

# ASSESSMENT OF LAKE-SALT FOR SOURCING DIETARY MINERALS.

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## ABSTRACT

An analysis of 43 'kanwa' samples collected from various locations in northern Nigeria, Niger, Chad and Cameroun republics for elements and compounds is reported. Sodium occurred in major concentrations (1-44.5%) in all samples; potassium also occurred in all samples but at low levels (0.1-4.7%) except in two samples where it was major (19.6% and 36.5%); zinc (0.00028-0.00042%), magnesium (0.02-0.06%) and aluminium occurred very rarely; iron (0.008-0.04%) was common while chromium occurred only occasionally. Pink coloured samples always contained manganese (0.0016 - 0.0055%). The major compounds isolated from samples were sodium carbonate, sodium hydrogen carbonate, sodium sulphate and sodium chloride. The quantities of 'kanwa' that may be incorporated into food and feedstuffs to source the minerals for various livestock are suggested in line with the experimentally determined composition. Though 'kanwa' may be used to source sodium, potassium, iron and manganese in human food and animal rations, it is unsuitable for magnesium and zinc that require excessively large quantities.

## INTRODUCTION

'Kanwa' is the Hausa name for natron, a dried lake salt and a product of the salt industry in many parts of northern Africa. It is important as a commodity of trade across the sahara desert with a market extending into most of west and central Africa<sup>1</sup>. It has culinary importance in Africa as a food tenderiser in cooking and as an ingredient in the preparation of certain foods<sup>1,2,3</sup>. Medicinal potency has been claimed in various localities and it is an ingredient of traditional medicines for ailments like stomach ache, toothache and constipation<sup>4</sup>. It is also applied as an ingredient in snuff and certain local beverages and as a mordant in dyeing. 'Kanwa' is administered to animals either in drinking water or in their feed in the belief that it provides essential minerals. It is known to induce diarrhoea when taken in water and has been so applied as a purgative in man and livestock. In the Hausa-Fulani region of northern Nigeria, nursing mothers consume large quantities of 'kanwa' daily in a pap of guinea corn as part of the forty-day post-partum practice in the belief that it increases the quantity and improves the

quality of breast milk<sup>5</sup>. This practice has been implicated in the incidence of the syndrome of peripartum cardiac failure (PPCF) among nursing mothers in this area<sup>6</sup>.

Literature reports on the chemical composition of 'kanwa' were based on few samples collected from single locations<sup>7,8,2</sup>. The wide variety of 'kanwa' in the market suggests the need to examine a larger number of samples collected over a wide area of distribution of the substance. For this study, samples were collected at various locations selected to reflect the different trade routes by which the commodity arrives in Nigerian markets. The objectives were to determine the chemical composition of 'kanwa' and to examine the speculation that 'kanwa' may be a source of minerals for man and livestock in the light of the chemical composition determined.

## EXPERIMENTAL

### *Samples and reagents*

Forty three samples of 'kanwa' comprising 4 from Chad, 4 from Cameroun, 20 from Niger and 15 from Nigeria were analysed for chemical composition using analytical reagent grade

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chemicals and distilled water. Numbers were assigned to identify samples for this study only.

#### Instruments

The flame photometer was a Gallenkamp Mark 11, F500 and the atomic absorption spectrophotometer was a Perkin Elmer model 306 using the air-acetylene flame.

#### Qualitative analysis

Each sample was observed with a hand lens for crystallinity before grinding and storing in a glass bottle. Flame test<sup>9</sup> and standard methods of qualitative analysis<sup>9, 12</sup> were applied to detect anions and metallic elements.

#### Quantitative analysis

All of each sample was coned and quartered to obtain a sample for quantitative analysis. 1.0g of the dry sample was dissolved in water or aqueous hydrochloric acid to obtain a clear solution; certain samples had to be boiled in hydrogen fluoride to obtain a clear solution. The solution was made up to 1 litre with water introducing 0.004M lithium chloride to suppress the ionisation interference observed in the determination of Na and K and 0.04M HCl to optimise signal enhancements in the determination of Fe, Mg and Zn. Sample solutions were analysed for Na and K by flame photometry and for Fe, Mg, Mn and Zn by flame atomic absorption spectrophotometry.

#### Bicarbonate/carbonate ratio

The combined carbonate and bicarbonate content of samples was determined by titrating the samples (0.5g) against standard (0.1M) hydrochloric acid using methyl orange as indicator. The ratio of bicarbonate to carbonate was then determined by titrating the samples with the hydrochloric acid using phenolphthalein and methyl orange successively as indicators. The results were checked against those obtained by pH titrations for a few samples.

#### Dietary supplementation

The masses of constituent mineral elements derivable from 1kg of 'kanwa' were

used with literature required daily intake of the minerals<sup>13</sup> to determine the daily consumption of 'kanwa' required by humans to source the minerals. A similar approach was adopted to determine the daily 'kanwa' requirement for various animal rations<sup>14-16</sup>

## RESULTS

#### Qualitative analysis

All samples except one were crystalline in texture. Sample colours varied between and within samples. Many samples were light coloured, ranging from white, through orange to transparent while others were dark in colour including grey, brown and black. Most of the light coloured samples were soluble in pure water while others required dilute acid (HCl); most of the dark coloured samples required digestion with hydrogen fluoride to give a clear solution.

CO<sub>3</sub><sup>2-</sup> was detected in all samples except one and occurred together with HCO<sub>3</sub><sup>-</sup> in all cases. Certain samples also contained SO<sub>4</sub><sup>2-</sup> and/or Cl<sup>-</sup> and PO<sub>4</sub><sup>3-</sup> was detected in one sample. Na and K occurred in all samples; Fe occurred in many samples while Mg, Mn, Zn, Cr and Al occurred in few samples. The results of qualitative analysis are presented with other data in Table 1.

Table 1: Results of chemical analysis

Sample No.	% in dry sample					Other metals present	Anions besides CO <sub>3</sub> <sup>2-</sup> & HCO <sub>3</sub> <sup>-</sup>
	Na	K	Fe	Mg	Mn		
1	15.2	1.5	0.035			Cr	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
2	28.3	2.5					Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
3	28.1	2.9					Cl <sup>-</sup>
4	28.4	0.3					Cl <sup>-</sup>
5	32.2	1.5	0.005		0.0044		Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
6	31.0	0.4					Cl <sup>-</sup>
7	31.7	4.7					Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
8	31.2	0.04		0.019		Al	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
9	32.1	0.02					Cl <sup>-</sup>
10	34.4	0.04					Cl <sup>-</sup>
11	28.1	0.9					Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
12	28.3	1.0					Cl <sup>-</sup>
13	31.7	0.3				0.0052	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
4	44.5	0.1	0.010			Al	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>

Table 1 contd.

15	22.1	19.6				Cl <sup>-</sup>
16	10.1	0.6	0.010			Cl <sup>-</sup>
17	28.4	3.9	0.019			Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
18	16.3	36.5		0.00042		Cl <sup>-</sup>
19	21.7	4.1				Cl <sup>-</sup>
20	20.0	3.7			Al	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
21	27.9	2.9				Cl <sup>-</sup>
22	34.1	2.6	0.016		Cr	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
23	18.4	1.5	0.011		Cr	Cl <sup>-</sup>
24	31.7	2.1				Cl <sup>-</sup>
25	30.0	2.5	0.010	0.039		Cl <sup>-</sup>
26	16.0	2.5				Cl <sup>-</sup>
27	28.1	2.9	0.008			Cl <sup>-</sup>
28	32.0	0.8	0.013	0.0016		Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
29	34.0	1.0		0.0050		Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
30	28.3	0.9	0.060			Cl <sup>-</sup>
31	37.1	0.4		0.0052		Cl <sup>-</sup>
32	30.0	3.1				Cl <sup>-</sup>
33	8.7	1.2				Cl <sup>-</sup>
34	27.4	1.5	0.041		Cr	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
35	31.7	0.9				Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
36	16.6	3.5				Cl <sup>-</sup> , PO <sub>4</sub> <sup>3-</sup>
37	12.5	1.2	0.010			Cl <sup>-</sup>
38	29.2	0.03	0.011	0.0046	0.00028	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
39	32.1	1.0				Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
40	32.1	1.2	0.015	0.0046		Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
41	1.0	0.1		0.00031		SO <sub>4</sub> <sup>2-</sup>
42	33.3	0.3	0.041			SO <sub>4</sub> <sup>2-</sup>
43	30.7	4.2	0.040	0.00030		Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>

Note: Blank spaces imply metal levels below detection.

### Quantitative analysis

Values determined for individual mineral elements are presented in Table 1 and summarised in Table 2. The bicarbonate to carbonate ratios ranged between 1:1 and 2:1

Table 2: Summary of metal concentrations in samples.

Element	Concentration range (%)	Mode (%)	Occurrence (% samples)
Na	1 - 44.5	30	100
K	0.1 - 36.5	3	100
Fe	0.008 - 0.041	0.01	32.5
Mg	0.019 - 0.060	0.040	11.6
Mn	0.0016 - 0.0055	0.005	16.2
Zn	0.00028 - 0.00042	0.0003	9.3
Al	-	-	2.3
Cr	-	-	9.3

### Dietary supplementation

The required daily intake of 'kanwa' that would source minerals in the diet of humans and livestock is presented in Tables 3 and 4 respectively.

Table 3: Required daily consumption of 'kanwa' for humans

Element	Mass per kg of 'kanwa' (g)	ADI (g)	Daily required 'kanwa' (kg)
Na	300	2.2(0.44)	0.007(0.0015)
K	30	4.0	0.133
Fe	0.1	0.010	0.100
Mg	0.4	0.350	0.875
Mn	0.05	0.005	0.100
Zn	0.003	0.015	5.000

## DISCUSSION

'Kanwa' is common in arid and semi-arid regions and has been described as natron or trona<sup>4,6</sup> and presented chemically as a mixture of sodium carbonate and sodium bicarbonate<sup>8</sup> or sodium carbonate and sodium sulphate<sup>4</sup>. While these compositions may relate to hand-picked, sorted or semi-purified components of 'kanwa', the unmodified substance is a complex mixture of different substances.

CANDISC was further used to analyse the output data from FASTCLUS. CANDISC performs canonical discriminant analysis and does both univariate and multivariate one-way analysis of variance. From these, canonical coefficients were calculated. An important provision of CANDISC is the PLOT algorithm which pairs the canonical variables<sup>12</sup> using 'K-means' clustering. Univariate test statistics from the canonical discriminant analysis were used in correlations.

Results of tests performed with some plant extracts on dogs and rabbits induced with particular ailment are shown in Table 4<sup>10,11</sup>. Extracts of the plants FSB, FPB and ASB, all members of cluster 3, were used and positive results were obtained<sup>10,11</sup>.

## RESULTS AND DISCUSSION

### Statistical analysis by elemental contents

Three clusters (groups) are discernible from the dendrogram. These are, CL9-(XAR, FPR, FTR\*); CL10-(MML, HCB) and FTR\* and FSB, CL6-(XQR, CPL, CPR, FSB, FSR, FPB) and CL-7 (MMR, ASB) denoted as clusters 1, 2 and 3 respectively. These have been classified according to the highest normalized RMS values. It is to be noted that the closer the variables cluster to the left of the dendrogram, the more similar they are; the further apart to the right the more dissimilar. Members of cluster 3 had the highest Ca and Mg concentrations with respective ranges of 21.0 - 32.5% and 0.48 - 7.3% (Table 1). Cluster 1 members had the highest K levels (12.0 - 18.0%) while Cluster 2 registered the lowest values of Ca, K and Fe according to this clustering procedure. The observed clustering is further highlighted by Fig. 2 which illustrated

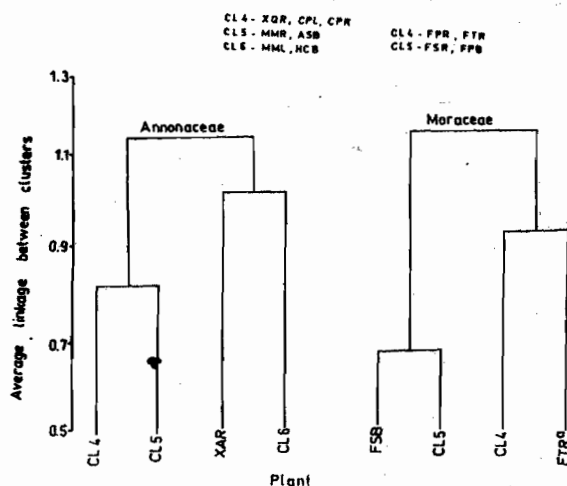


Figure 2: Average linkage cluster analysis by individual plant families

the tree diagram of the individual families. Agreement between Figs. 1 and 2 tends to negate the existence of outliers in the data set.

From table 2 it is noted that respective average values of K, Fe and Ca are highest for Cluster 1, 2 and 3 in that order. The clustering indices for these 3 clusters are within the range of  $13.04 \pm 4.35$ ,  $1.06 \pm 1.26$  and  $28.80 \pm 3.36$ . The RMS standard deviation between observations within the clusters 1, 2 and 3 are 1.73, 1.43 and 1.88 respectively. These help to reflect the nearness of cluster 1 to both 2 and 3. It is implied that some characteristics of the plant members of cluster 1 are found in members of both clusters 2 and 3. The absence of a cluster with only one member here once again negates the presence of an outlier within the data set. This is so because FASTCLUS is very sensitive to outliers and constitutes a useful procedure for outlier detection<sup>12</sup>.

CANDISC aids visual interpretation of the group differences. Fig. 3 illustrates the

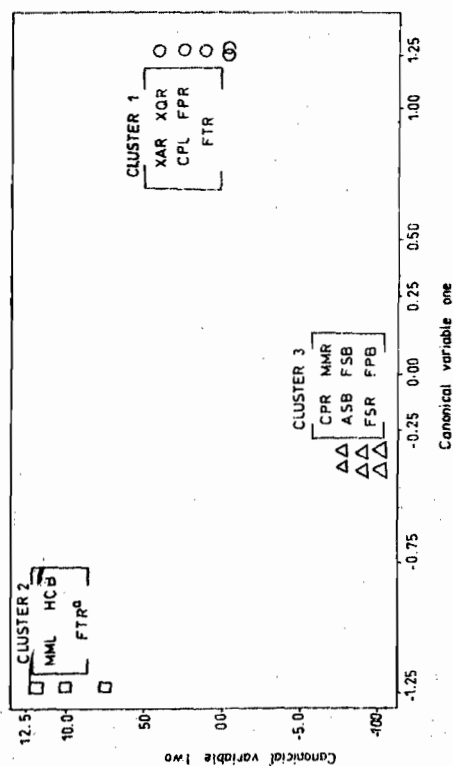


Figure 3: K Means cluster plot of Annonaceae and Moraceae plants showing location of the three clusters

"K-means" plot which further exhibits the existence of clusters. Members of the clusters here also agree with those produced by FASTCLUS. High  $R^2$  (squared multiple correlation) is desirable to differentiate between clusters. Thus the highest  $R^2$  shown for Ca (0.883), K (0.713) and Fe (0.140) formed the clustering indices. The values for Cd and Cu

were high because their raw values (Table 1) were relatively low for all the plant samples analysed.

There seems to be a disparity between CLUSTER/TREE AND FASTCLUS/CANDISC procedures in the placement of the plants XQR and CPL. These respectively placed both plants in groups 3 and 1. The seeming anomaly arises from the range of elemental values used as clustering indices. Whereas TREE used Ca range of 21.0-32.5%, (K range of 2.8-10%), FASTCLUS/CANDISC used 13.8-25% Ca and 8.1-18% K for clusters 3 and 1 respectively. These ranges invariably displaced the plants XQR and CPL from cluster 3 (TREE) to cluster 1 (FASTCLUS), as XQR extract had Ca and K values of 21.9% and 8.1% while CPL recorded 21.0% and 10.0% for both metals. Besides the agreement between the two SAS procedures were indeed remarkable. A summary of the grouping is provided in Table 3.

Table 3: Plant potency similarities suggested by clustering analysis.

Cluster Index	Plant samples included	
	Annonaceae	Moraceae
1) 8.1-18.0% K	XAR, XQR, CPL	FPR, FTR*
2) 0.43-2.5% Fe	MML, HCB	FTR*
3) 25.0-32.5% Ca	CPR, MMR, ASB	FSB, FPR, FPB

*Medicinal efficacies of the plant extracts on animals*

Extract doses of FSB (200mg/ml) and FPB (100mg/ml), both increased the B.P. of the dogs and acted as depressant on the CNS of rabbits. Observed differing degrees of the effect between the plants in both tests may be discussed in terms of some of their metal contents shown in Tables 1 and 4. Extracts of

Table 4: Test of certain plant extracts on animals

Observation	Effect of extract		
	FSB	FPB	ASB
1) Dog B.P.	+3.1%	+12.3%	NT
2) Rabbit CNS	Slightly depressing (200mg/ml)	Highly depressing (100mg/ml)	NT
3) Diarrhoea in rabbit	NT	NT	Positive antimicrobial activity

NT = Not Tested: ( ) = Dosage

FSB with higher Ca and Na levels but lower K, increased a dog's B.P. by 3.1% while FPB with lower K showed a four-fold increase over that of FSB. It appears that K has an overriding effect on the B.P. as Ca, apparently, could not emolliate the effect of K in FPB extract as it seems to have done on Na in FSB extract. For CNS, extracts of FSB with a hundred and fifty orders of magnitude higher Pb content compared to FPB proved to be a lower depressant even with double the extract doses compared to FPB.

Potassium and sodium have been specifically mentioned for these as they have been medically associated with elevated B.P. in man

while Pb has been associated with mental disorder especially in children<sup>20</sup>. It was also reported<sup>10</sup> that extract doses of ASB showed positive antimicrobial activity when tested on rabbits. ASB had relatively high levels of K and Ca but low levels of Na and Pb (Table 1).

A point to note here is that extracts of plants within a class reacted similarly towards respective ailments. If such classification is adopted it might save experimentation time as members of a group would then be subjected to similar tests. Moreover plants like FPB and ASB of different families but with similar elemental contents (Table 1) may show similar potency towards particular illnesses and ought to be tested as such.

### CONCLUSION

SAS computer packages used respective concentration ranges of K, Fe and Ca to efficiently place some of the Annonaceae and Moraceae families in three similarity clusters. The clustering exercise may have placed the plants in groups of equivalent potency towards particular illnesses treated by ATM.

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