

Specialized trace elements and volatile fatty acids interactions for enhanced methane production and biomethanization process stability during high organic loading rate

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Abstract

Volatile fatty acids (VFAs) and trace elements (TEs) interactions (VFAs*TEs) during biomethanization have effects that could be exploited to enhance anaerobic digestion (AD) of biomass. The goal of this study was to validate biocatalytic effects of specialized VFAs*TEs identified from a batch-derived Optimum TE Configuration (or simply 'Optimum') on high organic loading rate (OLR) involving mixed fruit residue (MFR) fed in semi-continuous AD operation. The specialized VFAs*TEs were formulated as Variants of the Optimum and included Optimum –Cobalt (Co) for specialized VFAs*Co effects, and Optimum +Selenium (Se) for specialized VFAs*Se effects. Four duplicate AD reactors were treated with formulations reflecting the Optimum and the Variants. Each duplicate reactor was semi-continuously fed with MFR at varying OLR until instability occurred. Methane production, total volatile organic acidity (FOS) / total alkalinity (TAC) and VFAs fingerprints were measured as main responses. The results showed that reactors of the Optimum and its Variants were unstable at OLR of 8g oDM/L/d, but stability was restored in the Optimum – Co (FOS/TAC values of 0.6 compared to 1.51 and 1.67 for Optimum and Optimum +Se respectively). The average specific CH₄ production (*Nml/g oDM*) of the Optimum and its Variants were Control: 431 ± 36 ; Optimum: 553 ± 16 ; Optimum –Co: 580 ± 12 ; and Optimum +Se: 545 ± 13 . Optimum –Co also had the lowest acetic acid and butyric acid accumulation, but had higher propionic acid concentration (0.7 g/L) compared to the Optimum (0.3 g/L) and Optimum +Se (0.4 g/L).

Keywords: Process stability, volatile fatty acids, trace elements interaction, organic loading rate, biocatalysis

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Introduction

Biomethanization involves biochemical processes and pathways that favor methane production during biomass degradation. Anaerobic digestion (AD) reactors are supplemented with trace elements (TEs) during biomethanization to maximize the processes of methane (CH₄) production, volatile fatty acids (VFAs) degradation and process stability (Braga et al., 2018; Mancini et al., 2018; Yazdanpanah et al., 2018). TEs improve biomethanization by

enhancing the biocatalytic potentials of metallo-enzymes (MEs) that are associated with AD, reduce digester acid accumulation and improve CH₄ production (Ezebuio and Koerner, 2017; Ezebuio et al., 2018; Da Silva et al., 2021). Managing digester acidification is important because VFAs accumulation beyond critical levels of 200 mmol/L may lead to biomethanization failure (Zhang et al., 2010; Bardi and Aminirad, 2020). It has been reported that different configurations of TEs serve

different AD phases due to the differences in the enzymology of the pathway(s) that characterize AD phases (Ezebuiri and Koerner, 2017; Ezebuiri et al., 2018).

Biochemical effects of TEs vary widely when used to enhance biomethanization and unexplainable variations due to TEs supplementation have limited the use of TEs for industrial optimization of AD processes (Ariunbaatar et al., 2016; Facchin et al., 2013; Qiang et al., 2013). Research showed that while individual TEs are important (Pobeheim et al., 2011; Ünal et al., 2012; Myszograj et al., 2018), the most important effects originate from the interaction between TEs (TE*TE) and between a TE and VFAs (VFAs*TE). These interaction effects influence substrate hydrolysis, acidification, acetogenesis and methanogenesis (Ezebuiri et al., 2018; Ezebuiri and Koerner, 2017). For most biomass used as substrates in AD, acidification during biomethanization is related to the organic loading rate (OLR). Controlling digester acidification is important because biomethanization instability has been reported to set in with accumulation of VFAs, especially propionic acid (Zhang et al., 2010).

Earlier researches involving derivation of optimum TEs concentrations and modelling of VFAs*TEs were conducted in batch mode using VFAs mixtures as substrate (Amani et al., 2010; Ezebuiri et al., 2018). It is known that digester solids influence bioavailability of TEs (Tokalioğlu et al., 2003; Alonso et al., 2009; Maharaj et al., 2018); hence, the use of actual biomass to confirm special VFAs*TEs effects has become necessary. This not only allows for detailed studies of the special TEs interactions using industry-grade biomass fed in more practical semi-continuous mode, but also enables the variation of the OLR of the biomass to determine the constraints or limits of biocatalysis of observed VFAs*TE effects. Therefore, the objectives of the current study include to:

1. Adapt specialized VFAs*TEs observed with batch-identified Optimum TEs Configuration (OTC or 'Optimum') for high organic loading in a semi-continuous AD operation using mixed fruit residue (MFR) as substrate;
2. Validate the process efficiency of the adapted TEs interactions in (1) with regards to improving process stability,

methane production and VFAs fingerprints; and

3. Determine the process efficiency constraints or boundaries of the adapted TEs interactions with regards to process stability, methane production and VFAs fingerprint.

Materials and Method

This study is a follow-up to an earlier publication (Ezebuiri et al., 2018). While current method and materials are discussed in detail, published method and materials are summarized and referenced.

Specialized TEs Interactions that Influence VFAs Degradation Rate:

The special interactions that influenced VFAs degradation rate were identified from the effects of an Optimum TEs Configuration derived from the published Response Surface Model (RSM) showing the synergistic and antagonistic influences of TEs on VFAs metabolism (Ezebuiri et al., 2018). The procedures for identification of an Optimum TEs Configuration (OTC) and derivation of the specialized TEs interactions, as they relate to the current study are summarized.

Identification of an Optimum TEs Configuration:

The OTC or "Optimum" (mg/L) include the Ni, Co, Se and Mo concentrations that simultaneously maximized VFAs degradation rate and CH₄ production in the biomethanization of *circa* 120 mmol/L VFAs mixture (medium VFAs level) comprising acetic-, propionic- and butyric acids. The OTC was identified from a mesophilic batch experiment involving 24 units of 1-litre reactors fed with VFAs mixture and supplemented with Ni, Co, Se and Mo in varying concentrations of low-, medium- and high levels. The supplementation matrix was designed statistically using the Custom Design Module of JMP 10. The experiment was run following the procedures recommended for the determination of methane potential of biomass (Verein Deutscher Ingenieure, 2006).

Liquid samples were taken at 3 to 4 days intervals from the reactors and analyzed for VFAs concentration and degradation rate. The data from VFAs degradation rates were further analyzed with the response surface model (RSM) to determine how Ni, Co, Se, Mo and VFAs

concentrations (factors) and their interactions influence VFAs degradation rate. To maximize the influence of the significant factors and derive the optimum factor concentration or the OTC, the desirability function proposed by Suich and Derringer, (1980) was applied to the RSM.

Derivation of Specialized TEs Interactions:
The specialized TEs were derived from Figure 1

that described the behavior of VFAs degradation rate at 37°C as a result of Ni, Co, Se and Mo supplementation to VFAs concentrations between 28 and 213 mmol/L as published in Ezebuiri et al. (2018). At $\alpha = 0.05$, VFAs*TEs terms with *p-value* ($Prob > |t|$) less than 0.05 or having negative estimate were considered specialized interactions.

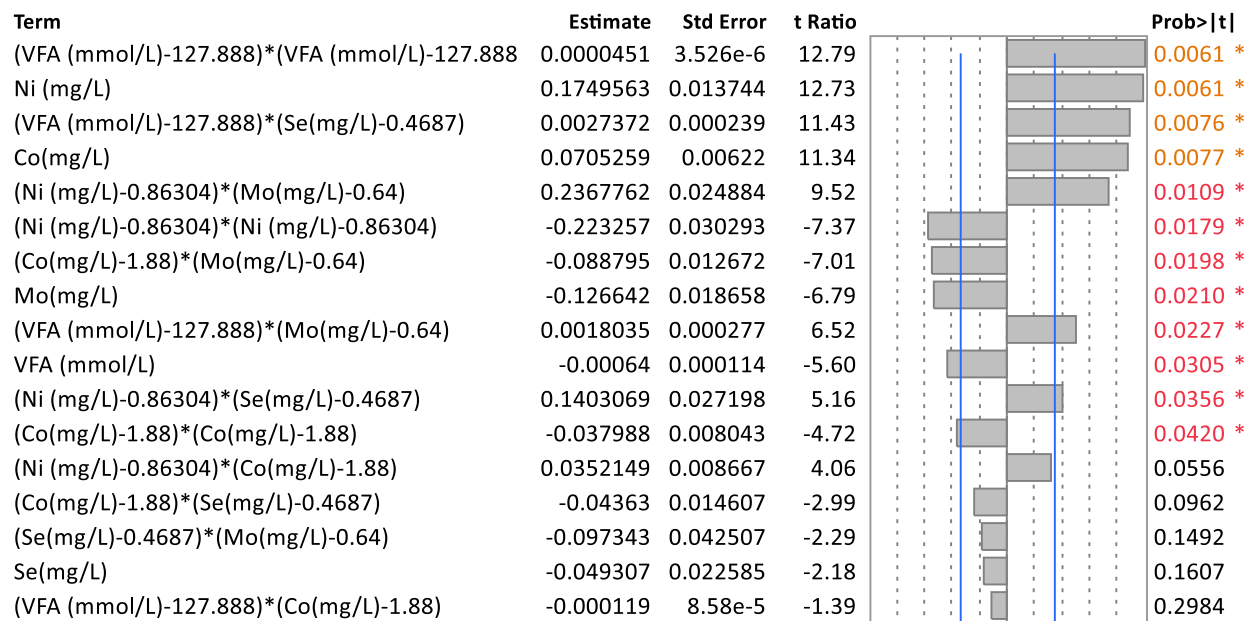


Figure 1 RSM terms (TEs and VFAs*TEs), estimates (magnitude and orientation of the terms) and significance of the terms that described the behavior of VFAs degradation rate at 37°C as a result of Ni, Co, Se and Mo supplementation to VFAs concentrations between 28 and 213 mmol/L (Adopted from Ezebuiri et al., 2018). Note: $Prob > |t|$ is the probability of obtaining the estimated value of the parameter if the actual parameter value is zero.

Adaptation of Specialized TEs Interactions: The current study adapted published OTC and specialized VFA*TE effects to high organic loading rate using industry-grade biomass mixture (mixed fruit residue) as substrate. The specialized VFA*TE were constituted as alternative conditions (or variants) to which the OTC could be subjected to further enhance biomethanization. VFA*Se, VFA*Mo and VFA*Co were considered specialized interactions from Figure 1, but VFA*Se and VFA*Co were further evaluated in the current study. Based on the orientation of the Estimates of VFA*Se and VFA*Co, the following variants of the OTC were derived:

Optimum +Se Variant: This Variant of the OTC highlights the VFAs*Se effect on process

resilience during acid accumulation at reactor acidification level above 200 mmol/L VFAs. The Optimum +Se Variant contains the OTC but also an increase in Se concentration from 0.50 mg/L that is optimum for VFAs concentration ≤ 120 mmol/L, to 1.50 mg/L that is modelled as more beneficial for VFAs levels ≥ 200 mmol/L.

Optimum -Co Variant: This Variant highlights the effect of Co supplement in VFAs*Co, which suggests that Co concentration should increase above the OTC at lower OLR and *vice versa* (Ezebuiri et al., 2018). It was proposed to validate the assumption that a decrease in Co concentration in the OTC will confer process resilience during acid accumulation at reactor acidification level ≥ 200 mmol/L VFAs. Hence, the Optimum -Co Variant contains the OTC but

the concentration of Co was decreased from 2.70 mg//L that is optimum for VFAs concentration \leq 120 mmol/L to 0.50 mg/L that

Table 1: TEs configuration of the 'Optimum' and its Variants used for reactor dosing in the semi-continuous AD experiment

Supplementation arrangements (Reactors)	mg/L			
	Ni	Co	Se	Mo
Control (R-1)	0.09	0.04	0.00	0.04
Optimum (OTC) (R-2)	1.46	2.70	0.40	0.50
Optimum -Co (R-3)	1.46	0.50	0.40	0.50
Optimum +Se (R-4)	1.46	2.70	1.50	0.50

Note: The Control TEs configuration is the TEs content and composition of the inoculum used for the current study

Validation and Determination of the Practical Constraints or the Process Efficiency Boundaries of the Specialized TEs Interactions:

To determine the practical biochemical constraints or boundaries of the OTC and its Variants namely Optimum +Se and Optimum -Co, AD experiments were conducted in semi-continuous mode in 2000 mL reactors using MFR as substrate. The experimental reactors contained TEs formulations (Table 1), nutrient solution, MFR and inoculum. The experimental MFR used in this study was a mixture of different fruit residues from a fruit processing company in Hamburg. It was blended

is modelled as more beneficial for VFAs concentration \geq 200 mmol/L. The OTC or Optimum and the Variants are shown in Table 1.

to consistent slurry and stored at -18°C until ready for use. Carbon (C) and Nitrogen (N) composition of the MFR were determined based on DIN EN 15104:2011-04 (E); and Ni, Co, Se and Mo contents of the MFR were determined as described in DIN EN 13346:2001-04. The inoculum used for the experiment was the digestate obtained from a mesophilic facility digesting maize silage composite. The TEs compositions of the inoculum were determined following the methods described for the MFR. The MFR was loaded to the mesophilic reactors in semi-continuous mode following the feeding schedule in Table 2.

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Table 2: Organic loading rate of mixed fruit residue used as substrate during the mesophilic semi-continuous AD experiment

Days	OLR (g oDM/L/d)	OLR Level	Cumulative OL (g oDM/L)
1-12	0.7	Low	8
13-18	1.3	Low	16
19-25	2.0	Low	28
26-37	4.0	Medium	68
38-42	8.0	High	100
43-46	6.0	High	124
47-49	Recovery period	-	124

Note:

OL, organic loading; OLR, organic loading rate

The test system used for the study was designed following the VDI 4630 guidelines (Verein Deutscher Ingenieure, 2006) and consisted of eight (8) units of 2000 mL glass reactors, each of which were connected to two eudiometers. The scheme of the test system is shown in Figure 2A and picture of the actual test

system is shown in Figure 2B. The experiment was performed in duplicates. Each duplicate reactor contained 400 mL of the inoculum; nutrient solution; TEs configuration of the OTC or either of its Variants (Table 1); and the MFR (Table 2). Except for the background TEs in the MFR and inoculum, no TEs was supplemented to the Control reactors. The content of each

duplicate reactor was made up to 1600mL with distilled water, thereby providing a headspace of 400 mL or 20% of the reactor volume.

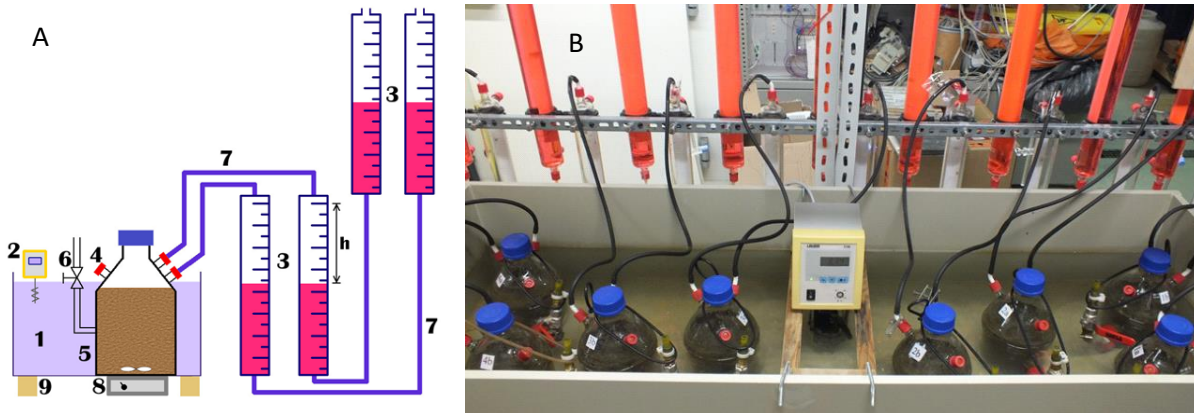


Figure 2(A) Basic scheme of the test system used for the validation investigations (1) water bath (2) thermostat (3) eudiometers (4) gas sampling port (5) glass reactors (6) valve for feeding and liquid sampling (7) reactor-eudiometer connecting tubes (8) magnetic stirrer and (9) water bath support; and **(B)** Actual picture of the validation test system.

The loaded reactors were placed in a water bath that was maintained at 37°C using a thermostat. A magnetic stirrer-bar was placed inside each reactor and eight magnetic stirrers were placed beneath the water bath such that each reactor was stirred by a magnetic stirrer through the magnetic stirrer-bar in the reactor. The loaded reactors were further flushed with dinitrogen gas for one minute to displace any oxygen (Verein Deutscher Ingenieure, 2006). During the experiments, liquid and gas samples were taken from the reactor every 3 to 4 days using a 50 mL syringe and biogas volume was measured within the same interval. The specific CH₄ production was derived as the ratio of cumulative CH₄ produced over a period to the total organic load fed to the reactor in the same period.

Table 3 summarizes the standard methods used for the analysis of the experimental parameters that include biogas volume (normalized), % CH₄, volatile organic acidity (FOS), total alkalinity (TAC), pH, and individual VFAs concentration.

Table 3: Equipment, standard methods and modification to standard methods used for sample analyses during the semi-continuous AD study

Parameter	Sample phase	Equipment/ Standard Method	Modifications to (Standard) Method
(1) Biogas volume	Gas	VDI 2006: Eudiometer method and normalization was as described therein.	-
(2) CH ₄ content (%)	Gas	Biogas Analyzer 5000, UK	-
(3) pH	Liquid	DIN 38404 – 5	-
(4) FOS	Liquid	PRONOVA (FOS/TAC) 2000 Analysentechnik GmbH & Co. KG	15mL instead of 50 mL

(5) TAC	Liquid	PRONOVA (FOS/TAC) 2000 Analysentechnik GmbH & Co. KG	15mL instead of 50 mL
(6) Individual VFA:	Liquid	PerkinElmer Autosystem: column HP-FEAP 30m, 0.25mm ID, 0.25 FD (JAS); Carrier gas: H ₂ 5.0 1mL/min; Detector: FID; Injector temperature 280°C; FID temperature 300°C; Split ratio 15:1; Liner 4mm fine split completed and packed with glass wool; Oven temperature program: 80°C (1min), 6°C/min, 170°C, 20°C/min, 200°C (2mins)	-
a. Acetic acid			
b. Propionic acid			
c. Normal butyric acid			
d. Iso-butyric acid			
e. Normal valeric acid			
f. Iso-valeric acid			

Results

Specialized TEs Interactions that Influence VFAs Degradation:

From Figure 1, the significant synergistic TEs and VFAs interactions (VFAs*TE) for optimized VFAs degradation were the VFAs*Se, Ni*Mo, VFAs*Mo and Ni*Se; and the antagonistic interaction was Co*Mo. With emphasis on the orientation of VFAs*TE (i.e., positive or negative values of Estimate), VFAs*Se, VFAs*Mo and VFAs*Ni were positive. This suggests that for optimum VFAs degradation, Se, Mo and Ni concentrations must increase beyond the OTC as VFAs levels increase in the digester. Se*Mo was negative and practically implies that the process of VFAs degradation requires either Se or Mo but not both. Hence, an increase in Se in an optimum supplementation formula intended for biocatalysis should be accompanied by a decrease in the Mo concentration. Furthermore, VFAs*Se was more positively significant for VFAs degradation (0.0076) compared to VFAs*Mo (0.227) and VFAs*Ni (0.5124). Therefore, VFAs*Se is considered a superior Variant of the OTC relative to VFAs*Mo and VFAs*Ni. The positive significance of VFAs*Se means that when VFAs concentration increases above 128 mmol/L, supplemented Se concentration should also increase above 0.5 mg/L.

Regarding negative influences, VFAs*Co was the only negative VFAs*TE in the VFAs degradation model. Although statistically insignificant ($\alpha = 0.05$) within the limits of the studied concentrations (Table 1), VFAs*Co has an inverse relationship with VFAs concentration. Hence, VFAs*Co is considered a probable

Variant of the OTC; and Figure 1 shows that for optimum VFAs degradation, VFAs concentration greater than 128 mmol/L requires Co concentration less than 1.88 mg/L.

Validation of Process Efficiency and Determination of the Efficiency Constraints of the Optimum TEs Configuration and its Variants

The MFR used in this study had a dry %DM content of 10.79 ± 0.08 ; C and N content (as %DM) of 45.52 ± 0.05 and 2.2 ± 0.03 respectively; and C / N of 21. The Ni, Mo and Co contents (mg/kg DM) of the MFR were 0.63 ± 0.08 , 0.6 ± 0.13 and 0.63 ± 0.18 respectively. The concentrations of Ni, Co, Se and Mo in the experimental inoculum have been shown in Table 1.

Methane Production at OLR of 2g oDM/L/d:

The specific CH₄ production of a reactor gives indications of its biochemical efficiency at converting carbon source to CH₄. Figure 3a-c shows the specific CH₄ production of the reactors (Table 1) at the different OLR (Table 2). Figure 3a shows Specific CH₄ production during the OLR of 0.7 – 2.0g oDM/L/d in the OTC and its variants. Optimum +Se reactor showed relative weakness in specific CH₄ production due to instability arising from the negative influences of higher-than-optimum Se concentration at comparatively low OLR (0.7 to 2g oDM/L/d compared to 4.0g oDM/L/d in Figure 3b). At the end of 2g oDM/L/d OLR, the specific CH₄ (Nml/g oDM) produced by the reactors were: **Control,**

589±39; **Optimum**, 612±6; **Optimum -Co**, 635±6; and **Optimum +Se**, 552±10.

production (Nml/g oDM) of the reactors were as follows: **Control**, 538±11; **Optimum**, 625±4; **Optimum -Co**, 645±10; and **Optimum +Se**, 610±5; and these amounted to 16%, 20% and 13% more methanogenic activities compared to the Control reactor.

Methane Production at OLR of 4g oDM/L/d: The specific CH₄ production during the OLR of 4g oDM/L/d is shown in Figure 3b. At the end of this period (Day 37), the specific CH₄

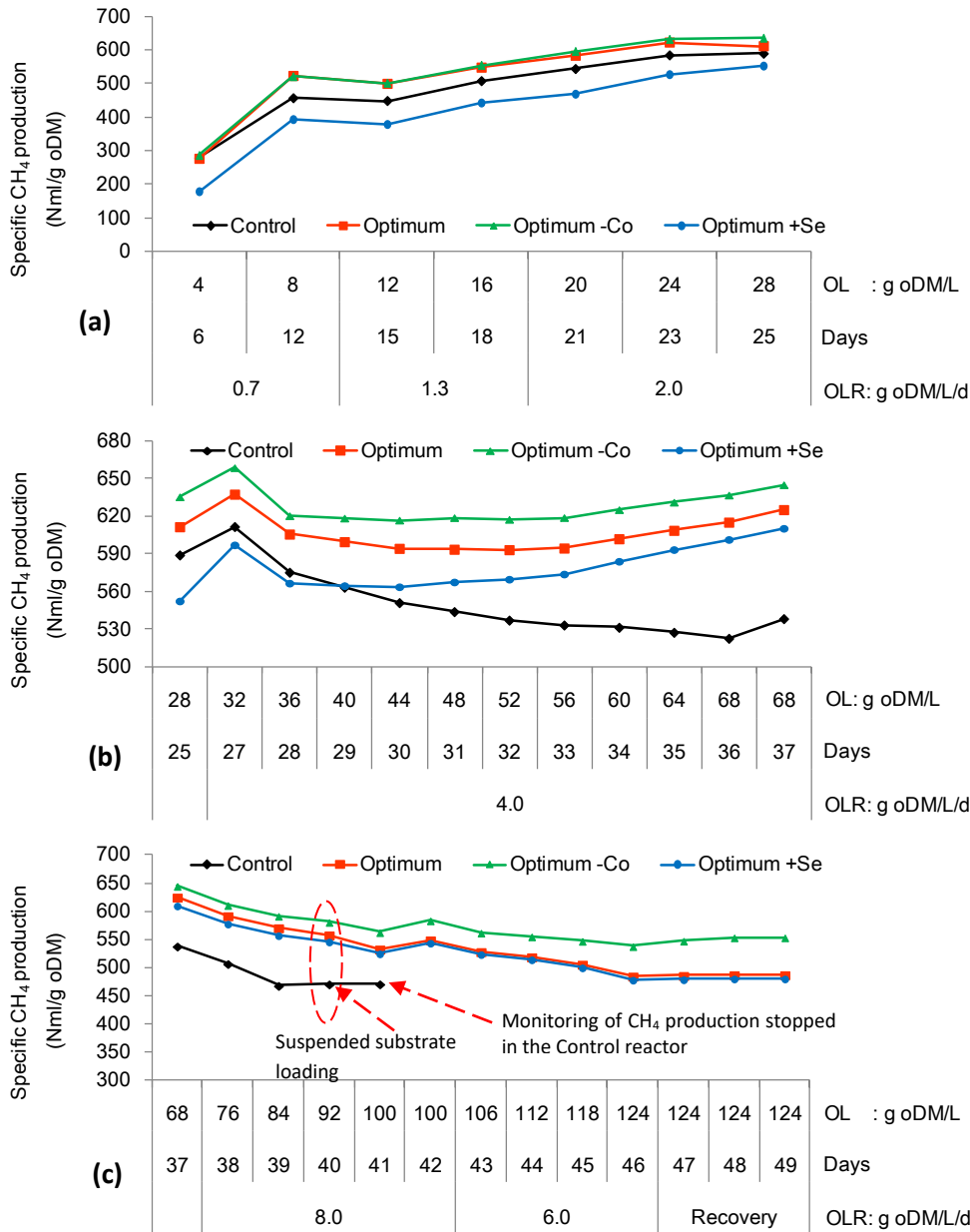


Figure 3a –c Specific CH₄ production of the OTC and its Variants due to Ni, Co, Se and Mo supplementation at different OLR: **(a)** OLR: 2g oDM/L/d **(b)** OLR: 4g oDM/L/d **(c)** OLR: 8g and 6g oDM/L/d

Methane Production at OLR of 8g oDM/L/d and the Recovery Period: Figure 3c shows

the specific CH₄ production between the high OLR of 8g oDM/L/d and the recovery period

when OLR was reduced to 6g oDM/L/d due to VFAs accumulation in the reactors. The specific CH₄ production (Nml/g oDM) of the Control reactor was 538±11 at the end of the 4g oDM/L/d (day 37), but decreased to 507±6 after the first OLR of 8g oDM/L/d induced reactor acidification (pH of 6.2, day 38). The specific CH₄ production (Nml/g oDM) of the Control reactor further decreased to 468±5 on day 39 even though pH was adjusted to 7.0 on day 38 using 1M NaOH. At this time (day 39), the TEs supplemented reactors had specific CH₄ production (Nml/g oDM) of 571±1 (**Optimum**); 591±7 (**Optimum -Co**); and 557±2 (**Optimum +Se**) compared to 468±5 in the **Control**. This corresponded to 22% (**Optimum**), 26% (**Optimum -Co**) and 19% (**Optimum +Se**) more CH₄ production per day than the Control reactor. Organic loading and CH₄ production measurement were suspended indefinitely on day 39 in the Control reactor due to acidification.

At the end of the OLR of 8g oDM/L/d (day 42), the specific CH₄ production (Nml/g oDM) of the TEs supplemented reactors were as follows: **Optimum**, 549±9; **Optimum -Co**, 584±2; and

Optimum +Se, 544±4. Instability in biometanization due to acidification induced general decline in specific CH₄ production in the TEs supplemented reactors and necessitated reducing the OLR to 6g oDM/L/d on day 43, and final suspension of digester loading on day 46. A three-day recovery period was allowed and parameter measurements were stopped on day 49 in all the TEs supplemented reactors. The specific CH₄ production (Nml/g oDM) of the TEs supplemented reactors at the end of the experiment on day 49 were: **Optimum**, 487±6; **Optimum -Co**, 554±2; and **Optimum +Se**, 481±5. The Optimum -Co reactor had 14% and 15% more specific CH₄ production than the Optimum and the Optimum +Se reactors respectively.

Process Stability (Acidity/Alkalinity): The ratio of volatile organic acidity to total alkalinity (FOS/TAC) was used as indication of process stability. Figure 4 shows an overview of FOS/TAC, OLR of the reactors and the total organic load fed to the Control, Optimum, Optimum -Co and Optimum +Se reactors.

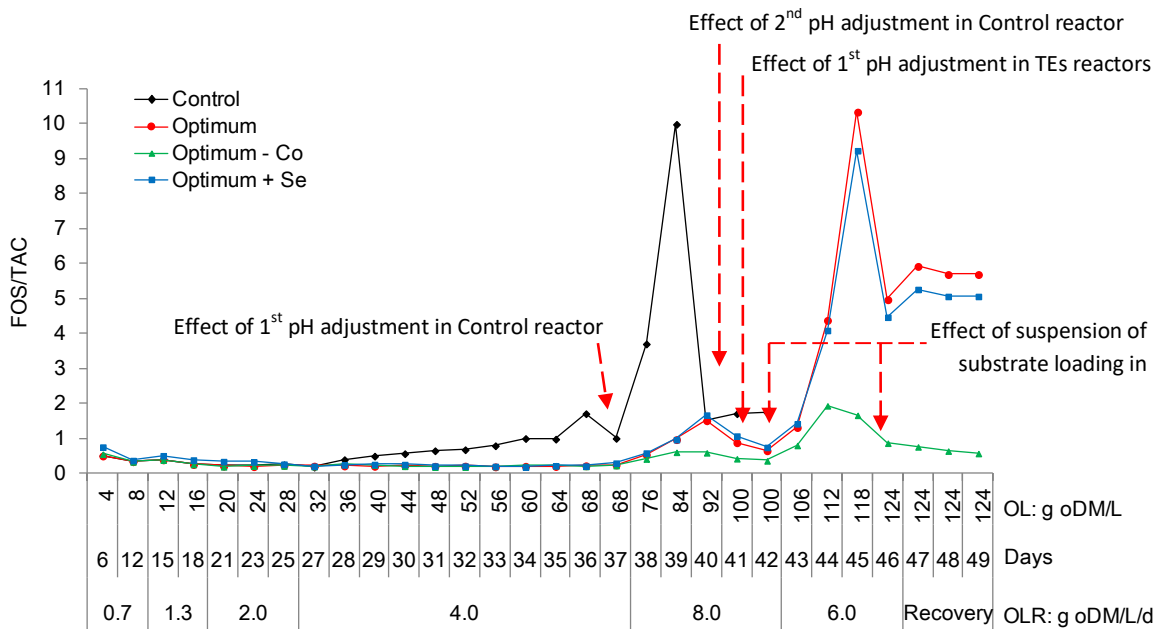


Figure 4 Overview of FOS/TAC of the reactors due to Ni, Co, Se and Mo supplementation and different OLR in the mesophilic semi-continuous experiment with mixed fruit residue.

Note: FOS/TAC values between 0.15 – 0.45 indicate stability in the biometanization process; and FOS/TAC > 0.6 indicate instability (Voss et al., 2009; Hach Lange GmbH, 2014).

Process Stability at OLR of 0.7g to 2.0g

oDM/L/d: When OLR was increased from 0.7 g to 1.3g oDM/L/d and maintained for 6 days, the FOS/TAC value in the Optimum +Se increased from 0.37 to 0.51 and stabilized at 0.39 at the end of the 1.3g oDM/L/d OLR (Figure 4, day 18). Conversely, within the same period, the FOS/TAC values in the Control, Optimum, and Optimum -Co reactors decreased to 0.26 from 0.3 that was measured at the end of 0.7 g oDM/L/d OLR. Increasing OLR from 1.3g oDM/L/d to 2g oDM/L/d and maintaining the OLR for 7 days did not increase FOS/TAC above 0.3 in any of the reactor, including the Optimum +Se reactor.

Process Stability at 4g oDM/L/d OLR:

Figure 4 shows the FOS/TAC trend at 4g oDM/L/d OLR in the four reactors. In the Control reactor, it was 0.18 in the start of the OLR of 4g oDM/L/d, but rose steadily to 1.71 at the end of this OLR when the total organic load was 68g oDM/L (day 37). This necessitated the first suspension of substrate loading followed by pH adjustment to 7.0 in the Control reactor. This intervention caused the FOS/TAC to decrease to 0.98. On the contrary, FOS/TAC value was less than 0.30 at all times of measurement in the Optimum, Optimum -Co and Optimum +Se reactors and remained so until the end of the OLR of 4g oDM/L/d.

Process Stability at 8g oDM/L/d OLR:

Figure 4 shows FOS/TAC trend in the Control

and its Variants. The Optimum -Co Variant had a peak FOS/TAC of 0.61 within the 8g oDM/L/d OLR, whereas the Optimum and Optimum +Se reactors had peak FOS/TAC values of 1.51 and 1.67 respectively. Due to the high FOS/TAC values in the Optimum and the Optimum +Se reactors, all the three reactors were operated one day without substrate loading (after loading of 100g oDM/L) in order to degrade accumulated organic load. The one-day break in organic loading induced recovery and returned the FOS/TAC values to 0.35 (Optimum -Co), 0.65 (Optimum) and 0.75 (Optimum +Se). Conversely, the Control reactor had a peak FOS/TAC of 9.99 at OLR of 8g oDM/L/d and this was accompanied by pH of 6.2. Adjustment of pH was initiated and resulted in the decrease in FOS/TAC value of the Control reactor from 9.99 to 1.51. Substrate loading was stopped in the Control reactor at a total organic loading of 84g oDM/L and recovery was monitored. Notwithstanding these interventions, the FOS/TAC in the Control reactor varied between 1.51 and 1.93 until the end of the study (day 49).

To evaluate the ability of the TEs configurations to restore normal biomethanization in the acidified digesters, the instability-inducing 8g oDM/L/d OLR was suspended and 6g oDM/L/d OLR was maintained until day 46 in the Optimum, Optimum -Co and Optimum +Se reactors. The FOS/TAC and pH values associated with the 6g oDM/L/d OLR are shown in Figure 5a - c.

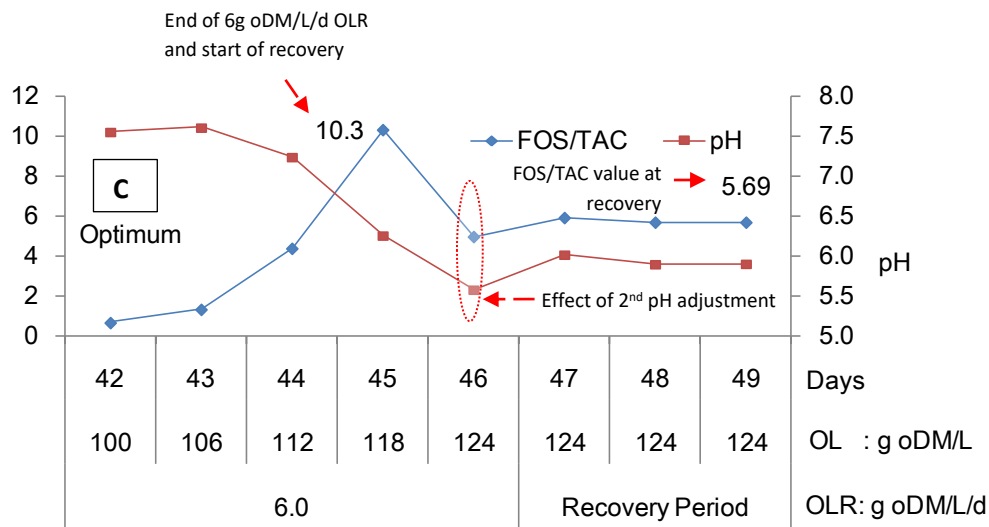
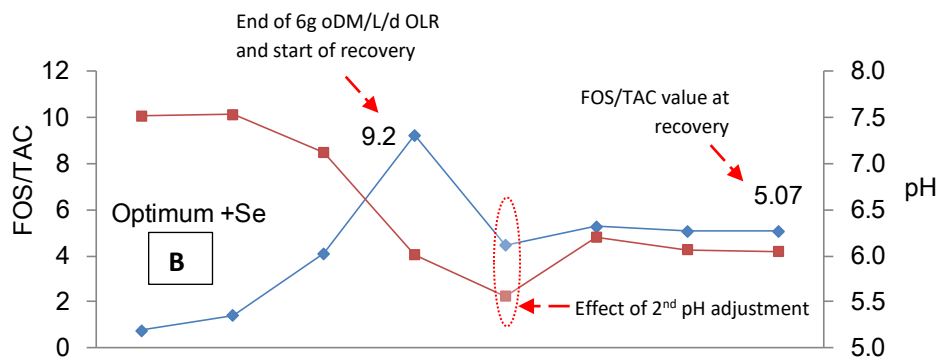
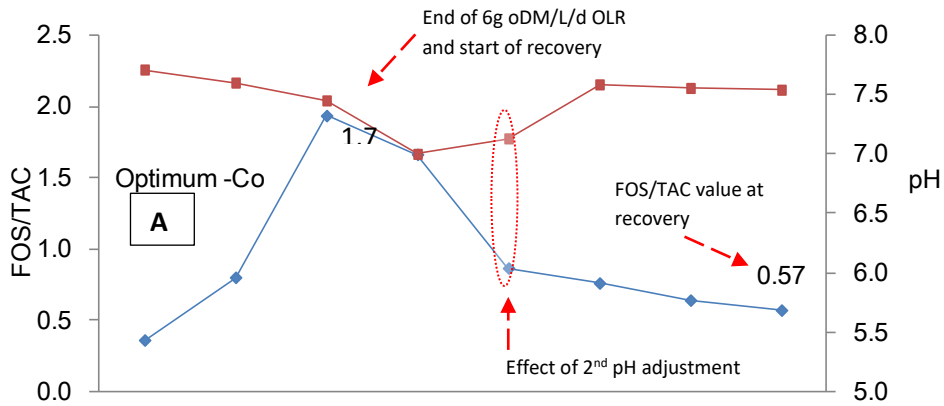


Figure 5a-c FOS/TAC of the (A) Optimum -Co, (B) Optimum +Se and (C) Optimum reactors due to Ni, Co, Se and Mo supplementation and OLR of 6g oDM/L/d in the mesophilic semi-continuous experiment fed with mixed fruit residue. (TEs configurations of the Optimum, Optimum +Se and Optimum -Co reactors are shown in Table 1)

Peak FOS/TAC of 9.2 and 10.3 were recorded in the Optimum +Se and the Optimum reactors respectively after substrate loading resumed at 6g oDM/L/d OLR. Peak FOS/TAC value of 1.66 was also recorded in the Optimum -Co within the same phase of OLR. The sharp rise in FOS/TAC was accompanied by noticeable decreases in pH values in Optimum +Se and Optimum reactors, similar to the Control treatment at failure point. The corresponding pH in the reactors at peak FOS/TAC were 6.26 (Optimum), 6.02 (Optimum +Se) and 7.0 (Optimum -Co). Consequently, substrate loading was stopped and pH was adjusted to 7.0 with 1M NaOH in the Optimum +Se and Optimum reactors. Notwithstanding the pH adjustment after organic loading was stopped (day 46), FOS/TAC in the Optimum +Se and Optimum reactors stayed at 5.07 and 5.69 respectively at the end of the study. Conversely, the FOS/TAC value dropped from 1.66 to 0.57 in the Optimum -Co reactor on day 49.

Individual VFAs Fingerprints of the Reactors: The plots of acetic acid, propionic acid, n-butyric acid, iso-butyric acid, n-valeric acid and iso-valeric acid (VFAs concentration) against Days are shown in Figure 6a-d for Optimum, Control, Optimum +Se and Optimum -Co reactors from day 12 (0.7g oDM/L/d) to 49 (recovery period). More pertinent due to associated instability are VFAs fingerprints during the 8g- and 6g oDM/L/d. The VFAs

concentrations were mainly due to accumulations of acetic acid, propionic acid and butyric acid.

In the Control reactor, acetic acid concentration (g/L) was between 4.2 ± 0.2 and 5.2 ± 0.7 during the OLR of 8g oDM/L/d (days 38 and 42). During the same OLR and days, acetic acid concentrations (g/L) were between 0.5 ± 0.05 and 2.0 ± 0.03 in the Optimum; 0.5 ± 0.04 and 2.5 ± 0.9 in Optimum +Se; and 0.6 ± 0.07 and 0.2 ± 0.02 in the Optimum -Co reactors. Peak propionic acid and butyric acid concentrations (g/L) during the OLR of 8g oDM/L/d were as follows: Control (1.2 for butyric only); Optimum (0.2 ± 0.03 and 0.09 ± 0.003 respectively), Optimum +Se (0.4 and 0.9 respectively), Optimum -Co (0.2 ± 0.02 and 0.06 ± 0.01 respectively).

During the OLR of 6g oDM/L/d, the individual VFAs had accumulated from the OLR of 8g oDM/L/d so that in the Optimum reactor, the individual VFAs concentrations (g/L) on day 49 were acetic acid, 6.3 ± 1 ; propionic acid, 0.25 ± 0.02 ; and butyric acid, 0.7 ± 0.05 . In the Optimum -Co reactor the VFAs concentrations (g/L) were acetic acid, 0.6 ± 0 ; propionic acid 0.7 ± 0.04 and butyric acid, 0.6 ± 0.05 . In the Optimum +Se reactor, these were (g/L), acetic acid, 6.1 ± 0.09 ; propionic acid, 0.3 ± 0.03 ; and butyric acid, 0.8 ± 0.02 . The Control reactor was not operational in this phase.

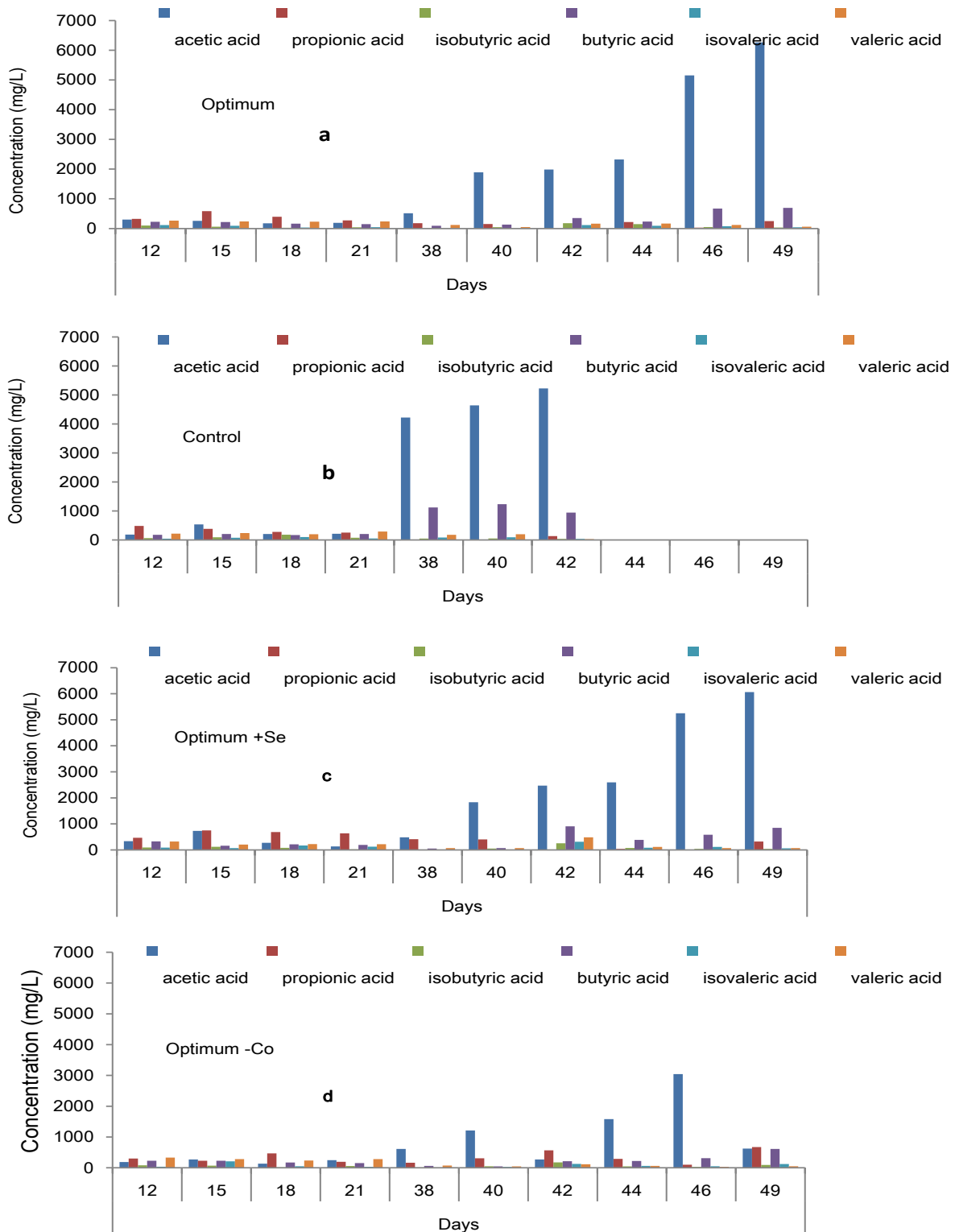


Figure 6a –d Individual VFAs fingerprints in the experimental reactors due to OTC and its Variants during the mesophilic semi-continuous experiment with mixed fruit residue: **a)** Optimum reactor; **b)** Control reactor; **c)** Optimum +Se reactor; **d)** Optimum –Co reactor

Although the Optimum -Co reactor had the lowest acetic acid and butyric acid accumulation compared to the Optimum and Optimum +Se reactors, it had a lower propionic acid oxidation capability compared to the other reactors. This is highlighted in Figure 6e for days 40 to 49

during the 8g -, 6g oDM/L/d and recovery period. Within this period, the peak propionic acid concentrations (g/L) of the reactors include: Optimum, 0.3; Optimum +Se, 0.4; and Optimum -Co, 0.7.

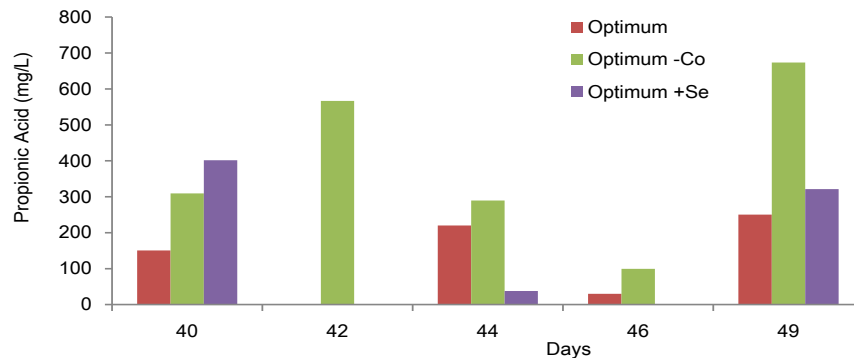


Figure 6e Propionic acid fingerprint during mesophilic biomethanization of mixed fruit residue in the Optimum, Optimum -Co and Optimum +Se reactors between days 40 and 49

Discussion

The results show potentials for maintaining process stability during high organic loading by manipulating and adapting previously identified VFAs*TEs. The observed biochemical behaviors of Optimum, Optimum -Co and Optimum +Se formulations could be explained in relation to the VFAs*Co and VFAs*Se. The biochemical behavior of VFAs*Se in this study is consistent with the RSM in Figure 1 and shows that increase in Se concentration in the Optimum +Se reactor (relative to Se Conc. in the Optimum) improved CH₄ production and FOS/TAC relative to the Optimum. The adaptation or lag period of 18 – 25 days prior to the observation of positive effect of the VFAs*Se is consistent with earlier findings on adaptation to TE toxicity during methanization (Ezebuiri and Koerner, 2017).

Antagonistic pre-adaptation effects of the VFAs*Se at the low OLRs (Table 2) as seen in Optimum +Se Variant of OTC is also consistent with the RSM in Figure 1 that suggests that relative to an OTC, Se concentration should decrease at low OLR and *vice versa*. The Se concentration in the Optimum +Se reactor was 1.5 mg/L, which is higher than 0.40 mg/L that is Optimum for 120 mmol/L VFAs or medium OLR. Hence, lower OLR such as 0.7g – 2.0g oDM/L/d require Se concentration < 1.5 mg/L. Supposedly, the inhibitory high Se concentration

of the Optimum +Se Variant (1.5 mg/L, compared to 0.4 mg/L in Optimum and Optimum -Co) was antagonistic and accounted for the observed poorer CH₄ production and lower process stability in the low OLR phase compared to the Optimum and the Optimum -Co reactors. From Table 1, 0.40 mg/L Se, as in Optimum and Optimum -Co, is most appropriate for medium VFAs level of 120 mmol/L (or medium OLR in Table 2). However, Optimum +Se Variant showed better CH₄ production (Figure 3a-c) and process stability (Figure 5a-c) than the Optimum at OLR ≥ 4.0g oDM/L/d. This is consistent with RSM in Figure 1 that suggests that a higher Se concentration relative to Se concentration in the OTC is required as VFAs concentration or OLR increases.

Similarly, from Figure 1, Co has an inverse and negative relationship with VFAs. The VFA*Co in Figure 1 was adapted to high OLR by lowering Co concentration from 2.7 mg/L in the Optimum, to 0.5 mg/L in the Optimum -Co. This was in anticipation of a corresponding positive biochemical effect during a high OLR. The adapted VFA*Co in the Optimum -Co Variant was responsible for the superior CH₄ production and FOS/TAC, especially at 6g oDM/L/d and the recovery period when FOS/TAC remained within the stability range in spite of the high OLR regime. Figure 6d and 6e showed that although the Optimum -Co Variant had a higher potential

for CH₄ production, process stability and recovery from acidification, it had weak propionic acid degradation ability. This observation is consistent with earlier publications on the role of Co in propionic acid degradation (Halarnkar and Blomquist, 1989; Ezebuoro et al., 2018). Hence, lower propionic acid oxidation potential as was evident in the Optimum –Co reactor (Co, 0.5 mg/L) compared to Optimum and the Optimum +Se reactors (2.7 mg/L Co each) was anticipated. At OLR of 4.0g oDM/L/d, the OTC outperformed its variants on CH₄ production and process stability. This is consistent with the RSM model in Figure 1 that produced the OTC in Table 1, which was designed to be most efficient for medium OLR (120 mmol/L) for all methanization processes.

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