

IMINOFORMYLATION PRODUCTS OF THIOAMIDES, PART 17 (1):
 SYNTHESIS OF 5-COUMAROYLTHIOPHENES FROM
 3-AMINOTHIOACRYLAMIDES AND THEIR DERIVATIVES

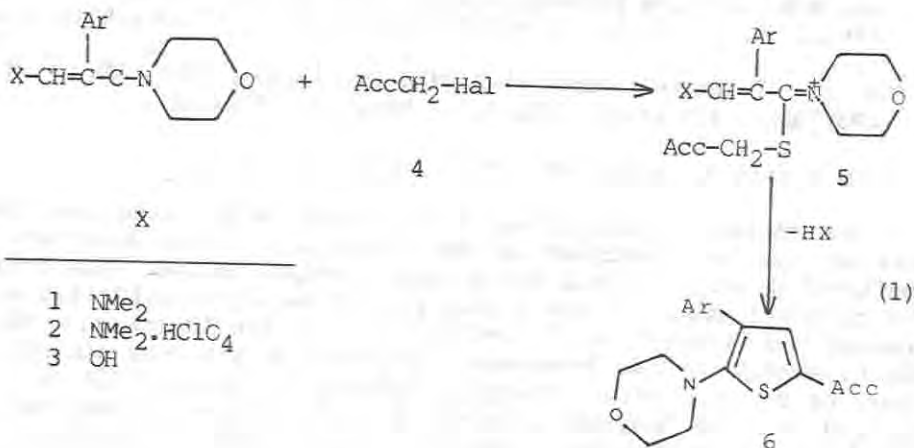
Jürgen Ibscher^{a*}, Alexander Knoll^a, Berhanu Abegaz^b and Peter Czerney^c, ^aSektion Chemie, Humboldt Universität zu Berlin, DDR-1040 Berlin, Hessische Str. 1-2, German Democratic Republic; ^bDepartment of Chemistry, Addis Ababa University, P.O.Box 1176, Addis Ababa, Ethiopia; ^cSektion Chemie, Friedrich-Schiller-Universität Jena, DDR-6900 Jena, German Democratic Republic.

(Received May 11, 1987)

ABSTRACT: The reaction of 3-amino and 3-hydroxythioacrylamides with 3-bromoacetyl coumarin results in the formation of new 2-coumaroyl-5-morpholinothiophenes. Their structure is elucidated by spectroscopic methods and by microanalysis.

INTRODUCTION

Recently we found an efficient synthesis of *N,N*-disubstituted 2-aminothiophenes **6** (2-4) based on the reaction of acceptor substituted methyl halides **4**, such as α -haloethylketones, α -haloacetates, bromonitromethane, or substituted benzyl halides, with 3-aminothioacrylamides **1**, their hydroperchlorates **2** or 3-hydroxythioacrylamides **3**. In some cases intermediate salts **5** could be isolated (2).



The reactants 1(5,6), 2(5,6) and 3(5,7) can easily be prepared from substituted thioacetamides which are iminoformylated by formamide chlorides. In continuation of our investigations in the synthesis of *N,N*-disubstituted thiophenes 6 following reaction (1) as well as of coumaroylheterocycles (8) we have achieved the preparation of coumaroylthiophenes.

RESULTS AND DISCUSSION

When ethanolic solutions of 3-bromoacetylcoumarine 7, 3-aminothioacrylamide hydroperchlorates 2 or 3-hydroxythioacrylamides 3 and triethyl amine are heated a smooth reaction (2) takes place. The crystalline products obtained are already the cyclized thiophenes 8. Intermediates, such as alkylation products 5 (Acc = 3-coumaroyl) were not isolated. High yields of the 2-coumaroylthiophenes 8 are achieved which can easily be purified by recrystallization.

Table: 2-Coumaroyl-5-morpholinothiophenes 8

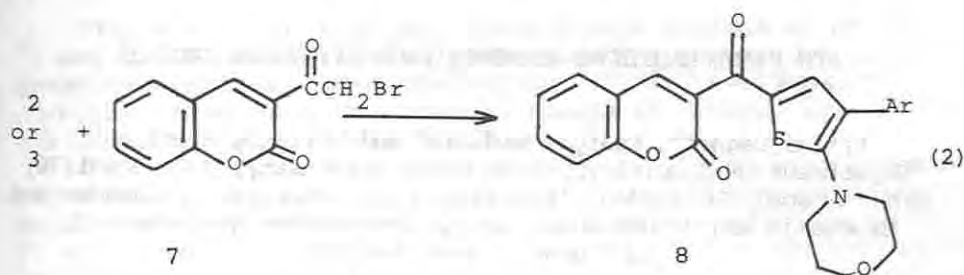
	Ar	Yield/reactant [%]	m.p. [°C]	¹ H NMR (CDCl ₃) [δ ppm]
8a ¹	C ₆ H ₅	77/2	200-202 (Ethanol)	3.01 (m; 4H), 3.65 (m; 4H), 7.38 (m; 1H), 7.96 (s; 1H)
8b ²	4-CH ₃ C ₆ H ₄	79/2	199-200 (Acetonitrile)	2.55 (s; 3H), 2.99 (m; 4H), 3.63 (m; 4H), 7.29 (m; 9H), 7.93 (s; 1H)
8c	4-ClC ₆ H ₄	94/2 94/3 ³	180-182 (Ethanol)	3.02 (m; 4H), 3.63 (m; 4H), 7.42 (m; 9H), 7.96 (s; 1H)
8d	4-CH ₃ OC ₆ H ₄	92/2	158-161 (Ethanol)	2.98 (m; 4H), 3.64 (m; 4H), 3.70 (s; 3H), 6.78 (d; 2H), 7.38 (m; 7H), 7.91 (s; 1H)

¹UV (Acetonitrile) λ_{max} nm (log ε): 224(4.26), 279(4.29), 387(4.23).
MS: m/e (relative intensity) 418(29.1), 417(M⁺, 100), 358(19.3), 186(33.7), 173(78.1), 115(28.5), 101(22.1), 89(27.4), 63(17.5).

²MS: m/e (relative intensity) 432(28.9), 431(M⁺, 100), 200(29.3), 173(78.1), 129(14.6), 115(11.5), 89(13.0), 44(15.9).

³1.01 g triethyl amine was used instead of 2.02 g.

The coumarylthienylketones 8 are stable yellow compounds that have not yet been described in the literature. Their structure is confirmed by microanalysis and by spectroscopic methods (see table). The carbonyl groups of the 8 give rise to peaks at 1610-1620 (C=O ketone) and 1720-1730 cm⁻¹ (C=O lactone) in the IR-spectra (KBr). The UV-spectra of the 2-coumaroylthiophenes 8 are very similar to those of corresponding 5-arylthiophenes 6 (Acc=arylCO) (2). A typical intense fragment peak of 173 (M⁺ - thienyl fragment) is found in the MS-spectra. The biological test of compound 8a revealed a significant inhibition of the angiotensin converting



enzyme ($IC_{50} = 18.10^{-6}$ mol).

EXPERIMENTAL

The mixture of 3-aminothioacrylamide hydroperchlorate **2** (0.01 mol), or 0.01 mol 3-hydroxythioacrylamide **3**, 2.67 g (0.01 mol) 3-bromoacetyl coumarin **7**, 2.02 g (0.01 mol) triethyl amine and 15 ml ethanol is refluxed for 3 minutes. After cooling to room temperature the product precipitates. It is filtered by suction and recrystallised.

REFERENCES

1. Part 16: A. Knoll and J. Liebscher, *Synthesis*, in press.
2. J. Liebscher, Berhanu Abegaz and Alemayehu Areda, *J. Prakt. Chem.*, **325**, 168 (1983).
3. J. Liebscher, Berhanu Abegaz and H. Hartmann, DDR-Patent, 201306 (1983); *C.A.*, **100**, 51590h (1984).
4. J. Liebscher and Alemayehu Areda, DDR-Patent, 202875 (1983); *C.A.*, **100**, 156489k (1984).
5. J. Liebscher and Berhanu Abegaz, *Synthesis*, 769 (1982).
6. J. Liebscher, DDR-Patent 206990 (1983); *C.A.*, **100**, 130698y (1984).
7. J. Liebscher, Berhanu Abegaz and H. Hartmann, DDR-Patent, 204086 (1983); *C.A.*, **100**, 209849m (1984).
8. P. Czerney and H. Hartmann, *J. Prakt. Chem.*, **325**, 551 (1983).