	Ar, N ₂	32,40	(CH ₃) ₂ CHF	Ne	54
BF ₃	Ar,Kr	41	C ₆ H ₅ CN	Ar	55
BF ₃	Ar	42,43	(C ₂ H ₅) ₂ O	Ar	37,56
NH ₃	Ar	44,45	(CF ₃ CH ₂) ₂ O	Ar	37,56
NH ₃	Ar,N ₂	32	(CH ₃) ₃ CCN	Ar	55

TABLE S3. A SELECTION OF BIMOLECULAR COMPLEXES FORMED BETWEEN BORON TRIFLUORIDE AND OTHER SMALL MOLECULES DETECTED BY INFRARED SPECTROSCOPY IN CRYOGENIC SOLUTIONS

Base	Solvent	References	Base	Solvent	References
N ₂	Ar	57	cyclopropene	Ar,N ₂	66
CO	Ar	58,59	cyclopropane	Ar,N ₂	67
OCS	Ar	60	propene	Ar,N ₂	64
COF ₂	Ar,Kr,N ₂	61	methylene cyclopropane	Ar	68
C ₂ H ₂	Ar	62,63	<i>cis</i> -2-butene	Ar	69
CH ₃ F	Ar,Kr	26	<i>trans</i> -2-butene	Ar	69
C ₂ H ₄	Ar,N ₂	64	2-methyl propene	Ar	69
CH ₂ CHF	Ar	51	methyl cyclopropane	Ar,N ₂	70
propyne	Ar	62,63	cyclopentene	Ar	71
allene	Ar,N ₂	65			

TABLE S4. A SELECTION OF THEORETICAL STUDIES OF BIMOLECULAR COMPLEXES FORMED BETWEEN BORON TRIFLUORIDE AND OTHER SMALL MOLECULES

Base	Theory/basis set	References	Base	Theory/basis set	References
Ne	MP2,MP4/LANL2DZ	73	CH ₃ SH	MP2/6-311++G(d,p)	99
Ar	MP2,MP4/LANL2DZ	73	CH ₃ CHO	various	95,111,118
Kr	MP2,MP4/LANL2DZ	73	CH ₃ COF	MP2/6-311++G(d,p)	111
Xe	MP2,MP4/LANL2DZ	73	CH ₃ NH ₂	MP2/6-311++G(d,p)	106
H ₂	MP2/6-311G(2df,2pd),TZ2p	74	CH ₃ PH ₂	MP2/6-311++G(d,p)	109
N ₂	various	75-78	CH ₃ C≡CH	MP2/6-31+G(d)	62,63
F ₂	MP2/6-31+G(d),6-31++G(d,p)	79	CH ₂ =C=CH ₂	MP2/various	65
Cl ₂	MP2/6-31+G(d),6-31++G(d,p)	79	c-(CH) ₂ CH ₂ ^b	DFT,MP2/various	66
Br ₂	MP2/6-31+G(d),6-31++G(d,p)	79	CH ₂ CH=CHO ^c	SCF/6-31G(d)	95
HF	various	9,79-84	HCOOCH ₃	SCF/6-31G(d)	95
HCl	various	79,83	C ₂ H ₅ F	SCF/various	82
HBr	MP2/6-31+G(d),6-31++G(d,p)	79	(CH ₃) ₂ O	various	89,95,97,99,119-121
ClF	MP2/6-31+G(d),6-31++G(d,p)	79	(CF ₃) ₂ O	AM1	121
BrF	MP2/6-31+G(d),6-31++G(d,p)	79	(CH ₃) ₂ S	SCF,MP2/various	99,120
BrCl	MP2/6-31+G(d),6-31++G(d,p)	79	CH ₃ CH=CH ₂	MP2/6-31+G(d)	64
CO	various	33,59,75,85-89	c-(CH ₂) ₃ ^d	MP2/6-31+G(d)	67

BF	SCF/4-31G	75	c-(CH) ₄ O ^e	SCF/6-31G(d)	95
NO	SCF/3-21G	14,15	(CH ₃) ₂ CO	various	95,111,112,118
CO ₂	MP2/6-31G(d)	88	c-(CH ₂) ₂ C=CH ₂ ^f	DFT/6-311++G(d,p)	68
CS ₂	SCF/3-21G	14	(CH ₃) ₂ NH	MP2/6-311++G(d,p)	106
HCN	various	17,45,76,89-93	(CH ₃) ₂ PH	MP2/6-311++G(d,p)	109
FCN	MP2/6-31G(d)	45,76	c-(CH ₂) ₃ O ^g	SCF/6-31G(d)	95
N ₂ O	MP2/6-31G(d)	45,94	CH ₃ CH ₂ CHO	MP2/6-31G(d)	112
OCS	MP2/6-31+G(d)	60	C ₅ H ₅ N	SCF/3-21G	29
H ₂ O	various	14,15,20,38,95-99	CH ₃ COOCH ₃	SCF/6-31G(d)	95
H ₂ S	various	14,15,98,99	CH ₂ =C(CH ₃)CHO ^h	MP2/TZ2P	89
SO ₂	MP2/6-31G(d)	22	CH ₃ CH=CHCHO ⁱ	SCF/6-31G(d)	95
C ₂ H ₂	various	62,63,74	CH ₂ =CHCOCH ₃ ^j	SCF/6-31G(d)	95
C ₂ N ₂	MP2/various	76,78,100	(CH ₃) ₂ CHF	DFT,MP2/aug-cc-pVTZ	54
BF ₃	SCF/4-31G,6-31G(d)	42,101,102	C ₆ H ₆	SCF/STO-3G	81
NH ₃	various	17,24,45,89,97,103-106	c-C ₅ H ₆ O ^k	SCF/STO-3G	115
NF ₃	SCF,MP2/6-31G(d),6-31+G(d)	107	CH ₃ CH=CHCH ₃ ^l	DFT/MP2/6-311++G(d,p)	69
PH ₃	various	103,105,108,109	CH ₃ CH=CHCH ₃ ^m	DFT/MP2/6-311++G(d,p)	69
H ₂ CO	various	95,110-112	(CH ₃) ₂ C=CH ₂ ⁿ	DFT/MP2/6-311++G(d,p)	69
H ₂ CS	MP2/6-311++G(d,p)	111	c-(CH ₂) ₂ CHCH ₃ ^o	DFT,MP2/various	70
F ₂ CO	DFT,MP2/various	61,111	CH ₂ =CHCOOCH ₃ ^p	SCF/6-31G(d)	95

HCOF	MP2/6-311++G(d,p)	111	c-(CH ₂) ₄ O ^q	SCF/6-31G(d)	95
HCCCN	MP2/6-31G(d,p)	76,78	(CH ₃) ₃ N	various	17,89,97,103,106
CH ₃ F	SCF,MP2/various	26,54,82,113	(CH ₃) ₃ P	MP2/various	103,108,109
CH ₃ Cl	MP2/6-311++G(d,p)	113	C ₆ H ₅ CN	DFT/various	122
CH ₂ NH	MP2/6-311++G(d,p)	111	c-(CH) ₂ (CH ₂) ₃ ^r	MP2/various	71
NH ₂ OH	MP2/6-311++G(d,p)	114	(CH ₃) ₃ CF	DFT,MP2/aug-cc-pVTZ	54
C ₂ H ₄	MP2/various	64,74	C ₆ H ₅ CHO	MP2/ TZ2P 6-31G(d)	89,118
CH ₂ CHF	MP2/6-31+G(d)	51	(CH ₃) ₂ C=CHCHO ^s	SCF/6-31G(d)	95
c-(CH) ₂ C=O ^a	SCF/STO-3G	115	(C ₂ H ₅) ₂ O	AM1,SCF/3-21G	14,15,56
CH ₃ CN	various	14,15,89,93,100,116	(CF ₃ CH ₂) ₂ O	AM1	56
CH ₃ NC	DFT/6-311++G(d,p)	50	(C ₂ F ₅) ₂ O	AM1	56
CH ₂ FCN	DFT/aug-cc-pVTZ	117	(CH ₃) ₃ CCN	DFT/various	122
CH ₂ ClCN	DFT/aug-cc-pVTZ	117	CH ₂ =C(CH ₃)COOCH ₃ ^t	MP4/STO-3G	123
CH ₂ BrCN	DFT/aug-cc-pVTZ	117	c-C ₆ H ₈ O ^u	SCF/6-31G(d)	95
CH ₃ OH	various	14,15,95,99			

^a cyclopropenone

^e furan

ⁱ 2-butenal

^m *trans*-2-butene

^q tetrahydrofuran

^u 7-oxanorbornene

^b cyclopropene

^f methylene cyclopropane

^j 3-buten-2-one

ⁿ 2-methyl propene

^r cyclopentene

^c acrolein

^g oxetane

^k dimethyl cyclopropenone

^o methyl cyclopropane

^s 3-methyl-2-butenal

^d cyclopropane

^h 2-methyl acrolein

^l *cis*-2-butene

^p methyl propenoate

^t methyl methacrylate

TABLE S5. A SELECTION OF BIMOLECULAR COMPLEXES FORMED BETWEEN
 NITROUS ACID AND OTHER SMALL MOLECULES
 IN CRYOGENIC MATRICES

Partner molecule	Major type of interaction	References	Partner molecule	Major type of interaction	References
Xe	OH...Xe	124	NH ₃	OH...N	136
N ₂	OH...N	124,125	CH ₄	OH...H ₃ CH	137
HF	FH...O(H)	126	SiH ₄	Si...O(H)	137
HCl	ClH...O(H)	126,127	GeH ₄	Ge...O(H)	137
CO	OH...C	125,128	C ₂ H ₄	OH... π	138
NO	OH...N	129	CH ₂ =CCl ₂	OH... π , OH...Cl	139
CO ₂	OH...O	130	CH ₂ NOH	ONOH...N(OH)CH ₂ , H ₂ CNOH...N(O)OH	140
CS ₂	OH...S	130	CH ₂ =C=CH ₂	OH... π , CH...N(O)OH	141
H ₂ O	ONOH...OH ₂	131	(CH ₃) ₂ S	OH...S	142
SO ₂	OH...O	132,133	(CH ₃) ₂ CO	OH...O	143
NO ₂	OH...O	134	CH ₃ SSCH ₃	OH...S	142
C ₂ H ₂	OH... π	135	(C ₂ H ₅) ₂ O	OH...O	143

TABLE S6. COMPUTED VIBRATIONAL SPECTRUM OF BF₃.HONO COMPLEX
STRUCTURE 1 (C_s)

Symmetry species	Mode	Wavenumber/cm ⁻¹		Intensity/ km mol ⁻¹	Approximate description ^a
		Harmonic	Anharmonic		
a'	v ₁	3708.1	3533.6	147.5	v(OH)
	v ₂	1655.7	1617.1	85.6	v(N=O)
	v ₃	1403.8	1395.0	231.3	v _a (BF ₃)
	v ₄	1337.5	1289.5	344.4	δ(HON)
	v ₅	901.9	821.2	168.1	v(N-O)
	v ₆	862.3	854.7	9.0	v _s (BF ₃)
	v ₇	674.6	640.7	78.8	δ(ONO)
	v ₈	597.2	611.2	264.5	δ _s (BF ₃)
	v ₉	473.9	469.6	8.7	δ _a (BF ₃)
	v ₁₀	202.8	156.2	4.1	<i>l</i> (geared)
	v ₁₁	126.4	101.1	1.5	<i>l</i> (antigeared)
	v ₁₂	56.6	55.0	15.5	v(B...N)
a''	v ₁₃	1459.4	1441.1	397.6	v _a (BF ₃)
	v ₁₄	658.4	589.4	83.6	γ(HON)
	v ₁₅	470.0	464.1	4.8	δ _a (BF ₃)
	v ₁₆	273.3	194.6	13.1	<i>l</i> (geared)
	v ₁₇	95.5	71.5	2.1	γ(OH...F)
	v ₁₈	40.3	14.1	0.5	<i>l</i> (antigeared)

^a v – stretching, δ – in-plane bending, γ – out-of-plane bending, *l* – libration, s – symmetric, a – antisymmetric.

TABLE S7. COMPUTED VIBRATIONAL SPECTRUM OF BF₃.HONO COMPLEX
STRUCTURE 2 (C₁). ALL MODES ARE OF a SYMMETRY

Mode	Wavenumber/cm ⁻¹		Intensity/ km mol ⁻¹	Approximate description ^a
	Harmonic	Anharmonic		
v ₁	3745.9	3563.4	96.5	v(OH)
v ₂	1713.3	1716.8	163.9	v(N=O)
v ₃	1446.1	1403.6	393.7	v _a (BF ₃)
v ₄	1426.1	1384.3	391.6	v _a (BF ₃)
v ₅	1222.7	1135.3	120.5	δ(HON)
v ₆	871.9	855.8	7.4	v _s (BF ₃)
v ₇	738.8	690.5	52.8	v(N-O)
v ₈	623.9	598.8	116.9	δ _s (BF ₃)
v ₉	585.3	549.9	156.4	δ(ONO)
v ₁₀	476.8	464.6	75.8	δ _a (BF ₃)
v ₁₁	470.5	466.2	6.6	δ _a (BF ₃)
v ₁₂	430.4	332.0	288.5	γ(HON)
v ₁₃	166.4	176.1	7.6	<i>l</i> (geared)
v ₁₄	153.9	163.8	7.1	<i>l</i> (geared)
v ₁₅	101.2	72.1	26.2	v(B...O)
v ₁₆	88.6	87.1	6.0	γ(B...OH)
v ₁₇	53.8	45.5	5.9	<i>l</i> (antigeared)
v ₁₈	26.6	28.9	0.6	<i>l</i> (antigeared)

^a v – stretching, δ – in-plane bending, γ – out-of-plane bending, *l* – libration, s – symmetric, a – antisymmetric.

TABLE S8. COMPUTED VIBRATIONAL SPECTRUM OF BF₃.HONO COMPLEX
STRUCTURE 3 (C_s)

Symmetry species	Mode	Wavenumber/cm ⁻¹		Intensity/ km mol ⁻¹	Approximate description ^a
		Harmonic	Anharmonic		
a'	v ₁	3746.5	3564.7	120.3	v(OH)
	v ₂	1633.5	1603.3	94.7	v(N=O)
	v ₃	1455.0	1431.9	396.2	v _a (BF ₃)
	v ₄	1310.0	1265.2	211.0	δ(HON)
	v ₅	888.5	846.6	127.2	v(N-O)
	v ₆	875.6	853.7	87.2	v _s (BF ₃)
	v ₇	679.3	656.6	10.4	δ(ONO)
	v ₈	635.9	632.7	371.2	δ _s (BF ₃)
	v ₉	473.7	471.5	11.1	δ _a (BF ₃)
	v ₁₀	132.2	115.9	4.3	<i>l</i> (geared)
	v ₁₁	83.5	74.8	8.7	v(B...O)
	v ₁₂	52.7	50.2	1.2	<i>l</i> (antigeared)
a''	v ₁₃	1448.2	1427.0	403.6	v _a (BF ₃)
	v ₁₄	614.3	571.9	98.4	γ(HON)
	v ₁₅	473.7	471.4	10.7	δ _a (BF ₃)
	v ₁₆	133.9	120.6	0.2	γ(B...ON)
	v ₁₇	58.5	64.8	3.4	<i>l</i> (geared)
	v ₁₈	26.2	31.4	9.1	<i>l</i> (antigeared)

^a v – stretching, δ – in-plane bending, γ – out-of-plane bending, *l* – libration, s – symmetric, a – antisymmetric.

TABLE S9. COMPUTED VIBRATIONAL SPECTRUM OF BF₃.HONO COMPLEX
STRUCTURE 4 (C_s)

Symmetry species	Mode	Wavenumber/cm ⁻¹		Intensity/ km mol ⁻¹	Approximate description ^a
		Harmonic	Anharmonic		
a'	v ₁	3749.5	3568.0	115.3	v(OH)
	v ₂	1639.1	1606.6	91.0	v(N=O)
	v ₃	1452.0	1428.8	394.6	v _a (BF ₃)
	v ₄	1300.5	1254.5	174.9	δ(HON)
	v ₅	882.5	869.0	11.8	v _s (BF ₃)
	v ₆	861.7	812.4	162.9	v(N-O)
	v ₇	672.8	657.4	112.7	δ _s (BF ₃)
	v ₈	633.8	620.1	181.2	δ(ONO)
	v ₉	472.0	470.3	8.7	δ _a (BF ₃)
	v ₁₀	97.5	90.1	1.2	<i>l</i> (geared)
	v ₁₁	94.4	85.7	5.6	v(B...O)
	v ₁₂	59.6	62.9	1.2	<i>l</i> (antigeared)
a''	v ₁₃	1455.6	1433.6	402.3	v _a (BF ₃)
	v ₁₄	605.8	569.7	99.4	γ(HON)
	v ₁₅	474.6	473.2	12.4	δ _a (BF ₃)
	v ₁₆	149.2	132.5	6.6	γ(B...ON)
	v ₁₇	54.8	54.9	4.1	<i>l</i> (geared)
	v ₁₈	32.2	46.7	0.5	<i>l</i> (antigeared)

^a v – stretching, δ – in-plane bending, γ – out-of-plane bending, *l* – libration, s – symmetric, a – antisymmetric.