



Method development of extraction and identification of Nitidine, (Benzophenanthridine alkaloid) from the barks of *Fagara chalybea*.

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ABSTRACT

Objective: *Fagara chalybea* is an important medicinal plant belonging to the family Rutaceae. The plant is well known for its anti-malarial, anti-microbial and anti-cancerous activity, which has been attributed to the presence of benzophenanthridine alkaloid nitidine in the plants. The present work aims to develop a method of Nitidine extraction and Identification from the bark of *Fagara chalybea* Engl.

Methodology and results: A simple, rapid and sensitive HPLC method has been developed for the qualitative determination of nitidine in the dried bark of *Fagara chalybea* after extraction. The calculated yield is 2.28%. The retention time of nitidine in the methanol was 27.639 min, and then this time was 27.393 nm in dichloromethane. The limit of detection and limit of quantization were found to be 2.18 and 7.29 µg/mL respectively, the correlation coefficient was 0.998.

Conclusion and application of results: The application of this method to the analyses of nitidine after extraction proved that the method is sensitive enough to detect low levels of analyses. To value traditional medicine, this method can be used as a tool for quality control of botanicals herbal formulations.

Key words: Nitidine, HPLC, *Fagara chalybea*, Malaria.

INTRODUCTION

Malaria is still a major threat to human life and health, especially in sub-Saharan Africa. It is a disease that mainly affects immunologically naive individuals, especially children under 5 years old, and women who are in their first or second pregnancy (Zhang *et al.*, 2014). Although recent estimates suggest that malaria mortality rates decreased by an impressive 47% between 2000 and 2013 globally, and by 54% in the World Health Organization's (WHO) African Region, malaria

remains a major public health problem in many countries (Djimde *et al.*, 2016). The chemotherapeutic arsenal for malaria treatments is limited to three main families of compounds: the quinolines (quinine, chloroquine, mefloquine), the antifolates (sulfadoxine, pyrimethamine) and the artemisinin derivatives. Widespread drug resistance resulted in the ineffectiveness of many anti-malarials and chemotherapy now requires drug combinations (Grellier *et al.*, 2008). Malaria vaccine development

has been hampered by the frequent antigenic variation of the *P. falciparum* parasites (Zhang et al., 2014; Bouyou-Akotet et al., 2015). Resistance of this parasite to virtually all of the currently available anti-malarial drugs is of great concern; consequently, new inexpensive drugs are urgently needed to address the global burden of malaria (Bouquet et al., 2012). Several natural or synthetic repellents are used against biting insects pests (Curtis et al., 1989). Synthetic products are generally sold in order to protect against mosquito bites. Nevertheless, they are prohibitively expensive for the communities concerned (Hougard et al., 1998) and have to be used carefully (Molmooi et al., 2010, Konan et al., 2003). Natural substances have provided the best anti-malarials that are currently available. These anti-malarials, including artemisinin and quinine, as well as numerous molecules derived from plants, are promising lead compounds that combat malaria

MATERIALS AND METHODS

Chemicals reagents: All solvents used for chromatographic analyses were HPLC grade reagent, Methanol (CHROMASOL, for HPLC, $\geq 99.9\%$), Dichloromethane (puriss, p.a, ACS reagent, ISO, $\geq 99.9\%$), Acetic acid (puriss, 99-100%), standard Nitidine and Amberlite IRA420®, were from Sigma Chemical Co (St. Louis, MO, U.S.A.).

Plant material and extraction procedures: The Barks of plant species (*Fagara chalybea*) collected from a National Park of Akagera in the East of Rwanda on October 2010, was identified by a resident botanist, University of Liege in botanical department. Powder of the bark (500g) was extracted for 3 days with 6 litre of methanol acidified by acetic acid (1%) at room temperature, followed by rapid paper filtration through Whatman paper. The resulting solutions were evaporated under vacuum at 70 °C by rotavapor to dryness. Twenty (20) g of dry methanolic residues were taken then dissolved in 100 mL of water and added 100 mL of dichloromethane for liquid-liquid extraction. After shake and decantation, the extract solution was concentrated to dryness in a rotary evaporator. The crude extract was conserved at 4°C until use. The required concentrations

RESULTS

Calibration curves: Five different concentration levels (50, 100, 150, 200 and 250 $\mu\text{g/mL}$) were obtained of standard solution, conveniently diluted with methanol.

(Lu et al., 2012; Lin et al., 2014). Nitidine is a benzophenanthridine alkaloid (Figure 1) found in specie of *Zanthoxylum nitidum* (knob wood). It was first isolated in 1959 from *Zanthoxylum nitidum* (Lu et al., 2012; Li et al., 2014). Nitidine was found again approximately 40 years later in a traditional Kenyan anti-malarial remedy and was more recently discovered in *Zanthoxylum rhoifolium*, a traditional remedy from South America. The widespread use of nitidine stresses the importance of this molecule in the field of malaria control (Bouquet et al., 2012; Lin et al., 2014; Lu et al. 2012). Many biological properties have been ascribed to Nitidine, including its use as anti-microbial, anti-cancerous and anti-malarial properties were also reported (Bouquet et al., 2012; Lin et al., 2014; Lu et al., 2012; Li et al., 2014). The present work aims to develop a method of Nitidine extraction and Identification from the bark of *Fagara chalybea* Engl. (Knob wood).

(4g) were taken again with each reagent methanol and dichloromethane for HPLC essay. (Li et al., 2014 ; Lu et al., 2012; Nono et al., 2014).

Chromatographic conditions: All analysis was done at ambient temperature (24 °C) under isocratic conditions. The stationary phase consisted of column: diphenyl (SS 250 x 4.6 mm), the mobile phase consisted of a mixture of acetonitrile/trifluoroacetic acid (80:20, v/v) and pH adjusted to 2.21. The flow rate was 1.0 mL/min and volume of injection was 10 μL . All solutions, including mobile phase, were sonicated during 20 min before use. The DAD detection was used (Baerheim-Svendsen & Verpoorte 1984; Makoto, 2007; Praveena et al., 2014).

Standard solutions preparation for calibration curves: Accurately weighed amounts of standards of nitidine equivalent to 5mg were transferred to 5mL volumetric flask. The volumes were completed with methanol. The resulting solutions were sonicated during 20 min and filtered through membrane filter. Final concentrations were 1 mg/mL. Aliquots of solution was accordingly diluted with methanol in order to obtain solutions with final concentration of 50 $\mu\text{g/mL}$, 100 $\mu\text{g/mL}$, 150 $\mu\text{g/mL}$, 200 $\mu\text{g/mL}$ et 250 $\mu\text{g/mL}$.

Each solution was injected in the chromatographic system ($n=5$) and mean values of peak areas were plotted against concentrations. The curve was adjusted

by linear regression with least mean squares method (Figure 1 and Table 1). (Santoro *et al.*, 2006). The correlation coefficient is 0.998

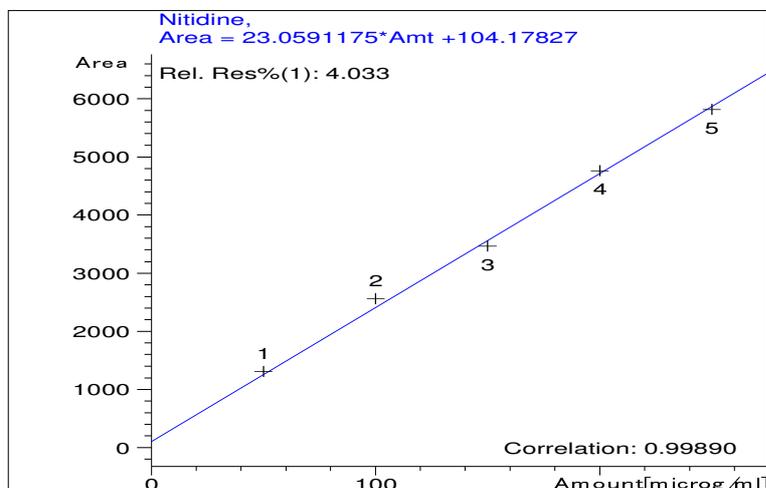


Figure 1: Calibration curve of nitidine in methanol.

Detection limit (DL) and quantification limit (QL): The DL and QL were calculated using Eqs. (1) and (2), respectively. The theoretically determined values of

detection and quantitation limits were crossed checked by actual analysis of these concentrations using proposed methods (Santoro *et al.*, 2006):

$$(1) \quad DL = \frac{SD * 3}{\alpha}$$

$$(2) \quad QL = \frac{SD * 10}{\alpha}$$

Where S.D is the standard deviation of curve and α is the slope of curve (Table 1).

was about 27 minutes in methanol and dichloromethane (Figure 2 and 3).

Chromatogram of sample standard of nitidine: Injected at different concentrations, nitidine retention time

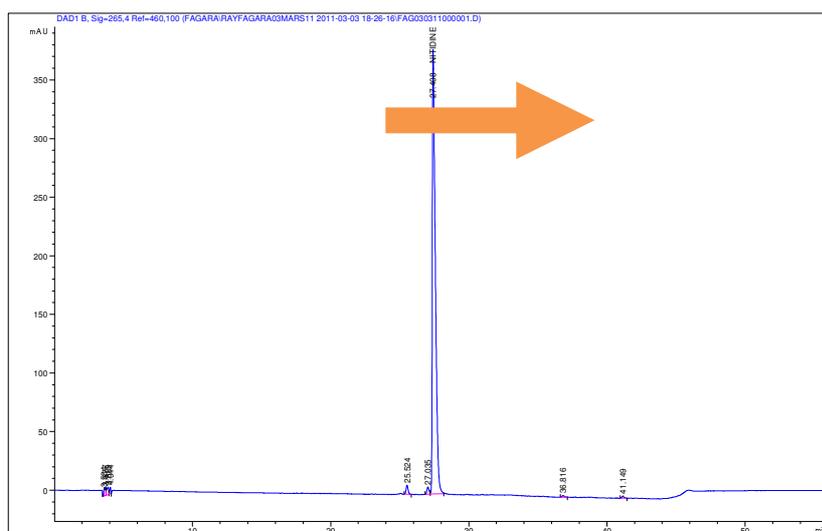


Figure 2: Chromatogram of sample standard of nitidine in methanol.

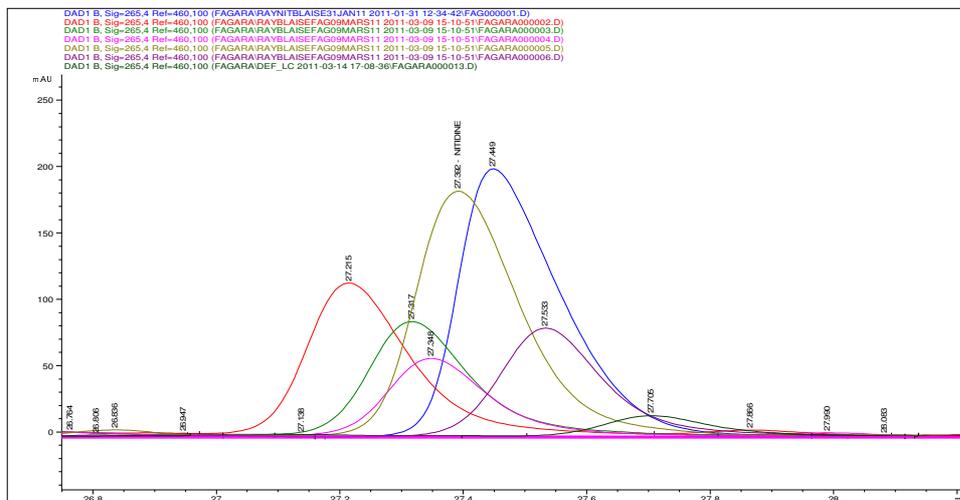


Figure 3: Superimposed spectra of nitidine standard in methanol at different concentrations.

Chromatogram of sample crude of the bark of *Fagara chalybea* powder: After injection of crude extract in methanol and dichloromethane, the peaks of nitidine are

distinct with retention time of about 27 min (Figure 4 and 5).

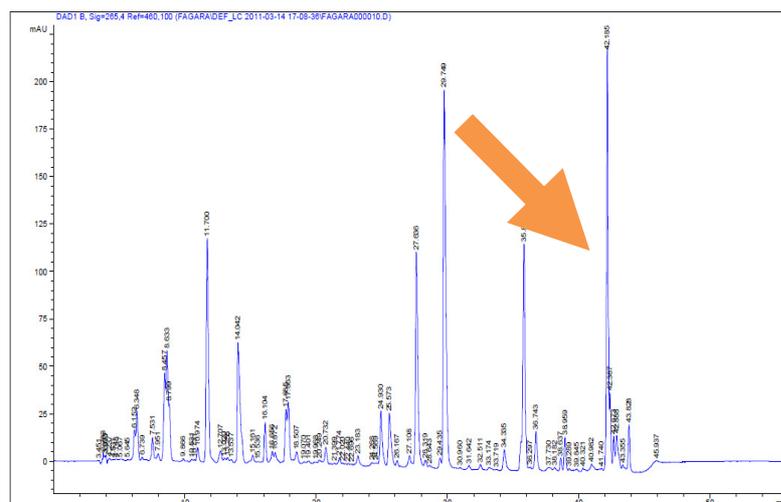


Figure 4: Chromatogram of sample crude (methanol extract) of the root bark of *Fagara chalybea* powder.

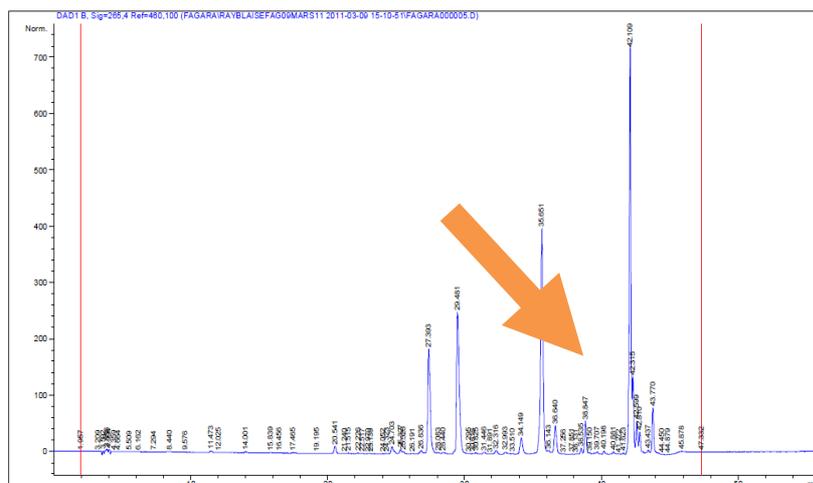


Figure 5: Chromatogram of sample crude (dichloromethane extract) of the root bark of *Fagara chalybea* powder.

Spectres of nitidine: Spectres (UV-vis) of standard nitidine and extract nitidine are similar (Figure 6).

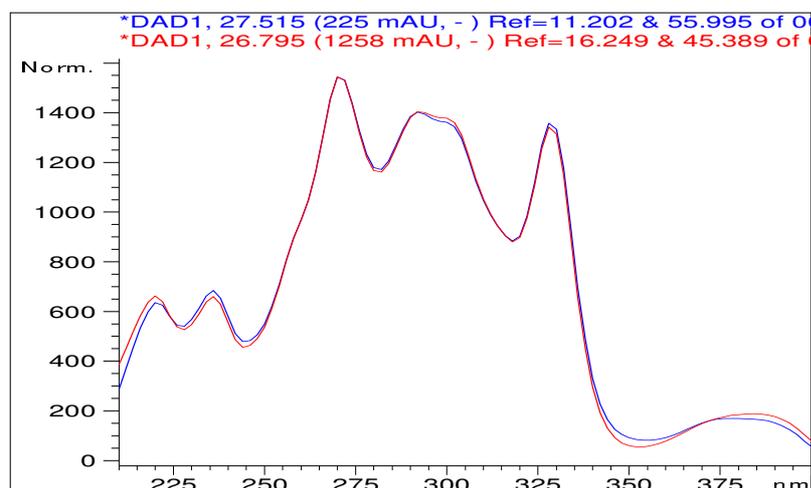


Figure 6 : Superimposed spectra of nitidine in standard and sample extract.

Table 1: Data for HPLC quantification and extraction of nitidine.

Parameter	Value
Linearity range (ng/spot)	50 - 250
Correlation coefficient (r)	0.9989
Slope	23.05
Intercept	104.17
LOD [$\mu\text{g/mL}$]	2.18
LOQ [$\mu\text{g/mL}$]	7.29
Nitidine yield (500 g)	1.8 %
Nitidine yield (0.39342 g)	2.28%

DISCUSSION

For preparative HPLC, 393.42mg of dichlorometanolique fraction were deposited on silica gel. This has afforded 9mg nitidine a yield of 2.28% by referring to the deposit.

With 500g powder of the plant, the calculated yield is 1.8 %. The retention time of nitidine in the methanol was 27.639 min, and then this time was 27.393 nm in

dichloromethane. The mobile phase acetonitrile/trifluoroacetic acid (80:20, v/v) gave good resolution for nitidine with a sharp and well defined peak. The bands in sample chromatograms was confirmed by the comparison of chromatogram and UV spectra obtained from the sample with that obtained from the standard solution. The peak corresponding to nitidine from the sample solutions had the same retention factor as that of standard nitidine. Specificity of the method was ascertained by comparing retention time values of samples with that of standard nitidine. No interference with peak of nitidine from other constituents of extracts was also observed indicating the specificity of the method. These results are similar to that obtained in a study on the quantitative determination of nitidine from roots and plant tissue of *Toddalia asiatica* (Linn.) the specificity of the method was ascertained by comparing R_f values and the UV-Visible spectra of samples with that of standard nitidine. No interference with peak of nitidine from other constituents of extracts was also observed (Praveena et al., 2014). Linearity data show concentration interval of studied nitidine in which the intensity of the detector response is proportional to

the concentration of the analyzed substance. The correlation coefficient obtained was 0.998. The calibration equation shows an excellent linearity with slope 23.05 and intercepts 104.17. In the study conducted by the team of Praveena in 2014 the author found a correlation coefficient equal to 0.9949. Their results can be superimposed on those obtained in this study (Praveena et al., 2014). Signal to noise ratios 3 and 10 were considered for DL and QL respectively and calculated from the slope of the calibration plot and the standard deviation of the response. DL and QL were found to be 2.18 and 7.29 $\mu\text{g/mL}$ respectively for nitidine, which ascertain the adequate sensitivity of the method. Santoro and collaborator in 2005 had obtained the DL and QL of samples of gatifloxacin; levofloxacin; lomefloxacin and pefloxacin (antibiotics in pharmaceutical preparations by high-performance liquid chromatography) were 0.13, 0.39; 0.15, 0.46; 0.17, 0.52 and 0.08, 0.25 $\mu\text{g/mL}$, respectively (Santoro et al., 2005). Their results are significantly lower than those obtained in the present study. This difference can be explained by the purity of the substance they used and chromatographic conditions.

CONCLUSION

This paper reported the development of a rapid, simple and specific analytical HPLC method for the qualitative estimation of nitidine. The application of this method to the analysis of nitidine after extraction in the barks of *Fagara chalybea* samples proved that the method is

sensitive enough to detect low levels of analyses in complex matrices. To value traditional medicine, this method can be used as a tool for quality control of botanicals herbal formulations.

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