

Structural Characterization of Gas Tungsten Arc Welded Stainless Steel at High Heat Input

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ORIGINAL RESEARCH

Abstract- Performance integrity of weldment differs considerably from the base metal due to microstructural changes, resulting from welding heat input effect. In this study, effects of high gas tungsten arc welding heat inputs on microstructure of 304L austenitic stainless steel (ASS) weldments were investigated. The weldments were produced based on ASTM A778/778M at GTAW speed, current and voltage of 1.7 mm/s, 200 A and 40 V respectively. After which, they were characterised, using X-ray diffraction (XRD) and optical microscopy. The experimental samples, both control and weldments were prepared for the XRD and microstructural analysis following ASTM E 975-03 and ASTM E3 – 11 respectively. Results showed that control sample and weldments were characterized by varying amounts of major and minor compounds. Crystal structure of the control sample is comprised of a mixture of face centred cubic (FCC) and body centred cubic (BCC) with major peak (111). While those of the 1.7 mm/s, 200 A and 40 V produced weldments were comprised majorly of FCC with major peak (511), BCC with major peak (211) and BCC with major peak (110) respectively. Microstructure of the control sample is homogenous, comprising of equiaxed – grained austenite (γ) matrix with precipitates of varying sizes (dark spots), which are distributed non-uniformly within the γ matrix and small amounts of δ -ferrite (dark lines) along the γ matrix grain boundaries. While microstructures of the weldments are heterogeneous, comprising of austenite (γ) matrix and ferrite (α) grains, which are dispersed within the γ matrix. Generally, grains of the fusion zones (FZs) microstructures are fine relative the heat affected zones (HAZs) microstructures. The FZs and HAZs microstructures are characterised by precipitates, δ -ferrite, inclusions and dendrites of varied number and size.

Keywords: austenite (γ) matrix, equiaxed – grained, ferrite (α) grains, fusion zone (FZ), heat affected zone (HAZ), microstructures, weldments.

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Consequently, in this study, high heat input inducing parameters were considered, because influence of heat input on either mechanical or corrosion performance of the weld joint is more severe at high heat input.

1 INTRODUCTION

304L austenitic stainless steel (ASS) is known for good combination of excellent corrosion resistance, good ductility and good weldability (Amer *et al.*, 2015; Hussain, 2010; Kožuh *et al.*, 2009). It has been successfully welded with shielded metal arc welding (SAW), submerged arc welding (SMAW), plasma arc welding (PAW) and gas tungsten arc welding (GTAW) techniques (Oyetunji *et al.*, 2013). Hence, its wide use as fabrication material for various industrial facilities, including oil and petrochemical fields, chemical plants, biomedical implants and food industries. However, during welding, microstructure, surface texture and composition of the weld joint is influence by the welding heat inputs and, in consequence, mechanical and corrosion properties of the weld joint differ considerable from the base plate (Abioye, 2017; Apurv and Vijaykumar, 2014; Devakumar and Jabaraj, 2014; Ramesh and Chauhan, 2014; Kožuh *et al.*, 2009). Therefore, adequate understanding of changes, occurring at the weld joint is necessary for making useful decision on the choice of welding parameters for production of the weldment either on industrial or domestic scale.

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Section D- MATERIALS AND METARLLUGY/CHEMICAL SCIENCES & RELATED SCIENCES

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2 METHODOLOGY

2.1 CHEMICAL COMPOSITION

Chemical compositions of the as-received 304L ASS plate and filler rod employed for the production of the weldments were obtained by optical emission spectrometry (AR 4 30 metal analyser), and the results are presented in Tables 1 and 2 respectively

Table 1. Chemical composition of the experimental as-received 304L ASS plate

Element	Wt. %	Element	Wt. %	Element	Wt. %
C	0.026	Cr	18.325	Cl	0.002
Si	0.511	Ni	8.469	Mo	0.069
Mn	1.311	Cu	0.135	V	0.083
S	<0.001	Nb	0.009	Ti	0.036
P	0.013	Al	0.018	Fe	75.916

Table 2. Chemical composition of the filler metal

Element	Wt. %
C	<0.03
Mn	1.650
Si	0.65
P	0.03
S	0.03
Cr	19.5-22.0
Ni	9.0-11.0

2.2 SAMPLE PREPARATION

Samples of dimensions 120 mm × 20 mm × 8 mm were cut from the as-received 304L ASS plate, using simple handsaw, and they were cleaned with acetone to remove lubricant and surface contamination. The samples were machined to butt – joint to make a single – V groove of 60° with root gap of 2.5 mm, this configuration was maintained to ensure good root penetration (Janunkar *et al.*, 2017) The welding parameters were fixed at 1.7 mm/s, 200A and 40V, corresponding to high heat input. Multipass welding was conducted on the samples, using Clark TIG welding machine and 2.5 mm diameter filler electrode. The values of current and voltage were read directly from the GTAW machine, while values of speed were obtained by calculation using the expression in Eqn. 1 [24]. The weldments were visually inspected for any weld defects and/or geometrical non-conformity. Table 3 summarises the procedure for the weldment production and Table 4 shows the quantities of heat input at the varied GTAW parameters, and Fig. 1 and Fig. 2 are samples of the butt-joint with single – V configuration and the produced weldment.

$$S = \frac{L}{T} \tag{1}$$

Where S is welding speed (mm/s), L is length (mm) and T is time (s). The time taken to produce each weldment was determined using a stop watch.

Table 3. Summary of the procedure for weldments production

S/No	Welding parameters		
1.	Speed (mm/s)	Current (A)	Voltage (V)
	1.7	160	30
2.	Current (A)	Speed (mm/s)	Voltage (V)
	200	4.6	30
3.	Voltage (V)	Current (A)	Speed (mm/s)
	40	160	30

Table 4. Quantity of heat input at the varied GTAW parameters

S/No.	Welding parameters			Heat input (J/mm)
1.	Speed (mm/s)	Current (A)	Voltage (V)	1694.5
	1.7	160	30	
2.	Current (A)	Speed (mm/s)	Voltage (V)	782.61
	200	4.6	30	
3.	Voltage (V)	Current (A)	Speed (mm/s)	834.78
	40	160	4.6	

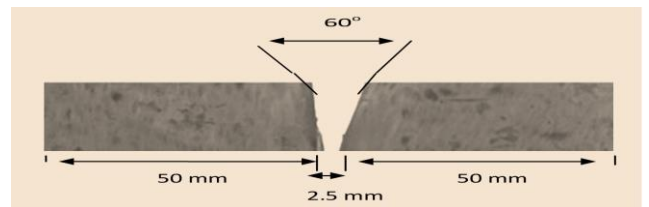


Fig. 1. Butt joint sample with single-V configuration

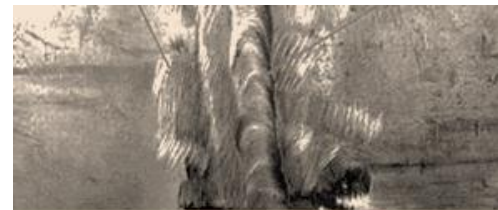


Fig. 2. Sample of the GTAW 304L ASS weldment

2.3 X-RAY DIFFRACTION (XRD) MEASUREMENT

The XRD samples were prepared in accordance with ASTM E 975-03. Phase constituents of the samples were obtained at room temperature (25°C), using Philips X’Pert diffractometer under Cu-Kα radiation with measuring accuracy of 5%. Data were collected over the 2θ range of 10o to 80o with a step size of 0.05° and an exposition time of 2 s per step, X’Pert High Score plus software was used to analyse the data.

2.4 OPTICAL MICROSCOPY

The metallographic samples were prepared based ASTM E3-11. Conventional metallographic grinding and polishing techniques were used to achieved the desired surface finish. The samples were etched in solution of 50 ml HCl + 50 ml HNO₃ + 50 ml water for two minutes. FZ and HAZ microstructures were viewed and captured, using optical metallurgical microscope (OMM) (Olympus GX51) with camera attached.

3 RESULTS AND DISCUSSION

3.1 X-RAY DIFFRACTOMETER (XRD) ANALYSIS

Fig. 3 and Fig. 4 are XRD diffractogram and stick pattern of the control sample respectively, and Table 5 and Table 6 are the peak list and identified pattern list respectively. From the results, The major peak, corresponding to X, Y and Z planes is (111), and minor peaks, corresponding to X, Y and Z planes are (200), (220), (311) and (222). Compounds of the major peak are Co_3Fe_7 , $\text{Fe}_{0.64}$ and $\text{Ni}_{0.36}$, those of the minor peaks are $\text{Cr}_{1.36}\text{Fe}_{0.52}$, Cr and Cr Si (Table 6). The sample is comprised of a mixture of face centred cubic (FCC), corresponding to austenite ($\gamma\text{-Fe}$) phase, and body centred cubic (BCC), corresponds to ferrite ($\alpha\text{-Fe}$) phase. The control is therefore not fully austenitic, but predominantly austenitic (Kožuh *et al.*, 2009; Honeycombe and Bhadeshia, 1995). And its chromium content of 18.325% (refer to Table 1) was indicative of its good to excellent corrosion resistance (Kožuh *et al.*, 2013). The x-ray diffractogram and stick pattern of weldment produced at the speed of 1.7 mm/s are shown Figs. 5 and 6 respectively, and the identified peak list and pattern list are depicted in Table 7 and Table 8 respectively.

From the results, the major peak, corresponding to X, Y and Z planes is (511), and the other identified minor peaks, corresponding to X, Y and Z planes are (111), (200), (220), (311), (222), (400), (331), (420), (422), (440), (531) and (600). Compounds of the major peak are C, Ni_3ZnCo_7 , $\text{Cr}_{1.36}\text{Fe}_{0.52}$, and those of the minor peak are Fe Ni, Mn Ni $_{16}\text{P}_7$, Si O₂, Mn₃Co₇, Fe₇S₈, Ni O and Cr.

The crystal structure of the weldment is FCC, corresponding to ($\gamma\text{-Fe}$) phase. Hence, it is fully austenitic (Kožuh *et al.*, 2009). Fig. 7 and Fig. 8 are the X-ray diffractogram and stick pattern of weldment produced at the current of 200A respectively, and the corresponding identified peaks list and pattern list are shown in Table 9 and Table 10 respectively, the major peak, corresponding to X, Y and Z plane is (211). Other identified minor peaks lists, corresponding to X,Y and Z planes are (200), (211), (220), (310) and (222). Compounds of the major peaks are Fe-Cr and Cr, and those of the minor peaks are Ni_3ZnCo_7 , Ni O, Fe Si, C, Fe₃O₄, Ni and Mn Si. The crystal structure of the weldment is BCC, corresponding to $\alpha\text{-Fe}$ phase. Hence, the weldment is comprised majorly of ferrite (Kožuh *et al.*, 2009). X-ray diffractogram and stick pattern of weldment produced at the voltage of 40V are depicted in Fig. 9 and Fig. 10 respectively, and the identified peak list and pattern list are shown in Table 11 and Table 12 respectively. The major peak, corresponding to X,Y and Z is (110). Other minor peaks, corresponding to X,Y and Z planes are (200), (220), (222), (321) and (332). Compounds of the major peaks are Fe-Cr and Cr, and those of the minor peaks are $\text{C}_{0.14}\text{Fe}_{1.86}$, $(\text{Ni O})_{.75}$ $(\text{Mn O})_{.25}$, Cr. Crystal structure of the weldment is BCC, corresponding to $\alpha\text{-Fe}$ phase. Hence, the weldment is comprised majorly of ferrite (Kožuh *et al.*, 2009).

In general, increasing volume fraction of ferrite (i.e. $\alpha\text{-Fe}$ phase) of the weldments relative to the control sample may be attributed to high welding heat inputs, and hence slow cooling conditions of the solidification process, during which, sufficient time was provided for transformation of $\gamma\text{-Fe}$ to $\alpha\text{-Fe}$ (Atapour *et al.*, 2015). The long peaks of the weldments relative to the control sample was due to grain coarseness that resulted from slow cooling condition of the solidification process. This is because the significant relationship that existed between grains refinement and peak shortening has been attributed to fast cooling conditions, accompanying solidification process (Fowless and Blake, 2008).

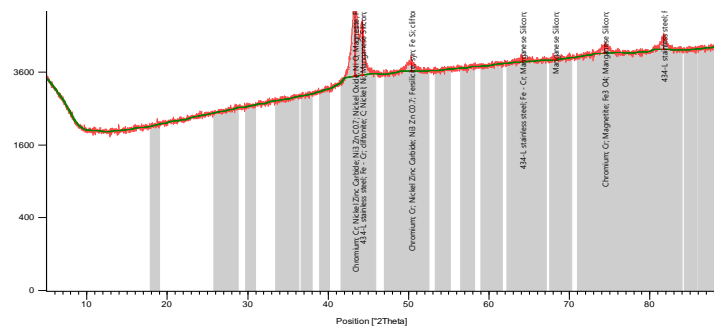


Fig. 3. XRD diffractogram of control sample

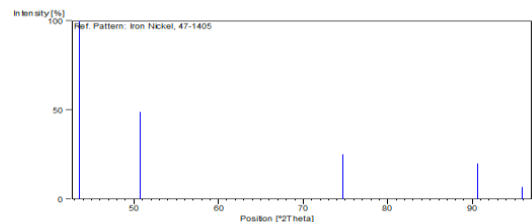


Fig. 4. Stick pattern of the control sample

Table 5. Identified peak list of control sample

S/N.	h	K	l	d(A)	I (%)
1	1	1	1	2.07400	100.0
2	2	0	0	1.79600	49.0
3	2	2	0	1.27000	25.0
4	3	1	1	1.08340	20.0
5	2	2	2	1.03730	7.0

Table 6. Identified patterns list of control sample

Score	Compound	Scale Factor	Chemical Formula
51	Cobalt Iron	0.015	Co ₃ Fe ₇
39	Iron Nickel	0.170	Fe _{0.64} Ni _{0.36}
26	Chromium Iron	0.019	Cr _{1.36} Fe _{0.52}
25	Chromium	0.012	Cr
25	Chromium Silicon	0.902	Cr Si

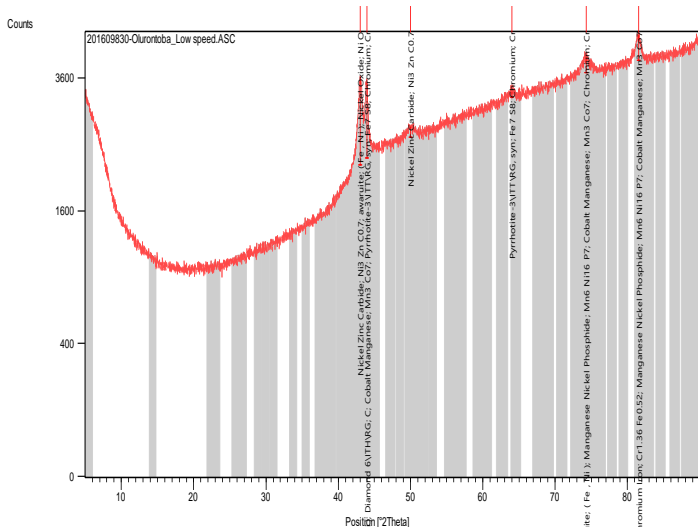


Fig. 5. XRD Spectrum of weldment speed 1.7mm/s

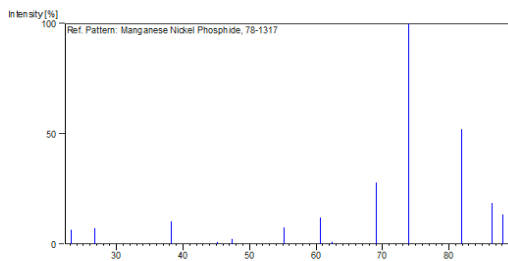


Fig. 6. XRD stick pattern of weldment at speed 1.7 mm/s

Table 7. Identified peak list of weldment at speed 1.7 mm/s

S/N	h	k	l	d[Å]	I [%]
1	1	1	1	3.83880	6.4
2	2	0	0	3.32450	7.2
3	2	2	0	2.35078	10.2
4	3	1	1	2.00475	0.1
5	2	2	2	1.91940	2.3
6	4	0	0	1.66225	7.6
7	3	3	1	1.52539	12.0
8	4	2	0	1.48676	0.2
9	4	2	2	1.35722	27.8
10	5	1	1	1.27960	100.0
11	4	4	0	1.17539	52.0
12	5	3	1	1.18676	18.6
13	6	0	0	1.10817	13.3

Table 8. Identified patterns list of weldment at speed of 1.7 mm/s

Score	Compound Name	Scale Factor	Chemical Formula
46	Diamond	0.804	C
45	Nickel Zinc Carbide	0.546	Ni ₃ Zn C _{0.7}
38	Chromium Iron	0.113	Cr _{1.36} Fe _{0.52}
39	Awaruite	0.247	(Fe , Ni)
27	Manganese Nickel Phosphide	0.204	Mn ₆ Ni ₁₆ P ₇
19	Silicon Oxide	2.107	Si O ₂
21	Cobalt Manganese	0.362	Mn ₃ Co ₇
22	Pyrrhotite-3\ITT\RG, syn	0.391	Fe ₇ S ₈
18	Nickel Oxide	0.420	Ni O
22	Chromium	0.157	Cr

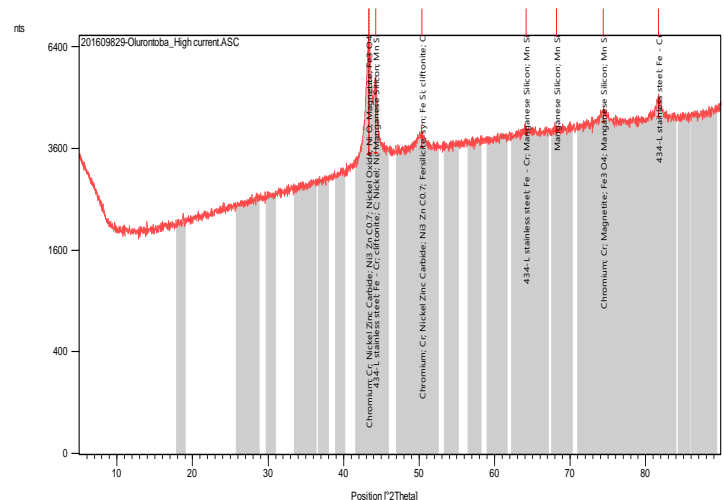


Fig. 7. XRD spectrum of the weldment at current of 200 A

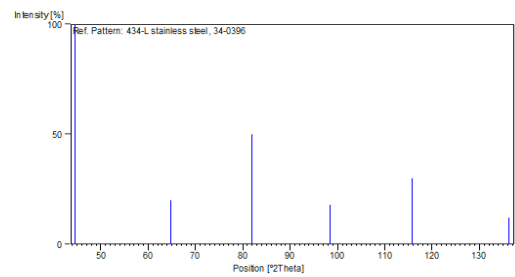


Fig. 8. Stick pattern of the weldment at current of 200 A

Table 9. Identified peak list of the weldment at current of 200 A

S/N	h	k	l	d [Å]	I [%]
1	2	1	1	2.03500	100.0
2	2	0	0	1.43800	20.0
3	2	1	1	1.17430	50.0
4	2	2	0	1.01700	18.0
5	3	1	0	0.90950	30.0
6	2	2	2	0.83020	12.0

Table 10. Identified pattern list of the weldment at current of 200 A

Score	Compound Name	Scale Factor	Chemical Formula
56	304-L stainless steel	0.212	Fe – Cr
52	Chromium	0.614	Cr
29	Nickel Zinc Carbide	0.206	Ni ₃ Zn C _{0.7}
21	Nickel Oxide	0.770	Ni O
18	Fersilicite, syn	0.100	Fe Si
9	Cliftonite	0.035	C
11	Magnetite	0.157	Fe ₃ O ₄
34	Nickel	0.162	Ni
18	Manganese Silicon	0.220	Mn Si

Table 11. Identified peak list of the weldment GTAW at current of 40V

S/N	h	k	l	d [Å]	I [%]
1	1	1	0	2.05000	100.0
2	2	0	0	1.45000	20.0
3	2	2	0	1.03000	10.0
4	2	2	2	0.84000	5.0
5	3	2	1	0.77000	30.0
6	3	3	2	0.62000	2.0

Table 12. Identified pattern list of the weldment at GTAW current of 40V

Score	Compound Name	Scale Factor	Chemical Formula
57	Chromium	0.860	Cr
36	Martensite	0.549	C _{0.14} Fe _{1.86}
25	Chromium	0.183	Cr
17	Nickel	0.416	(Ni O) _{.75}
	Manganese Oxide		(Mn O) _{.25}
24	Silicon Oxide	0.341	Si O ₂
33	Chromium Silicon	0.699	Cr Si ₂

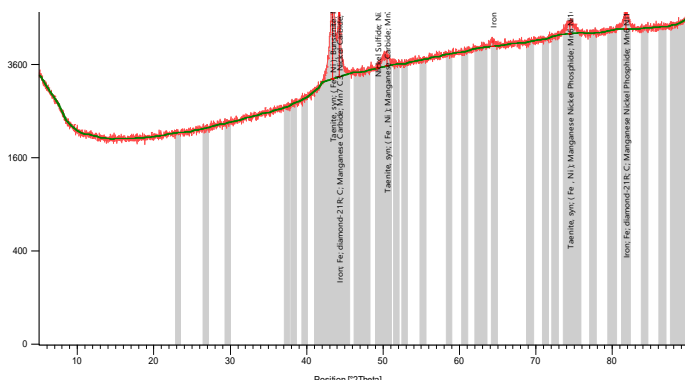


Fig. 9. XRD Spectrum of the weldment at voltage of 40 V

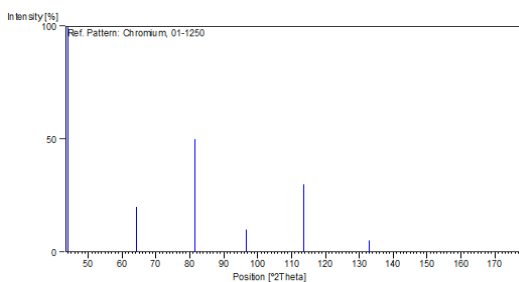


Fig. 10. Stick pattern of the weldment at voltage of 40 V

3.2 MICROSTRUCTURAL ANALYSIS OF THE SAMPLES

Plates 1(A-D) are microstructures of the control sample, and weldments at the speed, current and voltage of 1.7 mm/s, 200 A and 40V respectively. The control sample is characterized by austenite matrix of equiaxed – grains with small amount of δ – ferrite along the grain boundaries. In addition, non-uniformly distributed varying sizes of precipitates are visible within the austenite matrix. Microstructures of weldments, comprising FZs and HAZs are characterised by fine and coarse grains respectively. Similar to the control sample, the microstructures are characterised by a number of δ – ferrite and precipitations of varying sizes, and unlike the control sample, the weldments (FZs and HAZs) are comprised of inclusions and dendrites of different sizes. The small amount of δ – ferrite along the grain boundaries of the control sample may have been introduced intentionally to address problem of hot cracking that resulted from segregation of low melting point elements such as sulphur and phosphorus along the austenite grain boundaries (Lawrence et al., 2016; Kožuh et al., 2009), and

the precipitates may have resulted from thermal history of the sample (Amer et al., 2015; Kurt and Samur, 2013; Hussain, 2010; Kožuh et al., 2009). In general, weld cooling conditions are governed by heat, and hence microstructural heterogeneity and dendrites of different sizes, characterizing the weldment were expected, and they may have resulted from temperature gradient that was generated by the resulting welding heat input of the GTAW technique (Ghusoon et al., 2017; Pauli et al., 2016; Achebo, 2012), and chemical gradients that were due to the GTAW heat input may have contributed to the microstructural characteristics of the weldments (Lawrence et al., 2016; Tabish et al., 2014). The obvious inclusions within the FZs and HAZs of the weldments may be due to possible chemical reactions between dissolved metallic elements such as Fe, Mn, Al, Si and Cr and non-metallic elements such as S and C (Ramesh and Chauhan, 2014; Çalik, 2009). In addition, the inclusions may have resulted from oxidation at the weld pool surface, leading to the combination of some elements, including Mn, Si, Al with oxygen at high temperature to form a single phase oxide of $MnO - SiO_2 - Al_2O_3$ (Costa, 2018; Bhatti et al., 1984). In comparison, GTAW heat input at the welding speed of 1.7 mm/s is high relative to voltage of 40V and current of 200A correspondingly, as evident in the relative numbers of inclusion and dendrite, and number of peak and peak shortening (refer to Tables). Hence, structural characteristic features of the weldments were differently influenced at the GTAW speed, current and voltage.

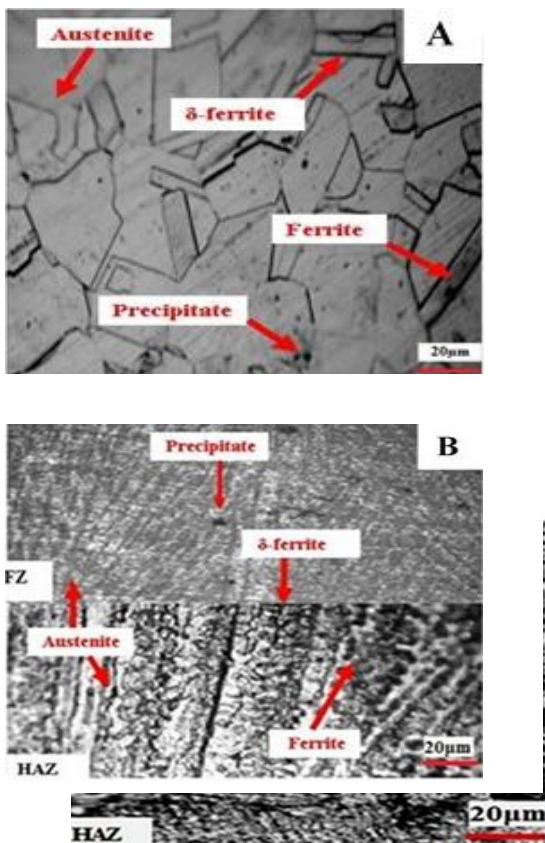


Plate 1 (A-D). Microstructures of the Control weldments at the speed of 1.7 mm/s, 200 A and 40 V of current and voltage respectively at magnification 400x

4 CONCLUSION

Based on the results of the investigations, the following conclusions were drawn:

1. The control sample and weldments are characterized by varying amounts of major and minor compounds.
2. Crystal structure of the control sample is comprised of a mixture of face centred cubic (FCC) and body centred cubic (BCC) with major peak (111). While those of the 1.7 mm/s, 200 A and 40 V produced weldments were comprised majorly of FCC with major peak (511), BCC with major peak (211) and BCC with major peak (110) respectively.
3. Microstructure of the control sample is homogenous, comprising of equiaxed – grained austenite (γ) matrix. Non-uniformly distributed precipitates of varying sizes (dark spots) and small amounts of δ -ferrite (dark lines) are found within the γ matrix and along the γ matrix grain boundaries respectively.
4. Microstructures of the weldments are heterogeneous, comprising of austenite (γ) matrix and ferrite (α) grains, which are dispersed within the γ matrix. Gains of the fusion zones (FZs) microstructures are fine relative to grains of the heat affected zones (HAZs) microstructures, and varied numbers and sizes of precipitates, δ -ferrite, inclusions and dendrites

are visible within the FZs and HAZs microstructures.

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