

The Geochemistry and Petrogenesis of the Iron-Bearing Sediments of Mfamosing, Southeastern (SE), Nigeria: Evidence from Major Oxides and Its Implication for Industrial Utilization

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Abstract: Iron ore, a critical resource for global industrial activities, plays a pivotal role in driving economic development and sustaining essential sectors such as construction, manufacturing, and infrastructure. Nigeria is endowed with substantial iron ore reserves, including the Mfamosing area, which has recently garnered attention for its untapped potential. However, limited comprehensive studies hinder a clear understanding of the iron ore occurrences and their industrial viability. This study addresses this gap by investigating the geochemistry of the iron-bearing metasediments in the Mfamosing area, utilizing X-ray fluorescence (XRF) to analyze major oxides. Field and laboratory studies were conducted, involving the collection of twenty-five (25) sediment samples from the Mfamosing area and subsequent XRF analysis. The results revealed a high content of Fe₂O₃ (hematite) in the range of 62.64–80.45 wt.%, indicating the dominance of iron-rich minerals. The presence of SiO₂, Al₂O₃, and other oxides suggests potential gangue minerals and aids in understanding the ore's composition. The petrogenesis study compares the geochemical characteristics of the Mfamosing iron ore with other iron-bearing formations globally. The findings indicate a sedimentary origin, with hydrothermal influence evidenced by Fe/Al and Fe/Si ratios. The low concentration of detrital materials further supports a primarily seawater-derived iron source. The iron ore has low concentrations of deleterious elements. Classification based on Fe₂O₃ content places most samples in the high-grade category, making them suitable as a primary raw material for steel production. Comparisons with other iron formations in Nigeria and worldwide affirm the Mfamosing iron ore's competitiveness on a global scale.

Keywords: Geochemistry Petrogenesis, Iron ore, Mfamosing, Calabar Flank.

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1.0 Introduction

Iron formations represent significant stratigraphic units primarily consisting of iron-rich chemical sedimentary rock often

referred to as ironstone (Kimberley, 1978a). They are mostly chemical sedimentary rock with over 15% Fe (Bekkar *et al.*, 2010; Cox *et al.*, 2013; Kimberley, 1978; Marion *et al.*, 2021; Planavsky, 1978; Marion *et al.*, 2021; Planavsky, 2012; Posth *et al.*, 2013; Yang *et al.*, 2016).

The intricate geological processes leading to the genesis of iron formations result in diverse deposits (Bekker *et al.*, 2014; Konhauser *et al.*, 2017; Zhang *et al.*, 2014). These include Banded Iron Formations (BIFs). This is a specific type of iron formation, which are sedimentary rocks recognized for their alternating layers of iron-rich minerals and silica-rich minerals, likely formed through the precipitation of iron and silica from ancient seawater in shallow marine environments (Clout & Simonson, 2005; Hagemann *et al.*, 2016; Nanda & Beura, 2021; Sun & Li, 2017; Taner & Chemam, 2015; Yin *et al.*, 2023). Magnetite deposits, characterized by the prevalence of the mineral magnetite (Fe_3O_4), often originate from magmatic processes or the concentration of iron by hydrothermal fluids (Moisescu *et al.*, 2014). Hematite deposits, primarily composed of hematite (Fe_2O_3), can form through various mechanisms, including alteration, hydrothermal precipitation, and metamorphism (Minitti *et al.*, 2005; Öztürk *et al.*, 2016; Xing *et al.*, 2021). Weathered iron ore deposits, like goethite ($\text{FeO}(\text{OH})$) and limonite ($\text{FeO}(\text{OH})\cdot n\text{H}_2\text{O}$), result from the weathering of primary iron minerals under atmospheric conditions (Santoro *et al.*, 2022). Kiruna-type deposits are extensive, high-grade iron ore deposits associated with intrusive magmatic rocks, while Superior-type iron deposits are linked to volcanic and sedimentary rocks in greenstone belts, formed through volcanic and hydrothermal processes. Iron ore deposits result from various geological processes, such as sedimentation, magmatism, and metamorphism, with dominant iron minerals including hematite, magnetite, limonite, and siderite (Rojas *et al.*, 2018). The mineral composition and textural features depend on the formation environment and subsequent

alterations. Iron ore deposits often contain associated elements like manganese, titanium, phosphorus, and sulfur, impacting quality and processing requirements (Liu *et al.*, 2019; Ochromowicz *et al.*, 2021).

Trace elements like arsenic, cadmium, and lead can pose environmental concerns, necessitating additional purification steps (Vishiti *et al.*, 2004; Nkuna *et al.*, 2022). Four types of iron formations are classified based on lithology, composition, associated rocks, and depositional environment: (i) Clinton type and Minette type consist of grained iron formations (GIFs), featuring a granular texture and the absence of laminations/banding; (ii) Algoma-type and Superior-type, known as banded iron formations (BIFs), are mainly banded cherty rocks and contribute significantly to global iron production (Jean-Lavenir *et al.*, 2023). The principal difference lies in the depositional environment, with Algoma-type formations deposited as chemical sediments along with other sedimentary rocks and volcanic material, while Superior-type formations were chemically precipitated on marine continental shelves and in shallow basins (Gourcerol *et al.*, 2016; Li *et al.*, 2022). Algoma- and Superior-type formations share similar mineralogy, with stilpnomelane present only in Algoma-type, indicating potential contamination with volcanoclastic detritus (Wang *et al.*, 2014).

Iron ore serves as a fundamental raw material with diverse applications across various industries. Primarily crucial for steel production, it undergoes smelting in blast furnaces to yield pig iron, the principal ingredient in steel manufacturing. The construction industry heavily relies on steel derived from iron ore for structural components, beams, and reinforcing bars in projects such as bridges and buildings. In transportation, iron and steel find applications in vehicle construction, contributing to the automotive sector, as well as in the production of railways, ships, and aeroplanes. The machinery and equipment manufacturing sector also depends on iron and steel for the fabrication of industrial



machinery and tools. Steel plays a vital role in energy infrastructure, supporting projects like wind turbines, power plants, and oil and gas pipelines. Consumer goods, ranging from appliances to furniture, utilize steel due to its strength and durability. Additionally, iron and steel have applications in containers and packaging, medicine (as supplements and in medical imaging), water treatment processes, fertilizers, and even in the art and decorative industries (Das *et al.*, 2007; Reddy *et al.*, 2019; Thombare *et al.*, 2016). With its versatility, iron ore remains a cornerstone in the global economy, impacting numerous facets of daily life and industrial processes. The rapid economic growth in certain countries like India and China has led to an increased demand for steel in the world market. Consequently, there has been a surge in Fe ore exploration to meet this demand. The banded iron formation (BIF) from the Archaean Ntem Complex (Congo Craton) in the Meyomessi Area, Southern Cameroon, as studied by Sylvestre *et al.* (2018), reveals that iron and silica are the main constituents. The total iron (TFe) contents range from 48.71 to 65.32 wt% (average of 53.29 wt %), making it relatively high-grade or medium-grade iron ore by global standards, suitable for industrial applications (Ndime *et al.*, 2018). The results of BIF from the Mamelles iron ore deposit in the Nyong unit, South-West Cameroon, as investigated by Teutsong *et al.* (2020), show that Fe₂O₃, SiO₂, and Al₂O₃ ranged from 72 to 76.40 wt%, 16.70 to 18.35 wt%, and 2.80 to 5.43 wt%, respectively. TiO₂ varies from 0.14 to 0.18 wt%, and P₂O₅ from 0.13 to 0.34 wt%. This study indicates that the Mamelles deposit contains medium-grade iron ores with acceptable contents in contaminants (Teutsong *et al.*, 2021). Studies reveal that Itakpe ore is richer in iron than either the Agbaja or the Corby ores. Itakpe is a fairly high-grade, acidic (or siliceous) ore, whereas Agbaja is a low-grade, high-phosphorus, acidic ore, and Corby is a low-grade, high-phosphorus ore that is basic (or calcareous) due to its high calcite content (Adedeji & Sale, 1984). The Maru BIF shows Fe₂O₃ ranging from 44.07 to 58.41%, SiO₂ from

24.74 to 42.67%, with relatively high values of Al₂O₃ (2.00 to 8.97%) and MnO (2.29 to 9.37%) (Adekoya, 1998). The Maru BIF is low to moderate grade and will require advanced techniques for processing.

Nigeria is endowed with substantial iron ore reserves, including the Mfamosing area, which has recently garnered attention for its untapped potential. However, limited comprehensive studies hinder a clear understanding of the iron ore occurrences and their industrial viability. This study addresses this gap by investigating the geochemistry of the iron-bearing metasediments in the Mfamosing area.

1.1 Location of the study area

The study was carried out within the Mfamosing areas, between latitude N 5° 5' 00" to N 5° 10' 0" and longitude E8° 24' 0" to E8° 25' 0". It is situated in the South Eastern region of Nigeria (Fig. 1). The area is about 30km away from the main settlement, and it is accessed by the use of motorcycles. These areas have been identified as holding significant iron ore occurrences hosted within the Mfamosing formation of the Calabar Flank.

1.2. Geologic settings

The earliest investigations into the geology of the Calabar Flank were conducted by Reyment (1955) and Dessauvage (1965). They established the basin's biostratigraphic context. Although Reyment (1955) initially regarded it as part of the Benue Trough, Murat (1972) later believed that the Calabar Flank experienced a distinct phase of tectonism and stratigraphic evolution compared to the adjacent Anambra Basin and the lower Benue Trough, thereby categorizing it as a separate basin (Ekwok *et al.*, 2021; Okon *et al.*, 2022). Nyong and Ramanathan (1985) defined the Calabar Flank as the easternmost segment of the Gulf of Guinea, which constitutes part of the Nigerian continental margin. This area is sandwiched between the Cameroon volcanic line to the east, the Ikpe platform to the west, and the Oban Massif and Calabar hinge line to the north and south, respectively.



The Calabar Flank is a relatively compact Nigerian sedimentary basin which lies along the fringes of the Gulf of Guinea. A significant divergence between the Calabar Flank and the southern Benue Trough originates from the initial rifting of the southern Nigerian margin. This event gave rise to two primary sets of faults, notably the NE-SW and NW-SW trending fault sets that characterize the Benue Trough and the Calabar Flank respectively (Murat, 1972). Within this context, Murat (1972) identifies the tectonic components of the Calabar Flank as including the Ikang Trough and the Ituk High, which represented a mobile depression and a stable mobile submarine ridge. They further postulated that the Calabar Flank is underlain by horsts and graben structures. Sedimentary processes within the Calabar

Flank commenced with the deposition of fluvio-deltaic clastics known as the Awi Sandstone, likely of Aptian age, overlying the Precambrian crystalline basement complex known as the Oban Massif. This phase was succeeded by a mid-Albian marine transgression, leading to the accumulation of the Mfamosing Limestone. This limestone was primarily deposited on horst structures and relatively stable platforms. (Ekwok *et al.*, 2019; Adamu *et al.*, 2021). Above the Mfamosing Limestone is a substantial sequence of black to grey shale known as the Ekenkpon Formation (Petters & Reijers, 1987). This formation displays minor occurrences of marls, calcareous mudstone, and oyster beds deposited during the late Cenomanian-Turonian. (Petters & Reijers, 1987).

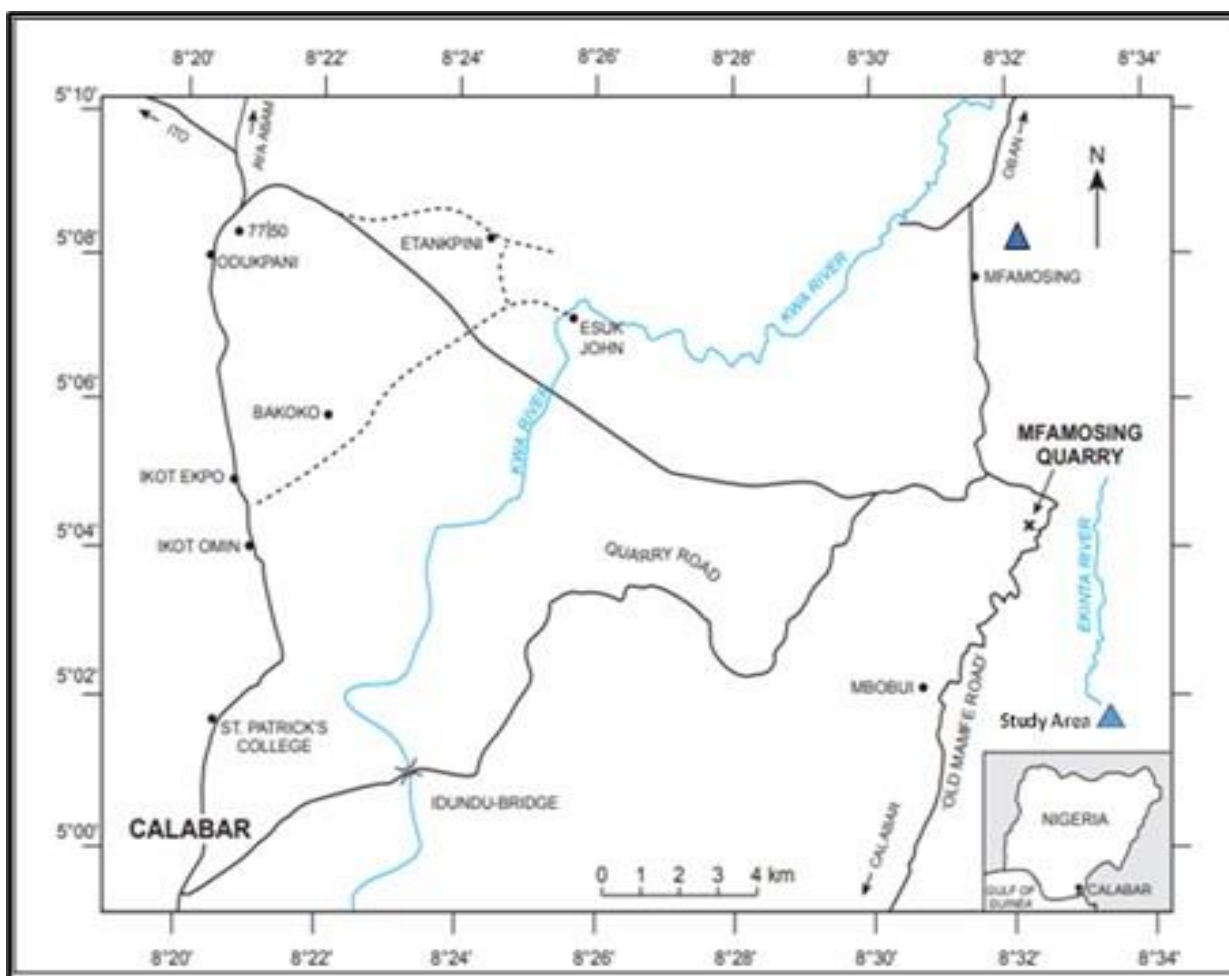


Fig. 1. Map of southeastern Nigeria showing the location of the study area (William *et al.* 2019)



Ukpong and Ekhalialu (2015) further confirm this age using foraminifera and palynomorph analysis. The Ekenkpon Shale is overlain by a thick marl unit referred to as the New Netim Marl. This unit is characterized by nodules and shale at its base, with intermittent thin layers of shale in its upper portion (Harry, 2022). Foraminiferal analysis indicates an early Coniacian age for the New Netim Marl (Anoh & Petters, 2014). Following an unconformity, the carbonaceous dark grey shale of the

Nkporo Formation overlays the New Netim Marl. This shale unit was deposited during the late Campanian-Maastrichtian interval and serves as the uppermost layer of the Cretaceous sequence within the Calabar Flank. Finally, the Nkporo Shale sequence is succeeded by a pebbly sandstone unit representing the Tertiary Benin Formation as seen in Fig. 2 (Edegbai *et al.*, 2019; Omietimi *et al.*, 2022; Uzoegbu *et al.*, 2023; Omang *et al.*, 2023; Omang *et al.*, 2023)

AGE	LITHOLOGY	DESCRIPTION
Recent Eocene -	Benin Formation	Loose sands, pebbly and arkosic
Maastrichtian L. Campanian -	Nkporo Shale	Dark grey, very fissile carbonaceous shale with gypsum bands and some calcareous nodules
Santonian	Santonian Deformation	Santonian deformational episode characterized by period of folding of pre-existing rocks and erosion and/or non deposition.
Coniacian	ODUKPANI GROUP	New Netim Marl
Turonian		Ekenkpon Shale
Cenomanian		Un-named Shale
Mid - Albian		Mfamosing Limestone
Neocomian - Aptian	Awi Formation	Reddish brown, coarse to medium grained arkosic sandstone. Pebbly at the base and exhibit fining upward succession in cycles, graded bedding.
Precambrian	Lower Cretaceous Basement Complex	Southeastern Basement Complex – Oban Massif composed predominantly of granite gneisses, granites and granodiorites.

Fig. 2. Stratigraphic chart of the Calabar Flank (Modified after Petters *et al.*, 2014)

2.0 Materials and Methods

The methodology encompassed comprehensive fieldwork, alongside geochemical analyses of rock samples. During the field study, twenty-five (25) rock samples weighing 5 to 10 kg before removing the weathered phases were meticulously collected from various rock outcrops within Mfamosing. Collected samples were cleansed to remove visible allogenic material and observed weathered phases before transportation to the laboratory. All the samples were subjected to geochemical analysis. For the geochemical analysis, approximately 2 kg of each representative rock sample was fragmented into thumb-nail-sized pieces using a hardened steel hammer. These pieces were then divided, with one part

reserved for reference purposes and the other crushed and ground to attain a particle size finer than -60 mesh, facilitated by a "jaw-

crusher". Each sample was powdered to -200 mesh and homogenized, exercising utmost care to prevent cross-contamination between samples. The homogenized ore samples were then quartered for subsequent analyses. Major element composition in the studied rocks was determined using the XRF technique.

Dehydrated powdered samples were fused with lithium tetraborate in a platinum alloy crucible at 1100°C to produce fused beads for analysis. The XRF raw data were corrected using "alpha factors" for inter-element correction. Major elements like Silica (SiO₂), Aluminum oxide (Al₂O₃), Iron (III) oxide (Fe₂O₃), Manganese oxide (MnO), Magnesium oxide (MgO), Calcium oxide



(CaO), Sodium oxide (Na₂O), Potassium oxide (K₂O), and Sulphur oxide (SO₂) were measured as oxides. Calibration was achieved using certified international reference rock materials.

3.0 Results and Discussion

The result of the geochemical analysis of the Mfamosing iron formation is presented in Table 1.

3.1. Major Oxide Geochemistry

The predominant bulk chemical composition of the analyzed sample reveals a substantial abundance of Fe₂O₃ (hematite), ranging from 62.64 to 80.45 wt.% and averaging at 70.98 wt.%. This signifies a notably high iron content, characteristic of hematite-rich deposits. The concurrent presence of SiO₂, within the range of 5.49 to 11.58 wt.%, suggests the occurrence of quartz—a common gangue mineral in iron ores known for its association with hematite. Additionally, the presence of Al₂O₃ (3.53 to 7.26 wt.%) points towards the potential existence of mineral phases such as clays or aluminosilicates, acting as gangue minerals (Hussin *et al.*, 2018). The observed low concentrations of oxides like MgO, CaO, MnO, and TiO₂ in the sample suggest a minimal influence of contamination from other rock types/minerals like carbonate rocks, apatite, pyroxene or hydrothermal alteration processes (Omotunde, 2020; Omang *et al.*, 2023; Vural, 2023). This emphasizes the purity and distinctiveness of the analyzed material.

An intriguing aspect arises from the weak negative correlation observed between Fe₂O₃ and SiO₂ ($r = -0.3$), Al₂O₃ ($r = -0.06$), and CaO ($r = -0.39$), as indicated in Table 2. This correlation implies an inverse relationship between these oxides and iron minerals. Such a trend suggests a possible chemical precipitation mechanism during the formation of iron ores, wherein iron oxides may have replaced or excluded SiO₂, Al₂O₃, and CaO (Fiege, 2019; Madondo *et al.*, 2021). This phenomenon could offer valuable insights into the geological processes that led to the ore's composition and formation. Comparative analysis with other iron formations, namely

Gangfelum BIF, Maru BIF, Muro BIF, and Kankun BIF (detailed in Table 3), highlights the distinctive nature of the analyzed sample (Bolarinwa, 2017). Specifically, the Fe₂O₃ content in the studied samples surpasses that of the mentioned iron formations, underlining its unique composition within the broader geological context. This disparity in iron content further accentuates the significance and potential economic value of the analyzed deposit.

3.2. Petrogenesis

The SiO₂-Al₂O₃-FeO ternary diagram, based on the classification system proposed by Govett *et al.* 1966 (Fig. 3), provides valuable insights into the age and genesis of the Mfamosing iron ore. The positioning of the ore in the diagram strongly indicates a post-Precambrian age, aligning with the observed characteristics of the deposit (Bafon *et al.*, 2023).

The identified weak negative correlation between Fe₂O₃ and SiO₂ ($r = -0.3$) aligns well with a sedimentary origin for the Mfamosing iron ore, as illustrated in Fig. 4.

In sedimentary settings, such correlations often suggest a transformation process where iron minerals precipitate, replacing silicate minerals like quartz. This interpretation adds depth to the understanding of the ore formation mechanism, with sedimentary processes playing a significant role in the composition of the deposit. The distinctive composition of the Mfamosing iron ore, characterized by extremely low concentrations of other oxides and the dominance of Fe₂O₃, which is in line with the attributes commonly associated with sedimentary iron ores

This is in contrast to magmatic iron ores, which typically exhibit a broader range of oxide compositions, including various metal oxides (Skirrow, 2022). The high Fe/Al (19.26) and Fe/Si (14.46) ratios, in comparison to the Si/Al ratio (1.37), strongly suggest a notable hydrothermal influence in the ore formation process, as indicated by (Hatton & Davidson, 2004; Jansson & Allen, 2011). These ratios serve as robust indicators of the involvement of hydrothermal fluids,



potentially contributing to the enrichment of iron in the ore.



Table 1: Geochemical Characteristics of Mfamosing Iron Ore

Sample ID	M14	M15	M16	M17	M18	M19	M20	M21	M22	M23	M24	M25	MEAN	Range
SiO₂	5.68	6.83	7.91	6.03	6.65	11.58	8.15	8.83	8.05	10.26	9.25	7.17	7.70	5.49-11.58
Al₂O₃	3.53	5.32	5.68	4.41	5.05	7.26	5.33	5.62	5.12	4.93	5.12	4.45	5.01	3.53-7.26
Fe₂O₃	72.20	70.10	73.12	71.79	72.35	70.32	73.15	70.60	71.44	70.19	73.14	74.75	70.98	62.64-80.05
CaO	0.01	0.05	0.10	0.02	0.12	0.43	0.11	0.07	0.09	0.20	0.13	0.01	0.10	0.00-0.43
MgO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.00
SO₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.00
K₂O	0.00	0.03	0.06	0.03	0.00	0.05	0.02	0.07	0.07	0.08	0.08	0.05	0.05	0.00-0.11
Na₂O	0.02	0.01	0.02	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.03
TiO₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.00
Mn₂O	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.00
P₂O₅	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00-0.00
LOI	13.52	13.43	13.24	14.25	12.90	12.30	12.48	13.00	12.85	14.07	12.01	13.98	13.52	12.01-15.70
Total	94.96	94.90	99.95	98.54	97.06	99.92	99.23	98.18	97.61	99.74	99.73	100.00	-	-
Si/Al	1.42	1.13	1.23	1.21	1.16	1.41	1.35	1.39	1.39	1.84	1.60	1.42	1.37	1.07-2.01
Fe/Al	27.08	17.45	17.05	21.56	18.97	12.83	18.17	16.63	18.48	18.85	18.92	22.24	19.26	

Table 2. Pearson's correlation matrix for major element oxides in the study area

	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	K ₂ O	Na ₂ O	LOI
SiO ₂	1						
Al ₂ O ₃	0.528194	1					
Fe ₂ O ₃	-0.29988	-0.05919	1				
CaO	0.806823	0.615551	-0.38987	1			
K ₂ O	0.3516	0.160537	-0.53162	0.34558	1		
Na ₂ O	-0.39446	-0.33234	0.341497	0.34252	0.26954	1	
LOI	-0.54975	-0.4018	0.428354	0.56692	0.30244	0.622269	1

Table 3. Comparison of the study area major oxides (wt-%) with BIF types in Nigeria (Anthony, 2018)

Location	This study	Gangfelum BIF	Maru BIF	Muro BIF	Kakun BIF
SiO ₂	7.70	41.98	34.35	57.66	20.17
TiO ₂	0.00	0.37	0.16	0.02	1.74
Al ₂ O ₃	5.01	1.41	4.25	0.28	8.12
Fe ₂ O ₃ (t)	70.98	53.91	54.3	42.02	56.8
MnO	0.00	0.09	4.83	0.06	0.07
MgO	0.00	0.02	0.13	0.01	3.58
CaO	0.10	0.05	0.11	0.02	6.68
Na ₂ O	0.00	0.18	0.01	0.01	1.1
K ₂ O	0.05	0.44	0.65	0.02	0.77
P ₂ O ₅	0.00	<0.01	0.09	0.05	0.27

The plotting of the iron ore samples within the deep-sea pelagic sediment field in the SiO₂-Al₂O₃ discrimination plots, following the classification by Bostrom *et al.*, 1973 (Fig. 5), further supports the hypothesis that the iron in the Mfamosing deposit primarily originated from dissolved iron in seawater (Tchouakui *et al.*, 2022). This interpretation implies a limited input of detrital materials such as quartz and aluminium-rich clays, emphasizing the marine influence on the ore's composition. The low Al₂O₃+TiO₂ value (5.01 wt.%), in contrast to the Fe/Al and Fe/Si ratios, serves

as additional confirmation of the reduced input of detrital materials, highlighting the purity of the iron ore. Furthermore, the role of hydrothermal fluids in ore formation is evident in Fig. 6, illustrating the potential interaction of hot, mineral-rich fluids from hydrothermal vents with seawater, leading to the precipitation of iron oxides. This multifaceted approach to analyzing the Mfamosing iron ore provides a comprehensive understanding of its genesis, age, and the geological processes that shaped its unique composition.



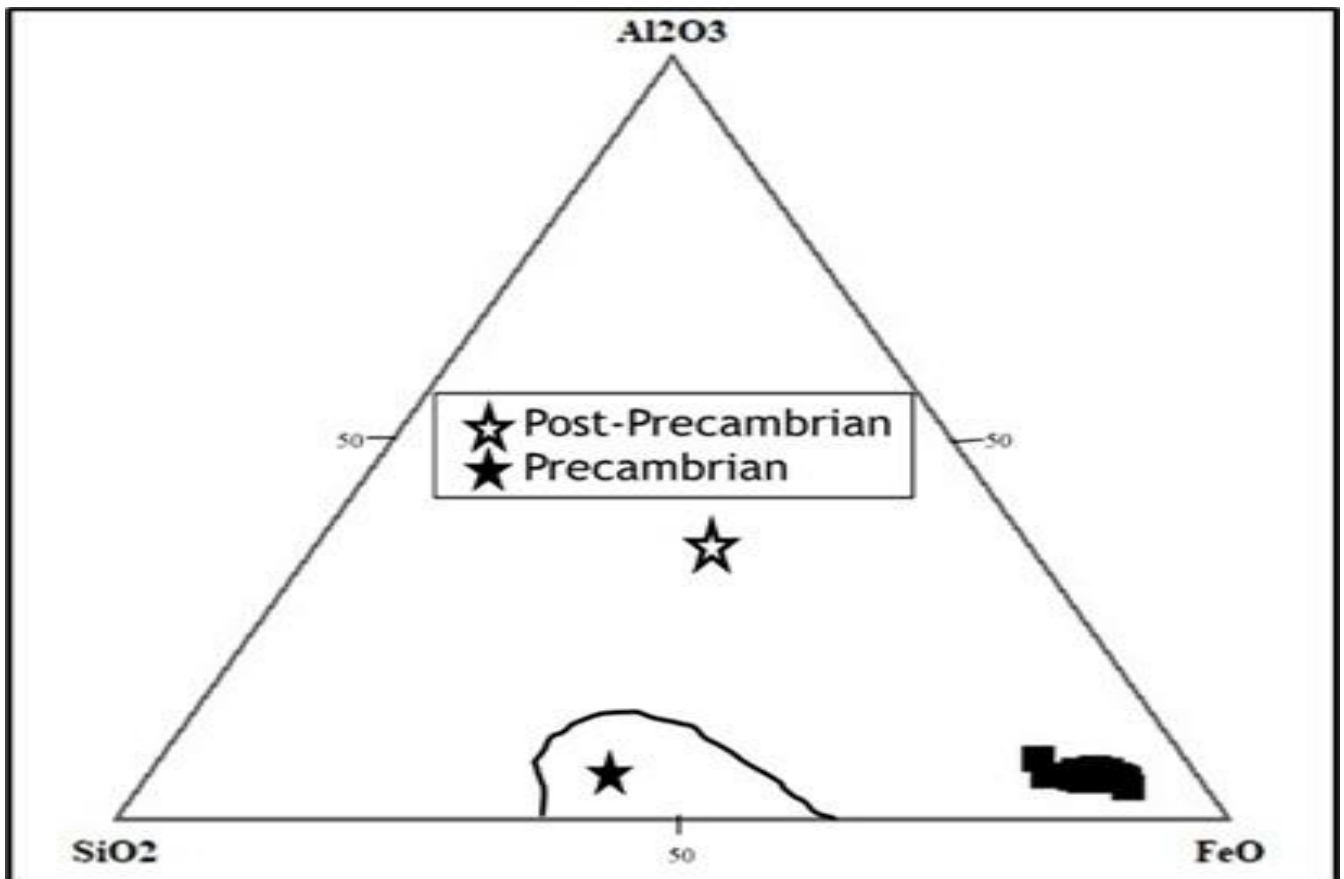


Fig. 3. SiO₂-Al₂O₃-FeO ternary diagram showing the post-Precambrian iron bearing metasediment in Mfamosing (field after Govett *et al.*, 1966)

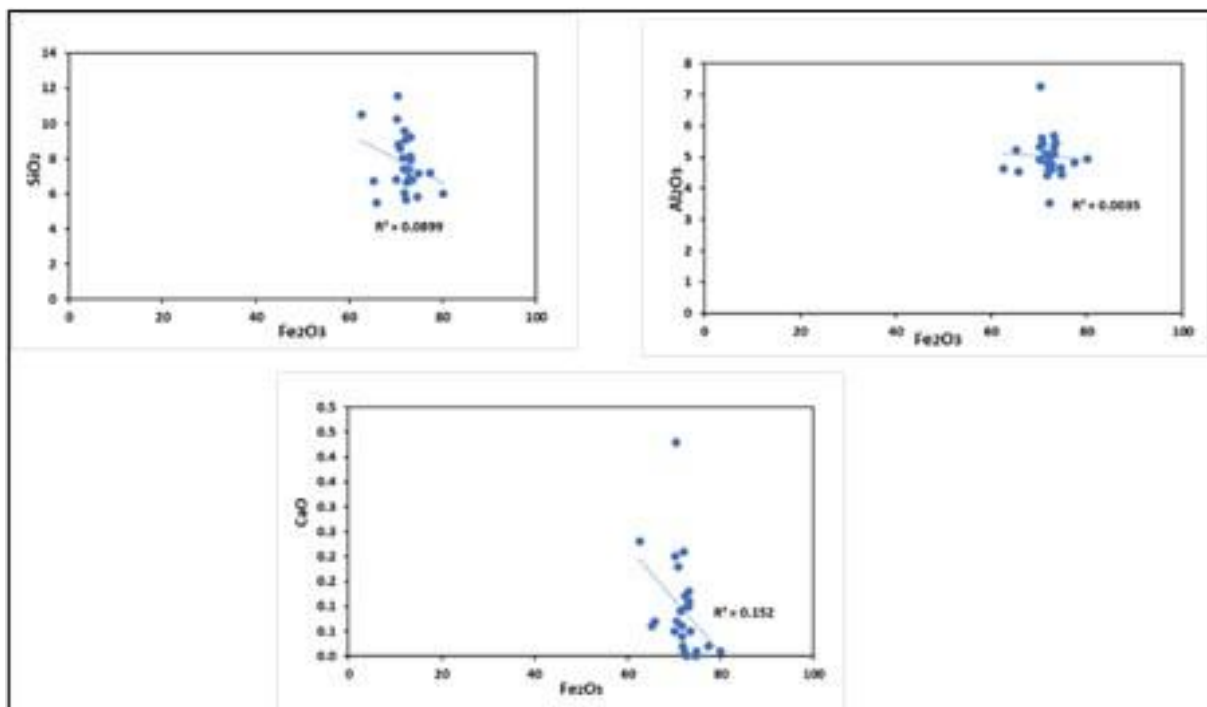


Fig. 4. Selected binary plots of major oxides against Fe₂O₃ for Mfamosing iron formation



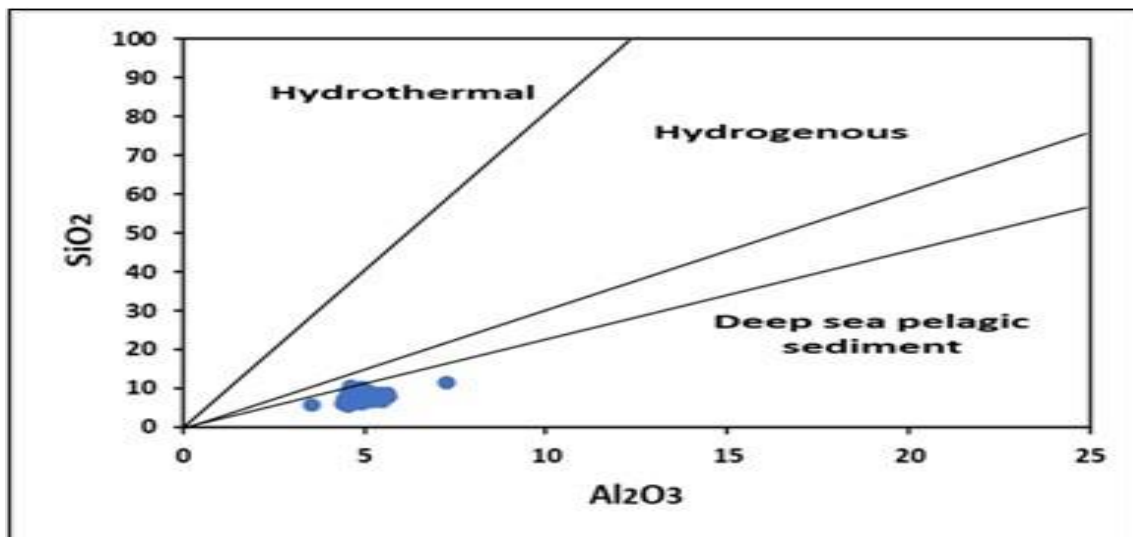


Fig. 5. SiO₂ - Al₂O₃ discrimination diagram indicating the deep-sea pelagic sediment of Mfamosing iron formation (Field after Bostrom et al., 1973)

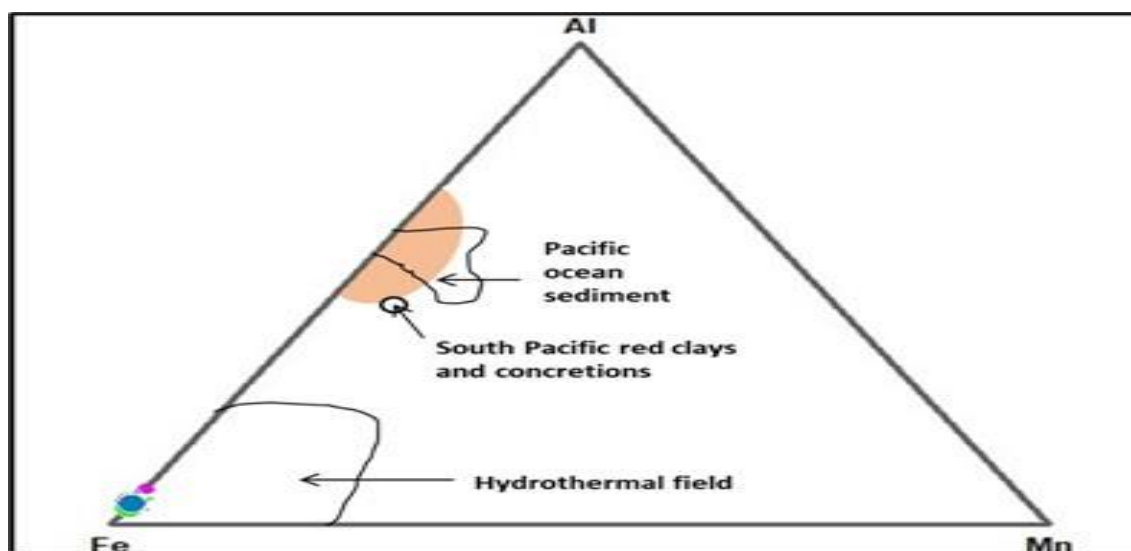


Fig. 6. Ternary Fe-Mn-Al plot showing hydrothermal affinity of Mfamosing iron formation (Field after Bonatti et al., 1975)

3.3. Implication for Industrial Utilization

The geochemical analysis of iron ore samples from the Mfamosing areas not only sheds light on the ore's composition but also proves instrumental in evaluating its suitability for various industrial applications. According to the criteria proposed by Guider (1981) and Dobbins *et al.* (1982), commercial iron ore should ideally have tolerable concentrations of deleterious elements, with limits set at 0.07 wt-% for phosphorus (P) and 0.1 wt-% for sulfur (S) (Riposan *et al.*, 2013). The remarkably low concentrations of these deleterious elements in the analyzed samples

indicate the ore's high quality, indicating that it is relatively free of impurities which is capable of adversely affecting industrial processes. Iron ores are conventionally classified into three categories, as outlined by Robinson *et al.* (2020): high-grade (Fe₂O₃ content above 65 wt-%), medium-grade (Fe₂O₃ concentration fluctuating between 52 and 65 wt-%), and low-grade (Fe₂O₃ values below 52 wt-%) (Angerer *et al.*, 2021). Most samples from the study area fall within the high-grade class, attesting to the exceptional iron content. This classification is supported by the whole-rock (MgO + CaO + MnO)/Fe₂O₃ total versus SiO₂/Fe₂O₃ total



discrimination diagram in Fig. 7, following the approach by Angerer *et al.* (2012).

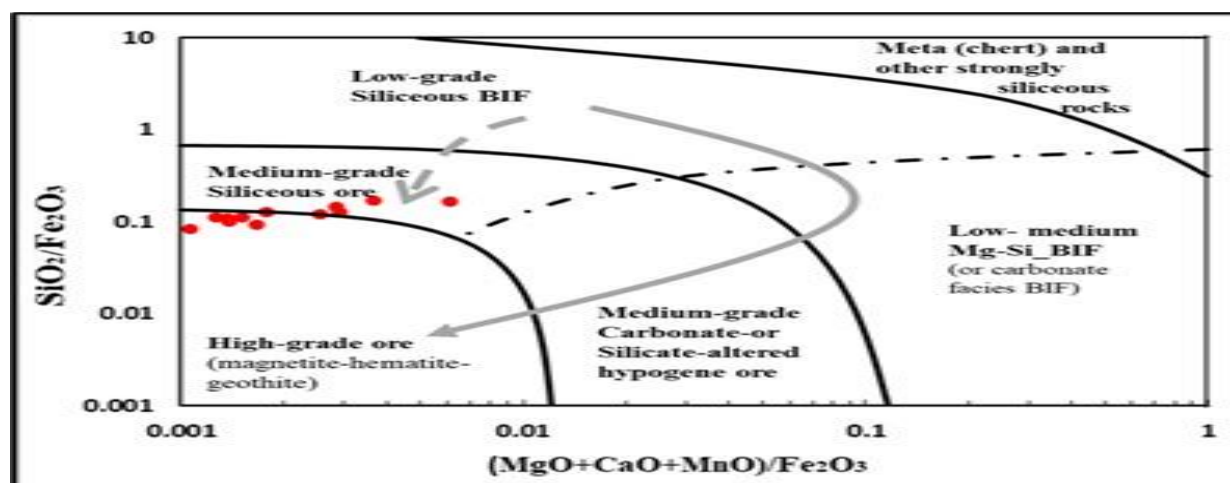


Fig. 7. Whole-rock (MgO + CaO + MnO)/Fe₂O₃ total versus SiO₂/Fe₂O₃ total discrimination diagram (Angerer *et al.*, 2012)

Comparative analysis with other iron formations in Nigeria, such as Gangfelum BIF, Maru BIF, Muro BIF, and Kankun BIF, revealing that the iron ore from the study area boasts higher iron content and a reduced presence of gangue minerals (Angerer *et al.*, 2022). Fig. 8, shows the spatial distribution of Mfamosing iron ore compared with

Gangfelum BIF, Maru BIF, Muro BIF, and Kankun BIF. Similarly, when compared to international iron formations like Endergue BIF, Elom BIF, and Um Arab IF, the study area stands out with its superior iron content and minimal gangue minerals as outlined in Table 4.

Table 4. Comparison of the study area major oxides (wt-%) with IF and BIF types areas of the world (Robinson *et al.*, 2020)

Location	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃ (t)	MnO	MgO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅
This Study	7.70	0.00	5.01	70.98	0.00	0.00	0.10	0.00	0.05	0.00
Endergue IF	36.77	0.19	1.95	56.29	0.10	0.77	0.16	0.0	0.07	0.07
Ngovayang IF	46.84	0.51	3.76	43.68	0.07	2.69	1.49	0.28	1.08	0.24
Elom BIF	39.98	0.03	0.48	58.19	0.07	0.06	0.02	0.01	0.02	0.08
Bikoula BIF	37.77	0.36	2.88	49.97	0.48	4.44	3.15	0.85	0.76	0.11
Nkout BIF	40.12	0.03	0.83	53.88	0.09	2.61	1.07	0.10	0.26	0.08
Um Anab (Egypt) IF	54.62	0.20	3.60	35.14	0.07	1.28	2.98	0.25	0.12	0.54
Jerome (USA) IF	18.16	0.04	1.04	77.78	0.13	0.09	0.92	0.09	0.22	/
Algoma BIF	50.50	/	3.00	41.33	0.22	1.53	1.51	0.31	0.58	0.21
Superior BIF	47.20	/	1.39	44.51	0.73	1.24	1.58	0.12	0.14	0.06



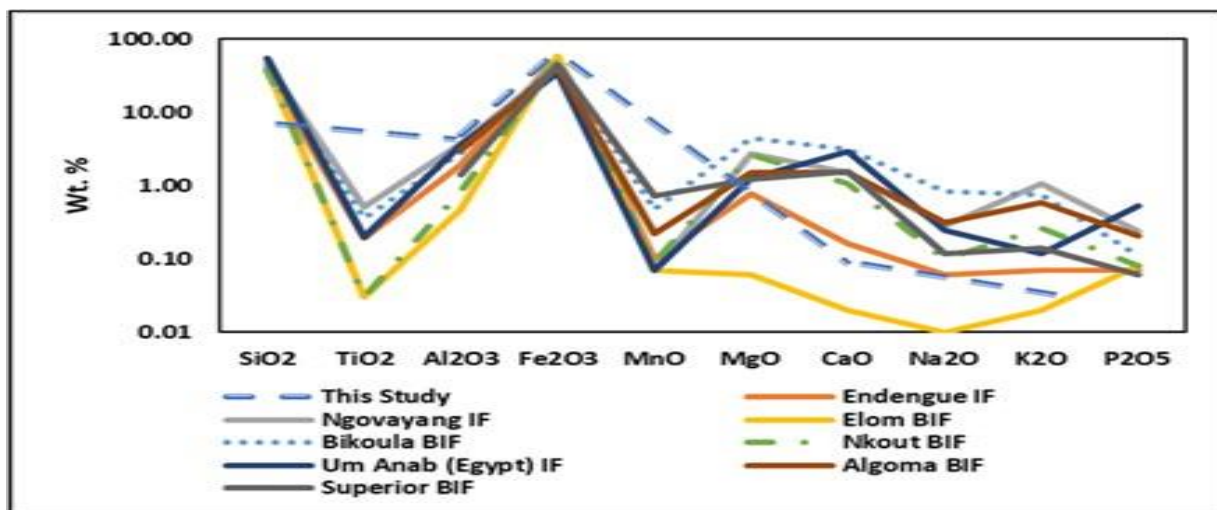


Fig. 8. Spider plot showing the variation of the study area major oxides (wt-%) with BIF types in Nigeria

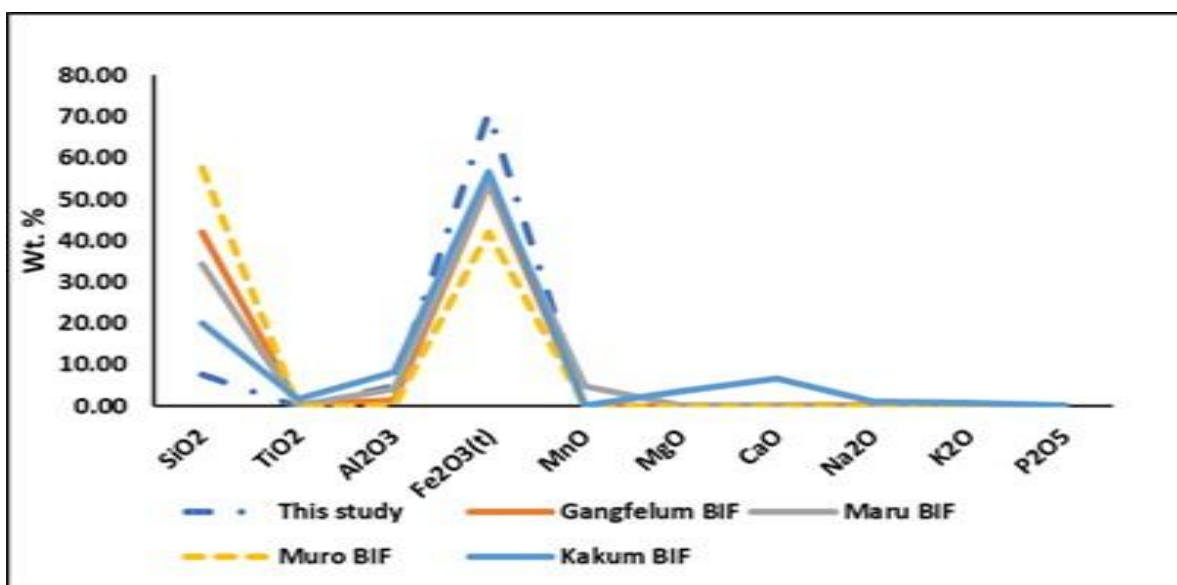


Fig. 9. Spider plot showing the variation of the study area major oxides (wt-%) with IF and BIF types areas of the world

The spatial distribution of Mfamosing iron ore compared with global data is shown in Fig. 9. The iron content of the Mfamosing iron ore is also competitive on a global scale, aligning with major productive iron formations worldwide, including Quadrilátero Ferrífero (Brazil), Carajás (Brazil), Hamersley (Australia), Krivoy Rog (Ukraine), Bailadila (India). Also, Steel, being one of the most versatile and crucial materials in the global economy, would benefit from the high-quality ore extracted from the Mfamosing region, positioning it as a valuable resource in the global industrial landscape.

4.0 Conclusion

The detailed analysis of the Mfamosing iron ore presents a compelling picture of a deposit distinguished by its remarkably high hematite content, averaging 70.98 wt.%. The coexistence of quartz and aluminosilicates as gangue minerals, coupled with minimal contamination from other oxides, underscores the ore's purity and uniqueness. The observed inverse correlations between Fe_2O_3 and SiO_2 , Al_2O_3 , and CaO suggest a potential chemical precipitation mechanism during ore formation, providing valuable insights into the geological processes shaping the deposit. Comparative analyses with other iron



formations highlight the economic significance of the Mfamosing deposit, emphasizing its exceptional iron content and distinctive composition globally. The ore's sedimentary origin is supported by the SiO₂-Al₂O₃-FeO ternary diagram, aligning with a post-Precambrian age. Hydrothermal influence, evidenced by high Fe/Al and Fe/Si ratios, further enhances the ore's richness and purity, emphasizing marine influence and minimal detrital material input. The ore's high quality, meeting industry standards for deleterious elements, classifies it as predominantly high-grade iron ore, making it a valuable resource for global steel production and cement industries. The Mfamosing iron ore stands out on both regional and international scales, positioning itself as a key contributor to the global industrial landscape.

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Compliance with Ethical Standards

Declaration

Ethical Approval

Not Applicable

Competing interests

The authors declare that they have no known competing financial interests

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Availability of data and materials

Data would be made available on request.

Authors' contributions

Benjamin Odey Omang: Project conceptualization, supervision and manuscript second draft. Temple Okah Arikpo: Data extraction and analysis. Eyong Gods'will Abam: plotting and interpretation of plots. Asinya Enah Asinya: Manuscript first draft. Godwin Terwase Kave: Manuscript proofreading. Anthony Adesoji Onansonwo: Revision

