INFLUENCE OF NIOBIUM AS SOLUTE ELEMENT ON HIGH TEMPERATURE MECHANICAL BEHAVIOUR OF LOW CARBON STEEL

G.J. ADETUNJI, J. A. AJAO AND A. B. ALABI

(Received 14 February 2006; Revision Accepted 17 July 2006)

ABSTRACT

The morphological changes during the tempering of low carbon Fe-Nb steel are described and related to the corresponding changes in the high temperature mechanical property. Compositional variation of microalloy caused resultant modification of the mechanical properties of the low carbon steel under investigation. The importance of solid solution, dispersion hardening and particularly grain refinement within the tempering temperature range (650°C – 750°C) is highlighted. The importance of carbide forming niobium in modifying the tempering behaviour is discussed. At 750°C tempering and water quench, enhanced matrix hardening occurs irrespective of niobium compositional différence.

KEYWORDS: Low carbon steel, tempering, mechanical properties, precipitation, microstructure.

1. INTRODUCTION

Tempering of steel provides a means of controlling the mechanical properties of steels and hence their use in engineering applications. Its importance can be highlighted when changes in mechanical properties are related to microstructural changes. Nutting (1983) provided adequate review on the tempering of carbon and alloy steels.

In pure iron, the $\delta \rightarrow \alpha$ reaction is extremely rapid but as is well known, the addition of alloving elements both in interstitial and substitutional solid solution retards the rate of transformation. Carbon has a large effect because of its much larger solubility in austenite than in ferrite. Hence during the $\delta \rightarrow \alpha$ transformation, carbon is precipitated as carbide and the rate controlling process in Fe-C alloys will be the diffusion of carbon in austenite (Honeycombe, 1981). Carbon acts principally by solid solution strengthening in non-stabilized steels but mainly by precipitation strengthening when Nb, Ti or V is present (Pickering, 1958). The formation of MX precipitates in austenitic stainless steels occurs when strong carbides/nitrides formers (Ti, Nb, V, Ta) are added to the alloy (Le Bon and de Saint Martin, 1975). MX precipitates usually form on dislocations within the matrix, on stacking-faults, on twin and grain boundaries. Stabilizers such as Nb or Ti have long been known to reduce the stability of carbon in austenite; (Sourmail, 2001). In fact, NbC precipitation in low carbon steels provides an effective way to increase the mechanical properties. It should be noted that detailed quantitative characterization of this phase is still scanty. This has informed the present study's investigation of the influence of compositional variation of Nb on the mechanical properties of high purity Fe-Nb alloy.

2. EXPERIMENTAL TECHNIQUES

2.1 Materials

The samples of high purity Fe-Nb alloys were provided by AERE, Harwell, United Kingdom in plate formeand cold rolled state with Nb compositions of 0.2 and 0.5wt%.

2.2 Heat Treatment

The samples were solution treated at 1050°C for 30minutes and subsequently quenched in air, oil and water. They were

then tempered at temperature range $600 - 750^{\circ}$ C in steps of 50° C for one hour.

2.3 Hardness Measurement

Monsanto tensiometer with Brinell Indentor attachment was employed in the determination of hardness. The stainless steel indentor has a radius of 5mm. A force of 0.75KN was applied för 30seconds and then removed. This operation was repeated several times in order to obtain statistically balanced data for all the samples. The indentation diameter was determined with the aid of an optical microscope.

2.4 Impact Test

The Charpy V-notch testing machine was employed in carrying out this test. Each of the samples was notched by the machine, making a V-notch of 45° at the center of the samples. The V-notched samples were mounted on the impact machine having its pointer touching the notch. The hammer arm was then released from a preset height of about 35cm to fracture the sample. The calibrated gauge on the machine indicates the energy required for fracture.

2.5 Characterization of the alloys

The microstructure of the alloys was examined using optical microscopy and scanning electron microscopy (SEM). The samples for SEM observations were slightly etched with an etchant consisting of 5g FeCl₃ + 10ml HCl dissolved in 50ml $\rm H_2O$.

3. EXPERIMENTAL RESULTS AND DISCUSSION

(Mechanical Behaviour of the Alloys)

Figures 1 and 2 show the plot of the hardness values against the tempering temperature for 0.2Wt%Nb and 0.5Wt%Nb alloys respectively quenched at various media. Generally, the hardness decreases as the tempering temperature decreases - with the hardness values higher in alloy with 0.5Wt%Nb. The fracture toughness is also found to increase as Nb content increases and also decreases as the tempering temperature increases (Figure 3).

- G.J. Adetunji, Materials and Electronics Division, Centre for Energy Research and Development, Obafemi Awolowo University, Ile-Ife, Osun State. Nigeria.
- J. A. Ajao, Materials and Electronics Division, Centre for Energy Research and Development, Obafemi Awolowo University
 Ile-Ife, Osun State. Nigeria.
- A. B. Alabi, Department of Physics, University of Ilorin, Kwara State, Nigeria.

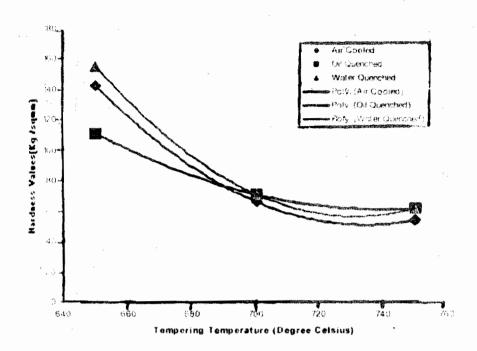


Figure 1: Hardness values against Tempering Temperature for Alloy A (0.2wt%Nb)

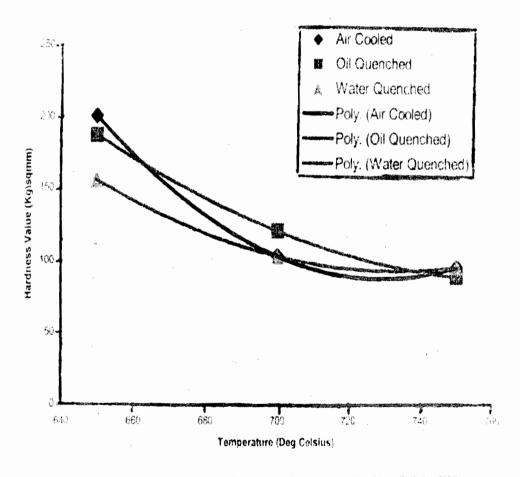


Figure 2: Hardness values against Tempering Temperature for Alloy B(0.5wt%Nb)

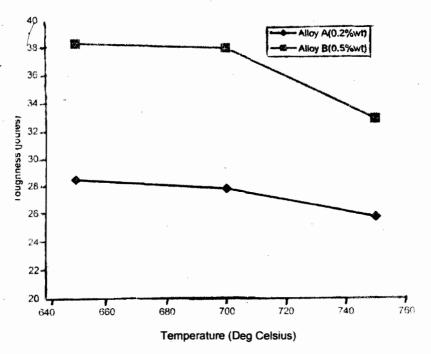


Figure 3: Toughness Against Tempering Temperature

As shown in the SEM of Plate 1, whitish precipitates of NbC could be seen in water quenched samples at 750°C. As reported by Sourmail (2001), dislocations are preferential nucleation sites for this phase. The NbC has a NaCl fcc (face

centred cubic) structure with lattice parameter of 4.47Å (Sourmail, 2001). It should also be noted that plates of $M_{23}C_6$ could be seen at grain boundaries and around NbC precipitates (Plate 2).



Plate 1: SEM showing the whitish precipitation of NbC in water quenched sampled at 750°C.



Plate 2: SEM showing plates of M23C6 surrounded by NbC precipitates

This could be explained by the possible nucleation of M23C6 on the Shockley bounding dislocation and growth in the stacking fault presumably helped by an enrichment of the stacking fault in chromium. As reported by Lewis and Hattersley (1965), the precipitation of M23C6 has been the focus of many investigations, motivated by its importance in terms of corrosion resistance. The presence of M23C6 on grain boundaries is often associated with intergranular corrosion (Ramaswamy and West, 1970; Beckitt and Clarck, 1967). Furthermore, a G-phase forms predominantly on grain boundaries in Nb-stabilized steels depending on the silicon content. It has an FCC structure with a lattice parameter of 11.2Å corresponding to a content of 116 atoms per unit (Powell et al, 1984; Kikuchi et al, 1986; Andren et al, 1980). This phase could not be easily observed in our present alloy. Carbides generally provide barriers to dislocation motion in order to enhance hardening. The Ashby - Orowan relationship can be used to assess the influence of carbides on hardness of martensitic steels (Irvine, 1970; Little and Henderson, 1958; Little et al, 1977):

$$\sigma_f = \sigma_o + \left[\frac{k}{s}\right] \log \left[\frac{d}{b}\right]$$
Where

Fracture stress σ_{o} Yield stress Interparticle spacing

Particle diameter

Burger's rector

constant

 σ_f

From the above relation, the larger the grains and the smaller the interparticle spacing, the higher the fracture stress. Though carbides provide barriers to dislocation motion to encourage hardening it should be noted that when niobium atoms are added, the niobium carbide particles' hinderance to dislocation cannot arrest the decrease in hardness due to the loss of the interstitial solute hence softening becomes predominant. It is known that the mechanical properties of tempered martensite structures depend upon the carbon content of the steel, the amount of retained austenite and the nature of the alloying elements. The presence of large amounts of retained austenite decreases the hardness of tempered martensite. martensitic lath structure could easily be observed in samples. air cooled at 750°C (Plate 3).

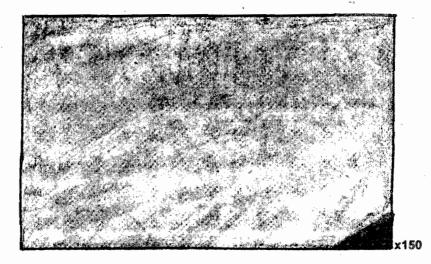
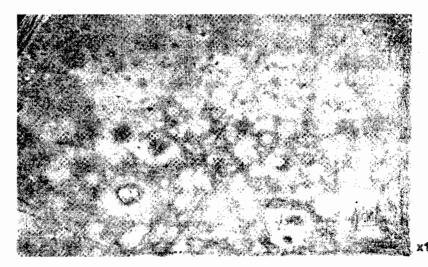


Plate 3: SEM showing the martensitic lath structure in samples quenched in air at 750°c



SEM showing the disc shape primary niobium carbides formed during solution treatment

However, the lower the carbon content, the greater is the ductility and toughness. But niobium which is a carbide former reduces the solubility of carbon in the austenite, hence reducing the carbon content of the matrix. It can then be suggested that softening of the matrix can be achieved easily when niobium is one of the alloying elements in low carbon steels. Its addition increases the tempering resistance and the refinement of the prior austenite grain size. Primary niobium carbides formed during solution treatments appear as disc shape in SEM (Plate 4). However, the secondary niobium carbides formed after tempering due to recrystallisation effect are located on the lath, and austenite lath packet boundaries junctions.

4. CONCLUSION

- The mechanical behaviour of low carbon Fe-Nb steels has been examined.
- As niobium contents increased, the fracture toughness was found to increase.
- However, the hardness values decreased as the tempering temperature increased. This is probably due to the inability of the niobium carbide particle to sufficiently arrest the dislocation motion.
- . The presence of the metastable phase $M_{23}C_6$ was also reported.
- It has been established that NbC precipitation in low carbon steels provides an effective way to enhance the mechanical properties of these alloys

REFERENCES

- Andren, H. O., Henjered, A. and Norden, H., 1980. Composition of MC Precipitates in a Titanium Stabilised Stainless Steel. J.Mater. Sci., 15:2365-2368.
- Beckitt, F R and Clarck B R 1967 The Shape and Mechanism of Formation of M₂₃C₆ Carbide in Austenite Acta Metall 15 113 129

- Honeycombe, R. W. K. 1981. "Steels, Microstructure and Properties". Edward Arnold, London Irvine, K. J., 1970. J. Iron and Steel Inst. 717.
- Kikuchi, M., Sakabibara, M., Otoguro, Y., Mimura, M., Takahashi, T., and Fujita, T., 1986. An Austenitic Heat Resisting Steel Tube developed for Advanced Fossil Steam Plant. In International Conference on Creep. Tokyo, April 14-18, 1986, pages 215 220.
- LeBon, A. B. and de Saint Martin, L. N., 1975, "Microalloying" 75 Greenwich, Conn. Climax Molybdenum Co.
- Lewis, M. H. and Hatterşley, B., 1965. Precipitation of M₂₃C₆ in Austenitic Steels. Acta Metalf., 13, 1159-1168.
- Little, E. A., Harris, D. R., Pickering, F. B. and Keown, S. R., 1977. Effects of heat treatment on structure and properties of 12wt% Cr. Steels. Metals Tech. 4: 1977, 205-217
- Little, J. H and Henderson, W. L. M., 1958. "Effects of second phase particles on the mechanical properties of steels" iron and Steel Inst. London, 23
- Nutting, J., 1983. "The tempering of carbon and alloy steels" Proc. Of Topical conf. on Ferritic Alloys" Snowbird, Utah, June 19-23, 1983.
- Pickering, F. B., 1958 "Precipitation processes in steels" The iron and Steel Inst. London, 23.
- Powell, D. J.: Pilkington R., and Miller D. A.: 1984. Influence of Thermal Ageing on Creep Properties of 20/25. Nbstabilised Steel. In Proc. Conf. Stainless Steels. 84. Gothenburg, Sept. 1984, pages 382-390.
- Ramaswamy, V and West D. R. F. 1970. NbC Precipitation in a 20%Cr-25%Ni-1%Nb Austenitic Steel. J. I. S. I., pages 391 - 394.
- Sourmail, T., 2001. "Precipitates in Creep Resistant Austenitic Stainless Steels" Mat. Sci. Tech. 17-1. 1-14