

## BIOSURFACTANT PRODUCTION AND CONCOMITANT DEGRADATION OF SPENT ENGINE OIL BY *Alcaligenes faecalis* ULAG3AF ISOLATED FROM LAGOS LAGOON

Ashamu, T. O.<sup>1,\*</sup>, Omotayo A. E.<sup>1</sup> and Adekunle, A. A.<sup>2</sup>

<sup>1</sup>Department of Microbiology, University of Lagos, Akoka-Yaba, Lagos, Nigeria.

<sup>2</sup>Department of Botany, University of Lagos, Akoka-Yaba, Lagos, Nigeria.

\*Corresponding Author's Email: temitopeashamu@gmail.com

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### ABSTRACT

*Alcaligenes faecalis* strain ULAG3AF was isolated from the Lagos Lagoon, an estuarine environment. The aim of the study was to investigate the biosurfactant production potentials of this organism and explore the antimicrobial potential of the biosurfactant. The bacterial isolate was screened for its biosurfactants production potentials using the drop collapse, oil displacement tests and emulsification index. The strain was cultivated in an enrichment medium, Bushnell Haas Medium, supplemented with 1%, 2%, 3% and 4% v/v spent engine oil and grown at different temperatures including 15 °C, 25 °C, 30 °C, 35 °C and 40 °C and salinity (10 mg/l, 20 mg/l, 30 mg/l and 40 mg/l of NaCl). The optimisation process involved analysing the effect of substrate concentration, temperature, and salinity on biosurfactant production by the bacterium using the one-variable-at-a-time (OVAT) approach. Gas Chromatography and Mass Spectrometry (GC-MS) and emulsification index were used to determine the rate of hydrocarbon degradation and the concentration of biosurfactant produced by the organism respectively. The bacterium demonstrated the ability to biodegrade different polyaromatic hydrocarbon components of spent engine oil and produced biosurfactants with an emulsifying index of 80.9% when cultivated in a medium containing 2% v/v of spent engine oil, 2% salinity (20 mg/l of NaCl) and a temperature of 40 °C after 7 days. The biosurfactant produced was a glycolipid with antimicrobial activities against *Klebsiella pneumoniae*, *Pseudomonas aeruginosa*, *Bacillus subtilis* and *Staphylococcus aureus*.

**Keywords:** Biosurfactant, Glycolipid, Biodegradation, Gas Chromatography and Mass Spectrometry, Polyaromatic hydrocarbon.

### INTRODUCTION

Like chemical surfactants, biosurfactants are amphipathic molecules with both hydrophobic and hydrophilic moieties, which are produced by plants and microorganisms (Santos *et al.*, 2016). However, unlike chemical surfactants, they are less toxic and more environmentally friendly because of their degradability and stability over a wide range of environmental conditions, such as pH, temperature, and salinity (Franzetti *et al.*, 2018). As a result, there has been a growing interest in the study of biosurfactants over the past three decades. One of the important applications of biosurfactants is in the bioremediation of hydrocarbon-polluted environments (Peele *et al.*, 2016). Hydrocarbon pollutants are released into the environment as a result of the exploration, use, storage and disposal of petroleum hydrocarbon and its products, thereby affecting the ecology of such environments owing to their toxicity (Oyetibo *et al.*, 2017). While bioremediation offers a solution to the menace of

hydrocarbon pollutants, the process is often limited by the hydrophobicity of the pollutants, which limits their bioavailability for degradation by microorganisms (Babaei & Habibi, 2019). Biosurfactants can increase the bioavailability of hydrocarbon pollutants by emulsifying them via pseudo-solubilisation and direct interfacial contact (Santos *et al.*, 2016). Typically, microorganisms that produce biosurfactants are hydrocarbonoclastic organisms that can metabolize hydrocarbons as their sole source of carbon (Joy *et al.*, 2017). The estuarine environment is a potential reservoir for hydrocarbonoclastic microorganisms, and consequently, biosurfactants producers because they harbour large deposits of hydrocarbon pollutants generated from anthropogenic activities in and around the estuarine (Di Gregorio *et al.*, 2016).

The Lagos Lagoon, which is the largest Lagoon in West Africa, is a brackish coastal lagoon that is exposed to diverse anthropogenic activities

including water transportation, industrialization and waste disposal (Alo *et al.*, 2014). As a result, the lagoon is subjected to a significant level of hydrocarbon and metal pollutants (Oyeyiola *et al.*, 2013) which is further enhanced by the unregulated environmental protection laws in the country (Edmund Emeka, 2015). For instance, Sogbanmu *et al.* (2016) reported that the Apapa Port, which is the busiest port along the Lagos Lagoon, is heavily contaminated due to the numerous petroleum tank farms located along the Port's terminals. Alo *et al.* (2014) also highlighted that the contamination of this large water body adversely affects the ecology of the Lagos Lagoon which is also a source of fisheries for the city. In a study by Sogbanmu *et al.* (2019), the authors reported that fish from four locations around the Lagos Lagoon were minimally contaminated with Polyaromatic Hydrocarbons (PAHs), which can pose a health risk to humans if such fishes are consumed.

Given the heavy contamination of the Lagos Lagoon, it is a rich source of hydrocarbon-degrading, biosurfactants-producing microorganisms (Di Gregorio *et al.*, 2016). Several studies have reported the isolation of hydrocarbon-degrading bacteria from the Lagos Lagoon (Adebusoye *et al.*, 2007; Buraimoh *et al.*, 2020; Obayori *et al.*, 2009, 2021; Obi *et al.*, 2016). Biosurfactants-producing microorganisms isolated from estuarine environments like the Lagos Lagoon are desirable because the microorganisms from the estuarine environment can withstand extreme conditions owing to the fluctuating pH, salinity and temperature in that environment, thereby increasing the applications of the biosurfactants (Antoniou *et al.*, 2015). Furthermore, hydrocarbon-contaminated environments also have extreme pH and salinity (Fathepure, 2014), thus, biosurfactants-producing organisms which can tolerate and produce biosurfactants under such conditions are important. This study, therefore, aims to investigate the biosurfactant production potentials of a bacterial isolate from the hydrocarbon-polluted Lagoon water; and explore the antimicrobial potential of the biosurfactant

## MATERIALS AND METHODS

### Sample Collection

Water samples were collected from the harbour region of three oil-contaminated sites on the Lagos Lagoon, Lagos State, Nigeria. The three sites included Apapa (3°22'40.233"E, 6°26'50.476"N), Kirikiri (3°18'22.674"E, 6°26'15.574"N) and Ijegan (3°14'40.547"E, 6°24'45.254"N). Approximately 1L of Lagoon water was collected in a sterile glass bottle and stored at 4 °C until use. The samples were collected at five different points at the sampling site and pooled together before being transported to the laboratory for further analysis.

### Physicochemical Analysis of Samples

Physicochemical parameters including pH and conductivity were determined with pH (Pro2Go Portable Mettler Toledo, Germany) and conductivity meter (INE-EC100B conductivity meter) respectively, and total hydrocarbon (THC) was measured gravimetrically as described previously (BSI, 1990; Chopra and Kanwar, 1998). The Varian 720-ES Inductively coupled plasma optical emission spectrophotometer was used for the heavy metal analysis while the nitrate and phosphate Analysis were done using the Hach Model DR/2010 spectrophotometer.

### Molecular Identification of Isolates

Bacterial DNA was extracted using the ABL-GA-GBB100 Presto Mini Genomic DNA Extraction Kit following the manufacturer's instructions. A NanoDrop spectrophotometer (Thermo Scientific, Wilmington, DE, USA) was used to quantify DNA concentration at 260/280 nm. The quality of the PCR amplicon was viewed using 1% agarose gel electrophoresis, and the DNA was stored at -20 °C. The conserved region (16S rRNA gene) targeted for analysis was the V4-V5 hypervariable region of the 16S rRNA gene, which was amplified using primers 515F-Y and 926R with Illumina adapters and barcodes. Pair-end sequencing was performed using a 600-cycle MiSeq® Reagent Kit V3 (Illumina, U.S.A.).

### Enrichment and Isolation of Hydrocarbon-Degrading, Biosurfactant-Producing Bacteria

The enrichment was carried out in 500 ml conical flasks containing 100ml Bushnell Haas Broth containing  $K_2HPO_4$  (1 g/L);  $KH_2PO_4$  (1 g/L);  $NH_4NO_3$  (1 g/L); NaCl (30.0 g/l);  $FeCl_3$  (0.5 g/l);  $CaCl_2$  (0.02 g/l),  $MgSO_4$  (0.02 g/l) as described by Adebusoje *et al.* (2007). The medium was amended with crude oil (1% v/v). This was then inoculated with 10 ml water sample and was sealed with a cotton plug before incubation at room temperature ( $28 \pm 2$  °C) at 150 rpm for 7 days. After 7 days, 10 ml aliquot was transferred into 100 ml of fresh Bushnell Haas broth amended with 1% (v/v) of crude oil and incubated under the same conditions. The transfer was repeated three times and 1 ml from the final transfer was plated out on Bushnell Haas agar and the pure isolates were selected for further studies.

### Screening of Isolates for Biosurfactant Production and Biodegradation Studies

Axenic cultures of the isolate was further incubated in Bushnell Haas broth amended with 1% (v/v) of crude oil at 150 rpm and 28 °C for 7 days. The broth was then centrifuged at  $10,000 \times g$  for 10 min at 4 °C and the cell-free supernatant (CFS) was filtered through Whatman filter paper No.1. The filtrate was then analysed for biosurfactant production using oil displacement test, drop collapsing assay and emulsification properties tests (Ram *et al.*, 2019). The experiment was set up in triplicate, and all broths in each set were sacrificed for total viable count (TVC) determination, biosurfactant quantification, and biodegradation studies every two days. The TVC was determined using the standard plate count technique to determine the microbial populations in the samples (Bloem, Hopkins and Benedetti, 2005), whereas the rate of degradation of hydrocarbons was determined by Gas Chromatography and Mass Spectrometry (GC-MS) using the Varian CP-3800 GC. The ability of the bacterial isolate to metabolize the Polyaromatic Hydrocarbons (PAH) in spent engine oil during the production of biosurfactants was determined. Using GC-MS, the quantities of selected PAH known to be present in motor engine oil were assessed in the samples over 14 days when the isolates were cultivated at 30 °C with

1% salinity and 1% (v/v) spent engine oil.

### Drop Collapse Test

Two  $\mu$ l of olive oil were added to a 96-well microtitre plate, and the plate was equilibrated for 1 h at 37 °C and 5  $\mu$ l of the CFS was added to the surface of the oil. The shape of the drop on the oil surface was observed after 1 minute. The culture supernatant that made the drop collapse was indicated as positive result and the drops that remained beaded were recorded as negative, and were compared with distilled water as control.

### Oil Displacement

This was done using the methods proposed by Youssef *et al.* (2004). Forty microlitre of distilled water was placed in a Petri dish and 15  $\mu$ l of spent engine oil was dropped on the surface of the distilled water to form an oil film. Ten microlitre of the cell-free supernatant was then gently dropped on the surface of the oil film. An area of clear halo was observed after 30 seconds and the diameter recorded.

### Emulsification Index

As described by Kiran *et al.* (2009), 2 ml of the spent engine oil was added to an equal volume of cell-free supernatant and homogenized for two minutes in a vortex at very high speed. The emulsification stability was measured after 24 h, and the emulsification index was calculated by dividing the measured height of the emulsion layer by the total height of the liquid layer and multiplying by 100. The emulsification activity of the isolates was compared to that of Tween 80. The EI was also used to quantify the amount of biosurfactant produced.

### Extraction of Biosurfactant

The biosurfactant in the cell-free filtrate was extracted using liquid-liquid extraction. The cell-free supernatant was acidified to pH 2.0 using 1 N HCl, and then mixed with an equal volume of chloroform and methanol (2:1) mixture. The organic phase containing the biosurfactant was then collected in a separation funnel. The aqueous phase was extracted three times to ensure a maximum yield, while the organic extract was dried using a rotary evaporator. Biosurfactant yield was determined using the equation proposed by Irerere *et al.* (2017).

Yield = Biosurfactant mass/ volume of cell-free supernatant

Eight strains which could utilise spent engine oil and produce biosurfactants were identified. The isolate with the highest biosurfactant yield and emulsification index was selected for further studies.

### Optimisation of Biosurfactant Production Using One-Variable-at-a-Time Method

Optimisation involved analysing the effect of concentration, temperature, and salinity (% NaCl) on biosurfactant production in the isolates using the one-variable-at-a-time (OVAT) (Rane *et al.*, 2017). Once one of the assessed parameters is optimised, it is set as the optimal level for the next round of OVAT studies to optimise other variables. The substrate/carbon source used was Spent Engine oil (SEO) added to the production medium (Bushnell Hass Medium) at different concentrations (1%, 2%, 3% and 4% volume per volume). The temperature range were 15 °C, 25 °C, 30 °C, 35 °C and 40 °C while the salinity was 10 mg/l, 20 mg/l, 30 mg/l and 40 mg/l of NaCl in the medium. These were the independent variables whereas the emulsification index was the dependent variable.

### Characterisation of Biosurfactants

The characterisation of the biosurfactant produced involves identifying the chemical structure of the biosurfactant and its bioactive properties and was performed using Attenuated Total Reflectance Fourier Transform Infrared (ATR-FTIR) spectroscopy to analyse the functional classes and types of bonds present in the biosurfactant produced. Spectroscopy was conducted using the Agilent Technologies Cary 630 FTIR Cary630 ZnSe spectrometer in the range of 4000 - 650  $\text{cm}^{-1}$ . The analysis was conducted in accordance with the manufacturer's instructions.

### Antimicrobial Activity of Biosurfactant

The antibacterial activity of the biosurfactant was determined using the agar well diffusion technique as described by Das *et al.* (2016). The bacterial strains used to test the antibacterial activity of the biosurfactant were *Staphylococcus aureus*, and *Pseudomonas aeruginosa* obtained from the Covenant University Microbial Collection Center. The test organisms were cultivated on Muller-Hinton agar plate into which wells had been made with sterile cork-borer. About 100  $\mu\text{l}$  of the biosurfactant was added into the wells and incubated for 24 hours at 37°C. Streptomycin was used for positive control whereas sterile distilled water was used for negative control. The antibacterial activity was evaluated based on the presence of a clear zone which indicates inhibition of growth and the diameter of the zone of inhibition was measured in millilitres.

## RESULTS

### Physicochemical and Microbial Analysis of Samples

Water samples collected from three sites on the Lagos Lagoon were heavily contaminated with hydrocarbons and heavy metals (Table 1). The Apapa site had the highest hydrocarbon pollution of the three sites. It also had the highest heavy metal contamination except for cadmium and chromium which are more abundant at the Ijegan site (0.16 and 1.10mg/l respectively). All three sites had heavy metal pollution at levels that were significantly higher than the national and (World Health Organisation) WHO standards. At the Apapa site, heavy metal contamination was Fe>Pb>Cr>Cd, at the Ijegan site heavy metal contamination was Cr>Pb>Fe>Cd while at the Kirikiri site, it was Cr=Pb>Fe>Cd. Surprisingly, however, the Apapa site had a lower concentration of hydrocarbon-utilizing bacteria compared to the Ijegan site, although it had the highest concentration of heterotrophic bacteria (Table 2).

**Table 1:** Physicochemical Analysis of Water Samples.

S/N	PARAMETER	WATER SAMPLE			NESREA	WHO
		CODES				
		AP1	JG1	KR1		
1	pH	6.80	7.20	7.13	-	-
2	Conductivity ( $\mu\text{s}/\text{cm}$ )	59,100	35,906	43,845	1000	NA
3	Salinity (mg/l)	35.73	20.96	32.73	-	-
4	THC (mg/l)	12.13	11.40	7.38	-	-
5	Nitrate (mg/l)	5.94	6.09	4.87	10-20	10
6	Phosphate (mg/l)	2.98	3.15	3.12	-	1
7	Potassium (mg/l)	2.08	3.10	3.01	-	-
8	Sodium (mg/l)	1.55	1.05	1.29	-	-
9	Lead (mg/l)	1.06	0.96	1.02	0.01	0.05
10	Cadmium (mg/l)	0.05	0.16	0.07	0.003	0.002
11	Iron (mg/l)	1.62	0.89	0.97	0.300	0.300
12	Chromium (mg/l)	1.04	1.10	1.02	0.05	0.150

**NB** – Total Hydrocarbon Content (THC), Not Available (NA), National Environmental Standards and Regulations Enforcement Agency (NESREA), World Health Organisation (WHO).

**Table 2:** Total Viable Count

Sample	Bacteria		Fungi	
	THB (CFU/ml)	THD (CFU/ml)	THF (CFU/ml)	THD (CFU/ml)
AP1	$8.01 \times 10^5$	$1.10 \times 10^4$	$1.15 \times 10^5$	$2.10 \times 10^3$
KR1	$3.14 \times 10^5$	$0.90 \times 10^4$	$1.30 \times 10^5$	$3.09 \times 10^3$
JGI	$3.91 \times 10^5$	$1.41 \times 10^4$	$1.70 \times 10^5$	$2.60 \times 10^3$

### Identification and Antimicrobial Sensitivity

#### Profile Biosurfactant-Producing Isolates

Isolate K3A1 (Table 3) was identified by 16S rRNA sequencing. Sequence alignment and phylogenetic tree analysis showed that the 16S rRNA sequence obtained in this study was

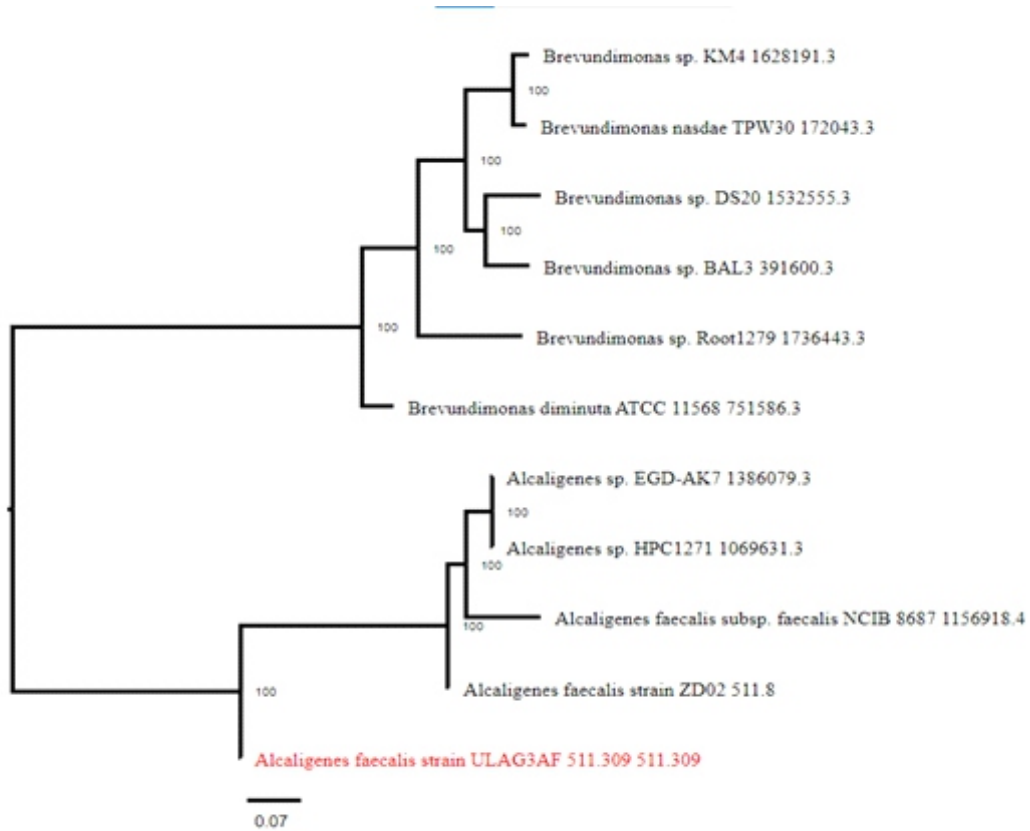
identified as *Alcaligenes faecalis* strain ULAG3AF (Figure 1). Antimicrobial sensitivity tests showed that *Alcaligenes faecalis* strain ULAG3AF was sensitive to all the antibiotics tested, except Cefotuxime, Nitrofurantoin and Ampliclox (Figure 2).

**Table 3:** Capability of isolates to produce biosurfactants based on drop collapse, oil displacement and Emulsification Index.

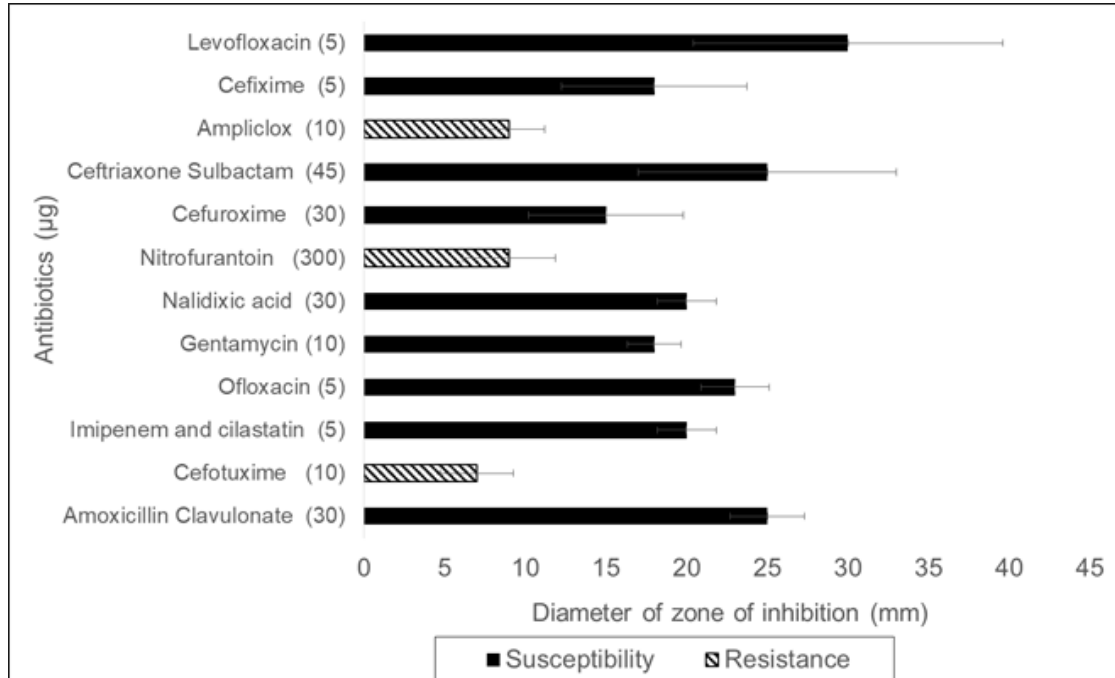
Bacterial Isolates	Drop Collapse Assay	Oil Displacement Test	Emulsification Index (%)	Yield in 1000 ml of Broth (ml)
<b>K3A1</b>	+	++	<b>68</b>	<b>2.2</b>
K3B4	+	++	63	2.2
K3C3	+	+	24	1.8
K3D1	+	++	53	2.0
K3E1	-	+	29	0.3
K3F7	+	+	42	1.8
K3G1	+	+	32	0.9
K3H1	+	++	45	0.3

**Drop Collapse Test:** (-) indicates rounded drops of oil. (+) indicates spread of the oil.

**Oil Displacement Test:** (-) indicates no spread, (+) incomplete spread ranging between 0.5 and 1.0 cm, (++) indicates significant spread of diameter ranging from 1.0 cm to 2.0 cm, (+++) indicates complete spread with diameter >2cm. The isolate selected for further studies are in bold.



**Figure 1:** Phylogenetic Tree of *Alcaligenes faecalis* strain ULAG3AF with the Unique Genome Identifier.



**Figure 2:** Antibiotics Resistance Profile of *Alcaligenes faecalis* strain ULAG3AF  
 NB: Resistant = Diameter of zone of inhibition is <15 mm; Susceptible = Diameter of zone of inhibition is ≥15mm.

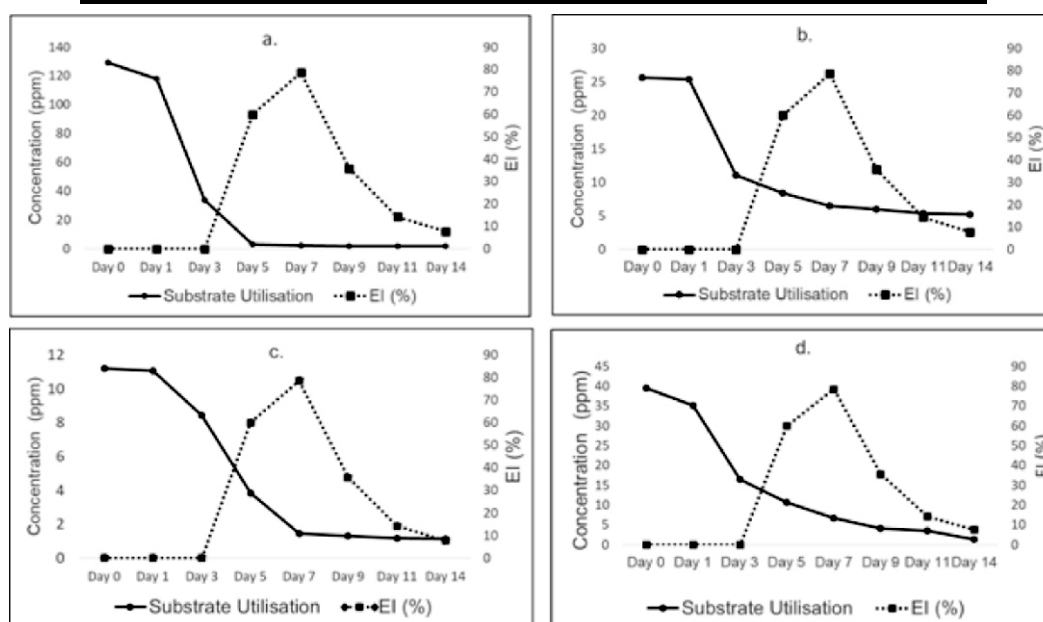
### Biodegradation Ability of the Isolate

*Alcaligenes faecalis* strain ULAG3AF utilised all the PAH during the production of biosurfactant with spent oil as a substrate, albeit at different rates (Table 4). *Alcaligenes faecalis* strain ULAG3AF

degraded between 50.34% (benzo (b) fluoranthene) 96.96% (benzo (a) anthracene) of the PAHs in 14 days. Significant utilisation of the PAHs was observed between days 1 and 7 when there was also significant emulsification activities, indicating biosurfactant production (Figure 3).

**Table 4:** The Degree of Degradation of Polyaromatic Hydrocarbons in Spent Engine Oil over 14 Days by *Alcaligenes faecalis* strain ULAG3AF.

Polyaromatic Hydrocarbon	Degree of Degradation (%)
Benzo(a)anthracene	96.96
Naphtalene	92.46
Dibenzo(a,h)anthracene	91.74
Fluorene	91.66
Indeno(1,2,3-c,d)pyrene	90.82
Acenaphtene	90.71
Pyrene	88.89
Acenaphtylene	88.41
Benzo(a)pyrene	88.12
Phenanthrene	88.05
Fluoranthracene	87.98
Anthracene	87.84
Benzo(g,h,i)perylene	82.87
2-methyl naphtalene	58.14
Chrysene	55.25
Benzo (b)fluoranthene	50.34



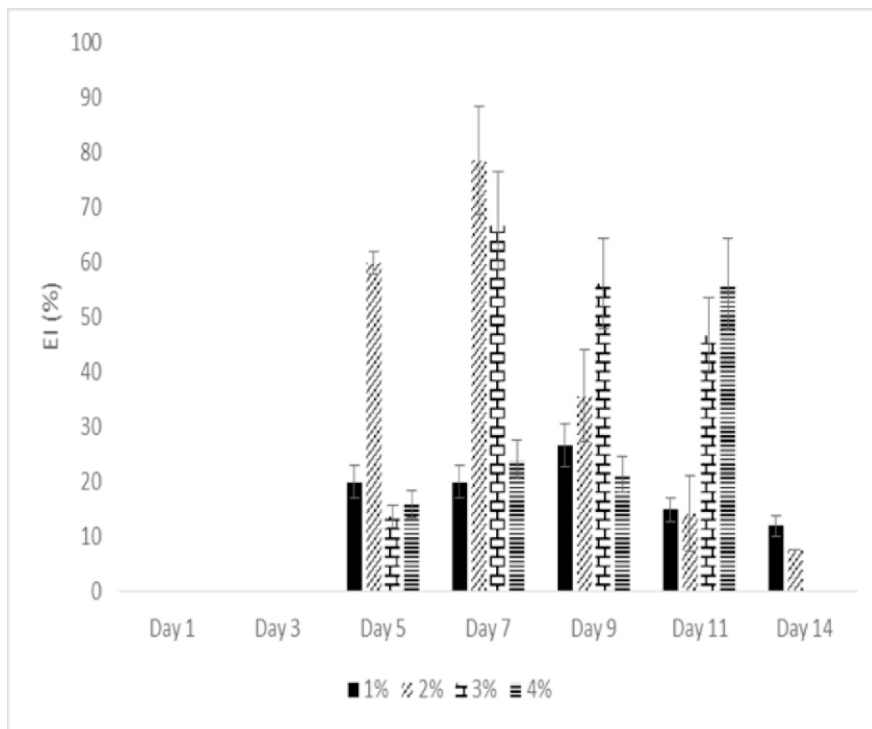
**Figure 3:** The Relationship between Emulsification Index (EI) and Utilisation of Naphtalene (a.), Dibenzo(a,h)anthracene (b.), Flourene (c.), and Benzo(a)anthracene (d.) by *Alcaligenes faecalis* strain ULAG3AF during Biosurfactant Production.

## Biosurfactant Production Process Optimisation

### Effect of Substrate and Concentration on Biosurfactant Production

Figure 4 shows the emulsification index (biosurfactant production) over 14 days with different concentrations of spent engine oil. At all concentrations, biosurfactant production began after day 3 and peaked between days 7 and 9 (Figure 4). After day 9, biosurfactant production

began to decline. Generally, biosurfactant production rate was directly proportional to the growth rate of the isolate. The highest Emulsification Index (EI) (78.9%) was observed on day 7 with 2% (v/v) spent oil. However, there was no significant difference between the EI values observed on day 7 with 2 and 3% (v/v) spent oil. With 3% and 4% v/v spent oil high EI was observed on days 9 and 11, respectively, suggesting delayed biosurfactant production with an increase in substrate concentration.



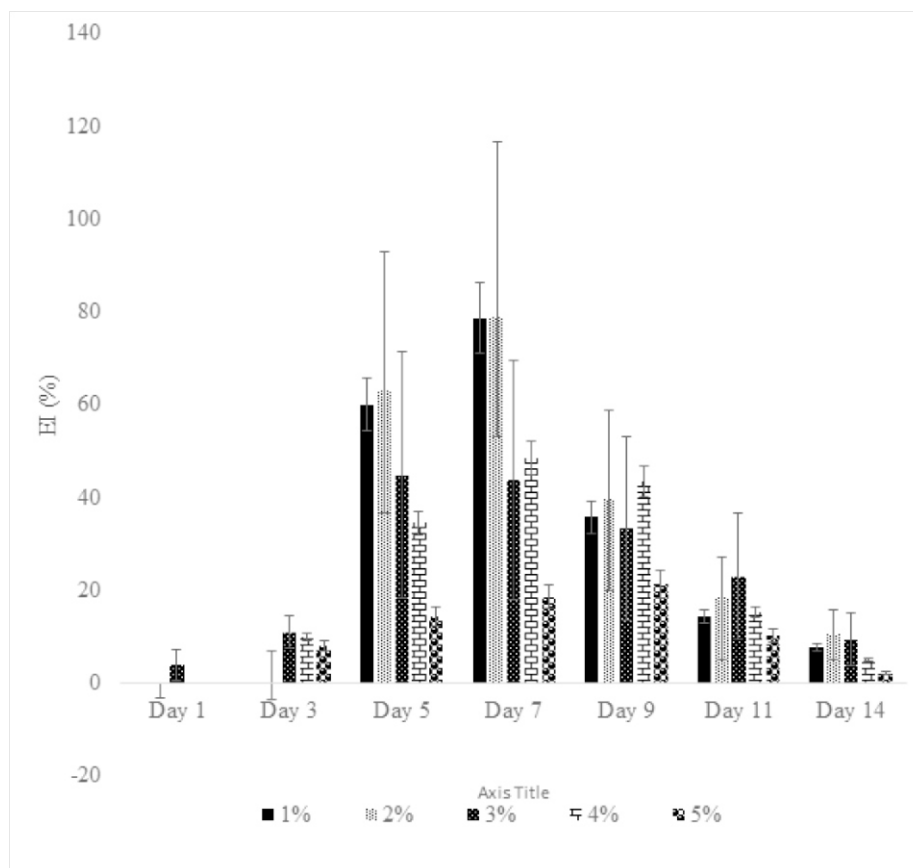
**Figure 4:** Biosurfactant Production (Measured in Emulsification Index) by *Alcaligenes faecalis* strain ULAG3AF in Spent Oil at 1%, 2%, 3% and 4% v/v. (NB: EI- Emulsification Index).

### Effect of Salinity on Biosurfactant Production

At 1 and 2% salinity (10 mg/l and 20 mg/l NaCl), bioproduction was not observed until day 5 when *Alcaligenes faecalis* strain ULAG3AF was cultivated with 2% (v/v) SEO at 30 °C (Figure 5). The onset of bioproduction appeared to occur earlier as the level of salinity increased. The late onset observed at 1 and 2% salinity notwithstanding, the highest concentrations of biosurfactant production was observed at these levels of salinity

once bioproduction commenced. The highest yield was observed on day 7 with 1% salinity producing an EI of 78.57% and 2% salinity an EI of 78.97%, suggesting that there was no difference in the ability of *Alcaligenes faecalis* strain ULAG3AF to produce biosurfactants at both salinity levels. However at higher salinity levels (3-5%), bioproduction rates are considerably lower although they span for a longer period. In addition, beyond day 7, bioproduction rates reduced at both 1 and 2% salinity levels.



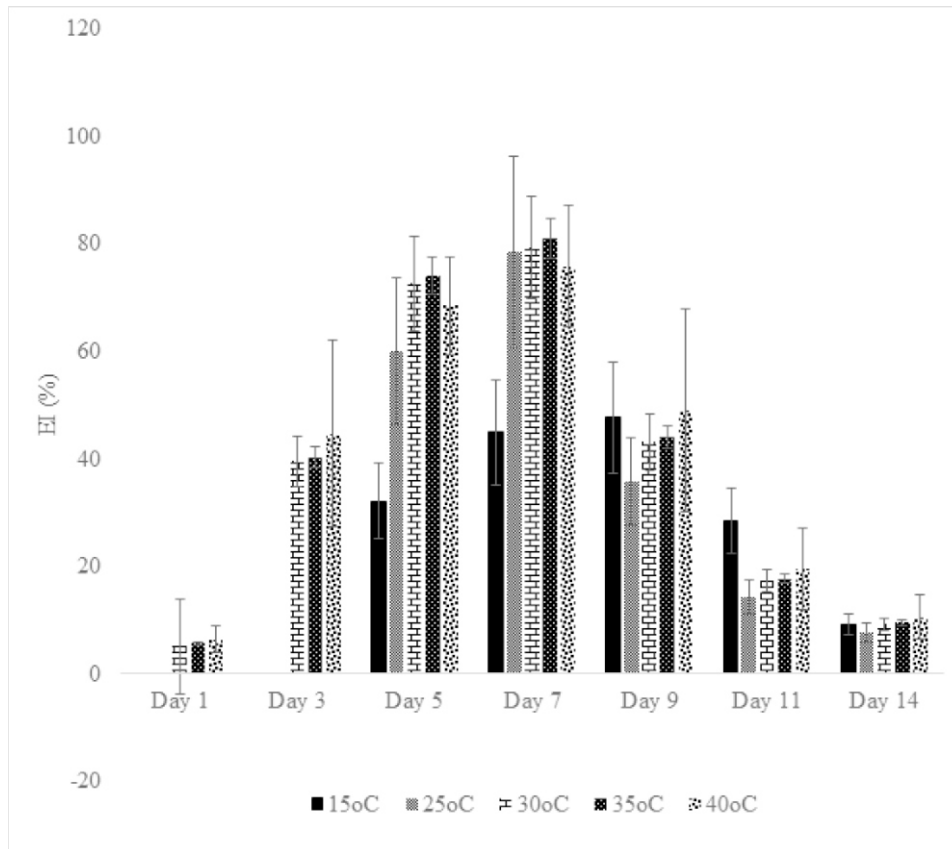


**Figure 5:** Biosurfactant Production (Measured in Emulsification Index) by *Alcaligenes faecalis* strain ULAG3AF in Spent Engine Oil at 1% (a), 2% (b), 3% (c), 4% (d), and 5% (e) Salinity.

### Effect of Temperature on Biosurfactant Production

When *Alcaligenes faecalis* strain ULAG3AF was cultivated with 2% salinity and 2% (v/v) spent oil, significant EI was observed from 25°C, and reached its peak at 35°C (80.94%) with a slight decline at 40 °C (75.85%) (Figure 6). However,

there was no significant difference between the yields obtained at the temperature ranging from 25 °C to 40 °C. Additionally, although biosurfactant production peaked on day 7 and declined significantly on day 9, there is no significant difference between the yields observed on days 5 and 7.

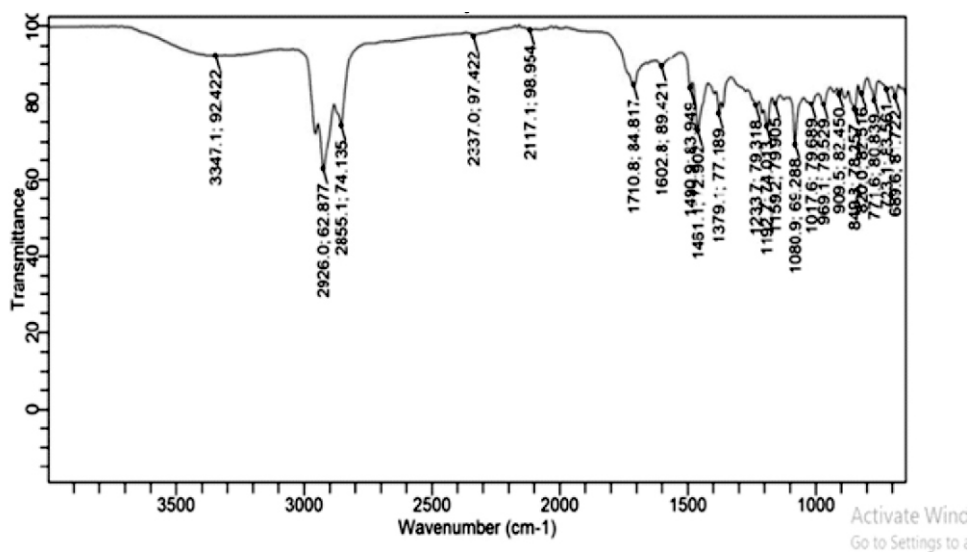


**Figure 6:** Biosurfactant Production (Measured in Emulsification Index) by *Alcaligenes faecalis* strain ULAG3AF in 2% v/v Spent Oil at 15 °C, 25 °C, 30 °C, 35 °C and 40 °C.

**Chemical Characterisation of Biosurfactants**

The Attenuated Total Reflectance Fourier Transform Infrared (FTIR) spectrum of the biosurfactant produced by *Alcaligenes faecalis* strain ULAG3AF is shown in Figure 6. The FTIR analysis of the biosurfactants produced by *Alcaligenes faecalis* strain ULAG3AF showed a total

of 22 absorbance peaks. Table 5 shows some peaks and the corresponding functional groups present in the biosurfactants based on evidence from literature (Das *et al.*, 2016). The biosurfactant produced by *Alcaligenes faecalis* strain ULAG3AF had OH, CO and CH bonds.



**Figure 7:** FTIR analysis of Biosurfactant Production by *Alcaligenes faecalis* strain ULAG3AF.

**Table 5:** Peaks and corresponding functional groups present in the biosurfactants produced by *Alcaligenes faecalis* strain ULAG3AF.

Organism	Wavelength	Bonds
	3347.1	OH bond or N-H bond
<i>Alcaligenes faecalis</i> strain ULAG3AF	2926, 2855	C-H bond
	2337	O=C=O Bond
	1080	C=O Bond

**Antimicrobial Activity**

The biosurfactant produced by *Alcaligenes faecalis* strain ULAG3AF showed antibacterial but not antifungal properties (Table 6). *Klebsiella pneumoniae* and *Pseudomonas aeruginosa* (Gram-negative bacteria) showed the least susceptibility to the biosurfactant, while *Bacillus subtilis* and

*Staphylococcus aureus* (Gram-positive bacteria) showed the higher susceptibility to the biosurfactant. However, the zones of inhibition with the biosurfactant were lower compared to the zones of inhibition observed with the chemical antibiotic, Streptomycin, which was used as the positive control.

**Table 6:** Antimicrobial Activity of Biosurfactant.

Organism Used	Diameter of Zone of Inhibition (mm)± SD		
	Biosurfactant	Positive Control	Negative Control
<i>Pseudomonas aeruginosa</i>	2.1 ± 1.05	24 ± 1.32	0
<i>Bacillus subtilis</i>	6.7 ± .41	19 ± 3.16	0
<i>Staphylococcus aureus</i>	4.8 ± 0.76	36 ± 3.17	0
<i>Klebsiella pneumoniae</i>	2.1 ± 0.20	31 ± 0.62	0
<i>Aspergillus niger</i>	0	15.61 ± 0.16	0

NB: SD- Standard Deviation, Positive controls: Streptomycin (10 µg/ disc) for bacteria, Fluconazole (1.0 mg/disc) for the fungus.

**DISCUSSION**

The presence of heavy metal and hydrocarbon pollutants at the sampling points is unsurprising given that the Lagos Lagoon is exposed to diverse anthropogenic activities, including water transportation, industrialization, and waste disposal (Alo *et al.*, 2014). In an earlier study, Sogbanmu *et al.* (2016) reported that the Apapa site of the Lagos Lagoon was the most contaminated site on the lagoon, which spans approximately 6000 km sq, owing to the presence of numerous petroleum tank farms located along the port's terminals.

In this study, the isolated hydrocarbon-utilizing and biosurfactant-producing bacterium was identified as *Alcaligenes faecalis* strain ULAG3AF. The isolate showed high emulsification indices and stronger oil displacement activity but weak drop collapse activity. Both oil displacement and

drop collapse activities are typically used for the qualitative identification of biosurfactant producers, whereas the emulsification index (EI) allows the quantification of the biosurfactant production potentials of the producing organism (Irorere *et al.*, 2017). The emulsification index shows a linear relationship between emulsification activity and the amount of biosurfactant produced; hence, it is ideal for quantifying the amount of biosurfactant produced (Mounira & Abdelhadi, 2015). In this study, multiple screening methods were used to identify biosurfactant producers, as this would help eliminate the drawbacks of individual screening methods, ensuring that true biosurfactant producers are identified (Walter *et al.*, 2010).

*Alcaligenes faecalis* has previously been isolated from different hydrocarbon-contaminated sites, including sediment and water from the Lagos

Lagoon, soil from mechanical workshops, and wastewater amongst others (Igwo-Ezikip *et al.*, 2009; Muftau *et al.*, 2021; Salam *et al.*, 2018). These studies demonstrate the hydrogen-utilizing abilities of the bacterium. However, despite its application in bioremediation, the isolate is a known opportunistic pathogen, and its presence in a body of water, such as the Lagos Lagoon, can pose an environmental threat (Tang *et al.*, 2020). Infections caused by opportunistic pathogens, which are often transmitted by water, can cause significant public health concerns, particularly if they are resistant to many antibiotics (Stec *et al.*, 2022). Furthermore, the use of biosurfactants from pathogenic bacteria is ethically unacceptable because of the possibility of the co-production of harmful toxins or pigments (Uzoigwe *et al.*, 2015). Although the *Alcaligenes faecalis* strain ULAG3AF isolated herein was sensitive to all the antibiotics tested, except Cefotuxime, Nitrofurantoin and Ampliclox, its potential pathogenicity may limit its applicability for bioremediation.

This isolate demonstrated the ability to utilize different polyaromatic hydrocarbons as substrates for the production of biosurfactants. The ability of the bacterium to utilize hydrocarbons may be attributed to the presence of hydrocarbon degradation enzymes in the organism (Adetitun *et al.*, 2020). Undugoda, Kannangara, and Sirisena (2016) reported the presence of the *phnG* gene, which is involved in phenanthrene degradation, and the *nabR* and *nabU* genes both involved in the naphthalene degradation pathways in the genome of *Alcaligenes faecalis*. The hydrocarbon-utilizing capability of the isolate makes it a candidate for the bioremediation of PAH-contaminated sites. As observed in this study, spent engine oil, which is a mixture of different PAHs, sustained both the growth and biosurfactant production potential of the isolate for 14 days, although the biosurfactant production rate began to decrease after day 7. The extended sustenance of the bacteria by the substrate (spent engine oil) was possibly due to the organisms assimilating the different PAHs present in the mixture at different rates. According to Brown and Jones (2024), the carbon use efficiency of substrates increases when there is substrate diversity, thereby, resulting in prolonged microbial carbon uptake and microbial carbon storage potential.

The results obtained herein show that at 1 to 5% salinity levels, *Alcaligenes faecalis* could produce biosurfactants, although the optimum salinity levels were between 1% and 3%. The estuarine environment, from whence the bacterium was isolated, has variable levels of salinity and undergoes constant reflux; hence, the ability of the isolate to withstand a range of salinity levels is unsurprising (Hillmann *et al.*, 2019). Additionally, this strain of *Alcaligenes faecalis* in this study grew and produced biosurfactant optimally at temperature range from 25 °C to 40 °C, suggesting that the organism can withstand fluctuating environmental temperature. This result is similar to findings reported by Arabo *et al.* (2021) and Najafi-Marghmaleki *et al.* (2018), who reported optimal production of biosurfactant by *A. faecalis* at 35 °C and 28 °C, respectively. Bharali *et al.* (2011) also reported that the *A. faecalis* strain in their study grew and produced biosurfactants optimally at 42 °C. This temperature is higher than the temperature reported in this study, suggesting that the organism is possibly mesophilic, considering that mesophiles can grow between 20 °C and 45 °C (Zuberer and Zibilske, 2021). However, Salehizadeh & Mohammadizad (2009) reported that at 50 °C, the strain of *Alcaligenes faecalis* grew and produced biosurfactants optimally. This suggests that the bacterium may be a facultative thermophile because of its ability to withstand high temperatures (Zuberer and Zibilske, 2021). The ability of *A. faecalis* to produce biosurfactants at such high temperatures is an attractive feature for biotechnological and bioremediation studies (Antoniou *et al.*, 2015).

The biosurfactant produced by *Alcaligenes faecalis* strain ULAG3AF had OH, CO, and CH bonds, suggesting that the biosurfactants are glycolipids (Hamed *et al.*, 2021). Bharali *et al.* (2011) also reported the production of glycolipids by *Alcaligenes faecalis*. Furthermore, glycolipids such as rhamnolipids have been reported to be produced by the uptake of hydrophobic substances, including pure samples of pyrene, chrysene, naphthalene, and fluoranthene (Patowary *et al.*, 2022) and hydrocarbon mixtures such as diesel and crude oil (Malavenda *et al.*, 2015; Maneerat *et al.*, 2019). According to Li *et al.* (2015), biosurfactants increase the hydrophobicity and

solubilisation of the cell surface, thereby increasing the bioavailability of the hydrophobic substrate to the microorganism, which then metabolizes or degrades it.

The biosurfactant produced by the strain of *Alcaligenes faecalis* in this study exhibited antimicrobial activities. However, it appeared to be more effective against gram-positive isolates than gram-negative isolates. These results contradict those reported by Bharali *et al.* (2011), who reported that the biosurfactant produced by their strain of *Alcaligenes faecalis* had a higher antimicrobial effect on gram-negative bacteria than gram-positive bacteria. The antimicrobial activity of biosurfactants is important for the survival of the biosurfactant producer to thrive in a competitive environment (Banat *et al.*, 2014). Different authors have proposed the mode of action of biosurfactants as antimicrobial agents. For instance, Gomaa (2012) reported that the addition of lipopeptide biosurfactant to *Staphylococcus aureus* cells resulted in the reduction of the protein and lipid content of the cells, possibly due to the biosurfactant preventing protein synthesis by binding to the 30S ribosome and preventing the aminoacyl-tRNA from binding to the 30S ribosome. Additionally, Cortés-Sánchez *et al.* (2013) biosurfactants can alter the structure of proteins in the cell wall, thereby increasing their permeability and resulting in cell lysis.

The findings from this study also showed that the inhibition zones produced by the biosurfactant was lower than those produced by chemical antimicrobials (positive controls). These findings suggest that biosurfactant may be required at higher concentration to be more effective. However, this study did not investigate the Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC) of the biosurfactant. While MIC is the minimum concentration of a substance that can elicit a bacteriostatic effect on a bacterium, MBC is the minimum concentration that is lethal to a particular bacterium (Bindayna and Al-Salman, 2023). Determining the MIC and MBC of the

biosurfactant would provide a quantitative assessment of the effectiveness of the biosurfactant as an antimicrobial agent (de Freitas *et al.*, 2019). Nonetheless, this study highlights the antimicrobial potential of the biosurfactant produced by *Alcaligenes faecalis* strain ULAG3AF.

## CONCLUSION

This study explored the hydrocarbon-degrading and biosurfactant-producing capabilities of *Alcaligenes faecalis* strain ULAG3AF, a mesophile which could produce biosurfactant optimally at 40°C. Using spent engine oil, this study demonstrated the ability of the bacterium to utilise different polyaromatic hydrocarbon components of spent engine oil while producing a biosurfactant with an emulsifying index of 80.9%, when cultivated in a medium with 2% v/v salinity (20 mg/l NaCl) and substrate at 40 °C in 7 days. The bacterium also demonstrated the ability to degrade a wide range of PAHs including pyrene, naphthalene and chrysene. Although the bacterium is a known opportunistic pathogen, a quality that could discourage its use for environmental bioremediation owing to the public health threat it poses, it showed susceptibility to a wide range of antibiotics. The biosurfactant it produces was also identified as a glycolipid with antimicrobial activities against selected bacteria and fungi.

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## CONFLICT OF INTERESTS

The authors declare they have no conflict of interest

## AUTHORS' CONTRIBUTIONS

A.T.O designed and conducted the research, analysed and interpreted the data and wrote the manuscript. A.E.O. designed and supervised the research and edited the manuscript. A.A.A. proofread and edited the manuscript.

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