

ONE-POT, MILD AND EFFICIENT MULTICOMPONENT SYNTHESIS OF NOVEL VARIOUS *SPIRO*-NITROGEN HETEROCYCLE COMPOUNDS

Maryam Adlu* and Issa Yavari

Department of Chemistry, Science and Research Branch, Islamic Azad University, Tehran, Iran

(Received December 9, 2021; Revised September 11, 2022; Accepted September 13, 2022)

ABSTRACT. One-pot, mild and efficient synthesis of various *spiro*-nitrogen heterocycle compounds, based on the reaction of ninhydrin and 1,2-diamino-benzene, (*indenoquinoxalin*), with *N*-heterocycle compounds and dialkylacetylenedicarboxylates is described. Using this approach, various *spiro*-nitrogen heterocycle compounds at a temperature of 50-60 °C in acetonitrile solvent, can be obtained very high yields.

KEY WORDS: Ninhydrin, 1,2-Diaminobenzene, *spiro*-Nitrogen heterocycle compounds, *Indenoquinoxalin*, *N*-heterocycle compounds, Dialkylacetylenedicarboxylates

INTRODUCTION

Heterocyclic chemistry is one of the most complex branches of chemistry and heterocyclic compounds, the largest and most diverse family of chemical compounds. The heterocyclic compounds have a stable structure that cannot be easily hydrolyzed or polymerized [1-3]. The heterocyclic compounds are a broad category of ring compounds that contain one or more non-carbon atoms, including nitrogen, oxygen, sulfur, or phosphorus in their structure [4-6].

Heterocyclic compounds play a vital role in biological processes and are widely found in natural compounds [7-11]. The main source of these compounds is plants. Heterocyclic compounds are used in the pharmaceutical industry. Some vitamins, proteins, and hormones have a heterocyclic structure. Multi-component reactions have always been a beneficial way of synthesizing heterocyclic compounds [12-17]. The quinoline, isoquinoline and *indenoquinoxalin* skeleton compounds are often used for the design of many synthetic compounds with diverse pharmacological properties such as antimicrobial, cytotoxic, HIV protease inhibitor, anti-inflammatory, anti-cancer, antitumor, antimalarial and anti-viral activities [18-20]. Yavari and co-workers had reported a huge number studies, utilizing quinoline, isoquinoline, *indenoquinoxalin* and acetylenedicarboxylate esters [21-27].

Meanwhile, *indenoquinoxalin* is one of the most important compounds for the synthesis of many heterocyclic compounds using multiple reactions [28-34]. The extent and dynamics of this part of organic chemistry has made these compounds a special place. Hence, researchers are using new and modern techniques to produce heterocyclic compounds [35-44].

According to the introduction, in this paper, using a gentle, effective and one-pot method, we reported an effective synthesis of *spiro*-heterocyclic nitrogen compounds based on the reaction between ninhydrin and 1,2-diaminobenzene, (*indenoquinoxalin*), with heterocyclic nitrogen compounds and dialkylacetylenedicarboxylate. Based on this, various *spiro*-nitrogen heterocycle compounds at a temperature of 50-60 °C in acetonitrile solvent, can be obtained very good yields.

RESULTS AND DISCUSSION

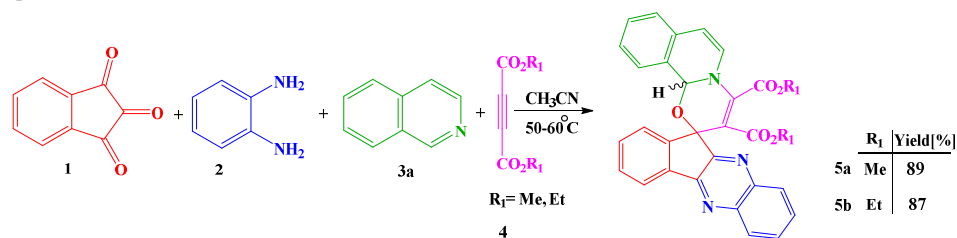
Initially, for the sample the reaction of ninhydrin **1** with 1,2-diaminobenzene **2** with isoquinoline **3** and dimethylacetylenedicarboxylate **4** was selected. In this case, solvent optimization and

*Corresponding author. E-mail: Maryam_adlu@yahoo.com

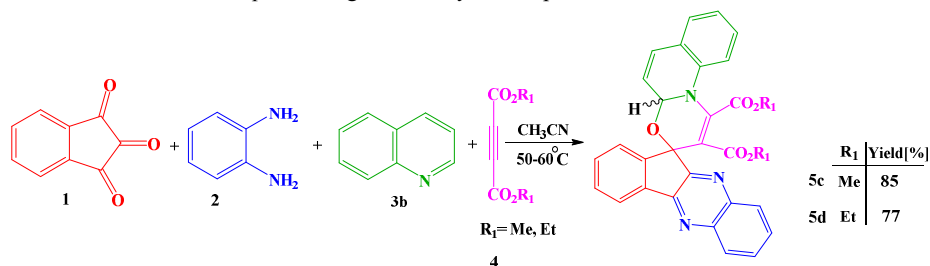
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reaction conditions were performed. Acetonitrile was found to be a suitable solvent and the best yield of **5a** was obtained at a temperature of 50-60 °C in acetonitrile. In the following, the reaction of ninhydrin **1** and 1,2-diaminobenzene **2** with nitrogen heterocycle compounds **3** and dialkylacetylenedicarboxylate **4** in an acetonitrile solvent is complete at 50-60 °C for 12 hours and *spiro*-heterocyclic nitrogen-containing compounds **5a-f** produce very good yields (Scheme 1-3).

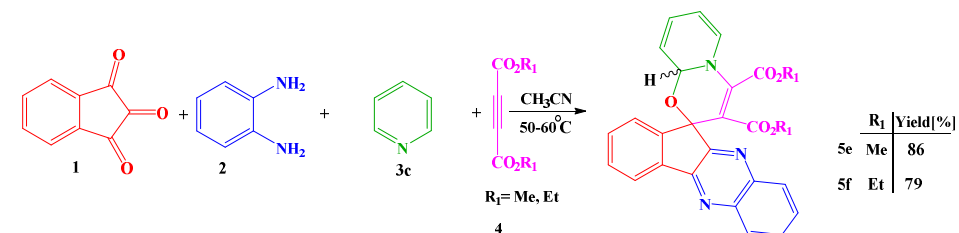
According to the results, products have two diastereomeric forms (60:40). But it is noteworthy that the reaction between dimethylacetylenedicarboxylate with pyridine or isoquinoline or quinoline and *N*-heterocycle compounds generates only one diastereoisomer, while with diethylacetylenedicarboxylate produce two diastereoisomers. Unfortunately, we could not separate these diastereoisomers (Scheme 4).



Scheme 1. Formation of *spiro*-nitrogen heterocycle compounds **5a-5b**.

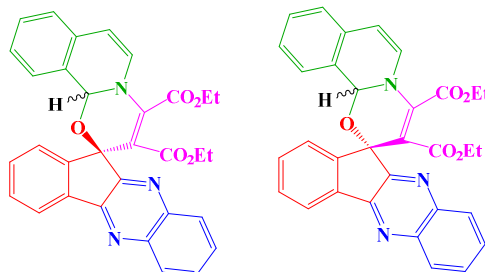


Scheme 2. Formation of *spiro*-nitrogen heterocycle compounds **5c-5d**.

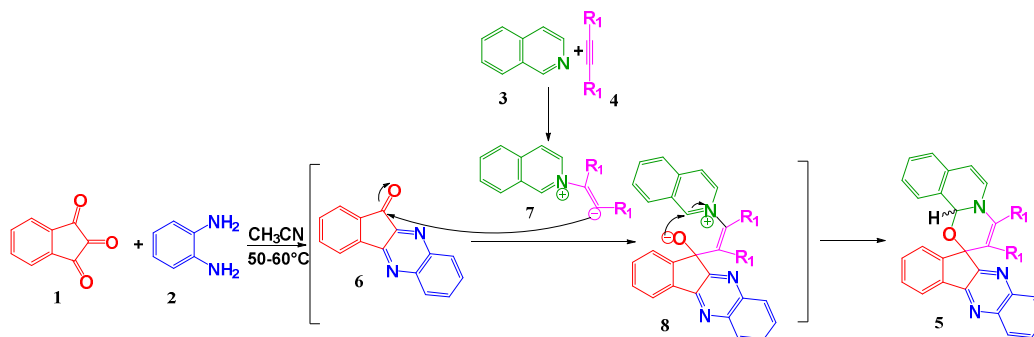


Scheme 3. Formation of *spiro*-nitrogen heterocycle compounds **5e-5f**.

The structure of the *spiro*-heterocyclic nitrogen compounds **5a-f** is deduced from the ¹H NMR, ¹³C NMR and IR spectra and elemental analysis. For example, the spectrum of the ¹H NMR composition of **5a** in CDCl₃ showed a singlet at δ 3.23 and a singlet at δ 4.07 for the methoxyprotons (2OMe), a doublet at δ 5.91 and a doublet at δ 6.57 for the methine proton (2=CH_{dihydroisoquinoline}), a singlet at δ 6.88 for the (CH_{diastereoisomer}) proton, multiplets at δ 7-8.40 for aryl protons(12H). The spectrum of the ¹³C NMR composition of **5a** in CDCl₃ showed 30 distinct resonances in agreement with the proposed structure.

Scheme 4. Two diastereoisomers of *spiro*-nitrogen heterocycle compounds **5b**, **5d** and **5f**.

The proposed mechanism for the synthesis of various *spiro*-nitrogen heterocycle compounds can be shown below (Scheme 5). Initially, in the reaction vessel, ninhydrin **1** reacted with 1,2-diaminobenzene **2** and produced an intermediate of the indenoquinoxalin **6**. At the same time, nitrogen isoquinoline **3** attacked the dialkylacetylenedicarboxylate **4** and produced an intermediate **7**. Then, the carbonyl group of intermediate **6** was attacked by carbon with a negative charge of intermediate **7** to furnish intermediate **8**. In the following, by forming a ring, the final product **5** is produced.

Scheme 5. A plausible mechanism for compound **5**.

EXPERIMENTAL

Chemicals and apparatus

Solvents, ninhydrin, 1,2-diaminobenzene, isoquinoline, quinoline, pyridine and dialkylacetylenedicarboxylate were obtained from Merck, Fluka, and Aldrich, and were used without further purification. Electrothermal-9100 was used to measure the melting temperature. Elemental analysis was performed with Heraeus CHN-O-Rapid. ^1H , ^{13}C NMR spectra were recorded with a Bruker DRX-500 Avance instrument using CDCl_3 as the deuterated solvent containing TMS as internal standard, at 300, 500.1, 125.8 MHz, respectively; δ in ppm, J in Hz. IR Spectra (v/cm^{-1}) were recorded as KBr pellets with a Shimadzu IR-460 spectrometer.

General procedure for the preparation of compounds 5a-f

To a mixture of (1 mmol) ninhydrin and (1 mmol) of 1,2-diaminobenzene in 10 mL of acetonitrile solvent, (1 mmol) isoquinoline was added and the solution was stirred. Then (1 mmol) of dialkylacetylenedicarboxylate in 2 mL of acetonitrile was dropwise added to the solution for 15 min at room temperature. Then the reaction mixture was allowed to warm to 50-60 °C and was stirred for 12 hours. The product was filtered and the solvent was removed under reduced pressure. The remaining residue was recrystallized and poured from ethanol.

Dimethyl 11b'H-spiro[indeno[1,2-b]quinoxaline-11,2'-[1,3]oxazino[2,3-a]isoquinoline]-3',4'-dicarboxylate (5a). Yellow powder; m.p. = 240; yield: 0.448 g (89%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1700, 1736 (2C=O); $^1\text{H NMR}$: δ 3.23 (3H, s, MeO), 4.07 (3H, s, MeO), 5.91 (1H, d, $^3J = 7.9$ Hz, =CH), 6.57 (1H, d, $^3J = 7.9$ Hz, =CH), 6.88 (1H, s, CH), 7-8.40 (12H, m, =CH); $^{13}\text{C NMR}$: δ 51.45, 53.41 (2OMe), 80.82 (CH), 105.12 (C_{spiro}), 122.72, 123.43, 123.97, 125.18, 126.37, 127.06, 127.78, 128.81, 129.07, 129.45, 129.73, 129.85, 130.10, 130.41 (14=CH), 132.33, 137.80, 141.36, 142.06, 145.50, 147.49, 154.23 (10C), 162.33, 163.90 (2C=O). Anal. calcd. for $\text{C}_{30}\text{H}_{21}\text{N}_3\text{O}_5$ (503.15): C, 71.56; H, 4.20; N, 8.35; O, 15.89%. Found: C, 71.54; H, 4.23; N, 8.32; O, 15.91%.

Diethyl 11b'H-spiro[indeno[1,2-b]quinoxaline-11,2'-[1,3]oxazino[2,3-a]isoquinoline]-3',4'-dicarboxylate (5b). Yellow powder; m.p. = 240; yield: 0.462 g (87%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1701, 1736 (2C=O); NMR data for the major isomer (60%): $^1\text{H NMR}$: δ 0.51 (3H, t, $^3J = 7.1$ Hz, Me), 1.49 (3H, t, $^3J = 7.1$ Hz, Me), 3.60 (2H, m, OCH_2), 4.53 (2H, m, OCH_2), 5.89 (1H, d, $^3J = 7.8$ Hz, =CH), 6.59 (1H, d, $^3J = 7.7$ Hz, =CH), 6.90 (1H, s, CH), 7-8.30 (12H, m, =CH); $^{13}\text{C NMR}$: δ 13.19 (Me), 13.96 (Me), 60.19 (OCH_2), 62.79 (OCH_2), 80.77 (CH), 104.79 (C_{spiro}), 122.33, 123.44, 124.09, 125.12, 126.45, 126.98, 127.82, 128.94, 129.08, 129.40, 129.84, 129.91, 130.28, 130.41 (14=CH), 131.81, 132.14, 137.12, 138.26, 141.32, 142.53, 145.75, 147.65, 149.01, 154.23 (10C), 163.19, 163.39 (2C=O); NMR data for the minor isomer (40%): $^1\text{H NMR}$: δ 0.57 (3H, t, $^3J = 7.1$ Hz, Me), 1.50 (3H, t, $^3J = 7.1$ Hz, Me), 3.69 (2H, m, OCH_2), 4.56 (2H, m, OCH_2), 5.91 (1H, d, $^3J = 7.8$ Hz, =CH), 6.59 (1H, d, $^3J = 7.7$ Hz, =CH), 6.90 (1H, s, CH), 7-8.30 (12H, m, =CH); $^{13}\text{C NMR}$: δ 13.11 (Me), 13.93 (Me), 60.24 (OCH_2), 62.79 (OCH_2), 80.23 (CH), 106.13 (C_{spiro}), 122.68, 123.91, 125.07, 125.21, 126.24, 126.96, 127.36, 128.79, 128.85, 129.46, 129.79, 129.86, 130.05, 130.12 (14=CH), 131.81, 132.14, 137.12, 138.26, 141.82, 142.38, 145.57, 147.65, 149.01, 154.23 (10C), 162.21, 162.42 (2C=O). Anal. calcd. for $\text{C}_{32}\text{H}_{25}\text{N}_3\text{O}_5$ (531.18): C, 72.31; H, 4.74; N, 7.91; O, 15.05%. Found: C, 72.28; H, 4.70; N, 7.93; O, 15.10%.

Dimethyl 4a'H-spiro[indeno[1,2-b]quinoxaline-11,3'-[1,3]oxazino[3,2-a]quinoline]-1',2'-dicarboxylate (5c). Yellow powder; m.p. = 240; yield: 0.427 g (85%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1707, 1713 (2C=O); $^1\text{H NMR}$: δ 3.27 (3H, s, OMe), 3.93 (3H, s, OMe), 5.95 (1H, dd, $^3J = 9.8$ Hz, =CH), 6.52 (1H, d, $^3J = 4.3$ Hz, =CH), 6.83 (1H, d, $^3J = 9.8$ Hz, CH), 7.05-8.16 (12H, m, =CH); $^{13}\text{C NMR}$: δ 52.02, 53.27 (2OMe), 81.03 (CH), 114.31 (C_{spiro}), 118.58, 121.79, 122.42, 122.45, 123.75, 128.65, 128.93, 129.22, 129.24, 129.27, 129.79, 129.84, 129.93 (14=CH), 130.42, 132.13, 135.89, 138.29, 141.41, 142.89, 143.67, 146.46, 154.01, 161.24 (10C), 164.02, 164.45 (2C=O). Anal. calcd. for $\text{C}_{30}\text{H}_{21}\text{N}_3\text{O}_5$ (503.15): C, 71.56; H, 4.20; N, 8.35; O, 15.89%. Found: C, 71.53; H, 4.25; N, 8.33; O, 15.88%.

Diethyl 4a'H-spiro[indeno[1,2-b]quinoxaline-11,3'-[1,3]oxazino[3,2-a]quinoline]-1',2'-dicarboxylate (5d). Yellow powder; m.p. = 240; yield: 0.409 g (77%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1726 (2C=O), NMR data for the major isomer (60%): $^1\text{H NMR}$: δ 0.60 (3H, t, $^3J = 7.1$ Hz, Me), 1.31 (3H, t, $^3J = 7.1$ Hz, Me), 3.72 (2H, m, OCH_2), 4.41 (2H, m, OCH_2), 5.95 (1H, dd, $^3J = 9.6$ Hz, =CH), 6.01 (1H, d, $^3J = 4.7$ Hz, =CH), 6.88 (1H, d, $^3J = 9.6$ Hz, CH), 7.02-8.20 (12H, m, =CH); $^{13}\text{C NMR}$: δ 13.18 (Me), 13.96 (Me), 60.20 (OCH_2), 62.78 (OCH_2), 81.03 (CH), 114.31 (C_{spiro}),

118.58, 121.79, 122.42, 122.45, 123.75, 128.65, 128.93, 129.22, 129.24, 129.27, 129.79, 129.84, 129.93 (14=CH), 130.42, 132.13, 135.89, 138.29, 141.41, 142.89, 143.67, 146.46, 154.01, 161.24 (10C), 164.02, 164.45 (2C=O); NMR data for the minor isomer (40%): ¹H NMR: δ 0.59 (3H, t, ³J = 7.1 Hz, Me), 1.31 (3H, t, ³J = 7.1 Hz, Me), 3.60 (2H, m, OCH₂), 4.32 (2H, m, OCH₂), 5.95 (1H, dd, ³J = 9.6 Hz, =CH), 6.01 (1H, d, ³J = 4.7 Hz, =CH), 6.88 (1H, d, ³J = 9.6 Hz, CH), 7.02-8.20 (12H, m, =CH); ¹³C NMR: δ 13.18 (Me), 13.96 (Me), 60.20 (OCH₂), 62.78 (OCH₂), 79.93 (CH), 114.31 (C_{spiro}), 118.58, 121.79, 122.42, 122.45, 123.75, 128.65, 128.93, 129.22, 129.24, 129.27, 129.79, 129.84, 129.93 (14=CH), 130.42, 132.13, 135.89, 138.29, 141.41, 142.89, 143.67, 146.46, 154.01, 161.24 (10C), 164.02, 164.45 (2C=O). Anal. calcd. for C₃₂H₂₅N₃O₅ (531.18): C, 72.31; H, 4.74; N, 7.91; O, 15.05%. Found: C, 72.35; H, 4.73; N, 7.89; O, 15.07%.

Dimethyl 9a'H-spiro[indeno[1,2-b]quinoxaline-11,2'-pyrido[2,1-b][1,3]oxazine]-3',4'-dicarboxylate (5e). Yellow powder; m.p. = 192 °C; yield: 0.390 g (86%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1737 (2C=O), 1259; ¹H NMR: δ 1.15 (6H, s, 2MeO), 5.32 (1H, dd, ³J = 6.7 Hz, =CH), 5.49 (1H, dd, ³J = 9.9 Hz, =CH), 6.20 (1H, dd, ³J = 9.9 Hz, =CH), 6.45 (1H, d, ³J = 7.5 Hz, =CH), 6.96 (1H, dd, ³J = 3.2 Hz, CH), 7.42-8.17 (8H, m, =CH); ¹³C NMR: δ 53.42, 57.99 (2OMe), 79.37 (CH), 101.69 (C_{spiro}), 107.28, 116.48, 122.42, 123.66, 124.84, 125.11, 128.93, 129.04, 129.73, 129.91, 130.34 (12=CH), 132.09, 138.28, 141.21, 142.58, 145.50, 147.28, 153.91, 161.70 (8C), 163.54, 163.77 (2C=O). Anal. calcd. for C₂₆H₁₉N₃O₅ (453.45): C, 68.87; H, 4.22; N, 9.27; O, 17.64%. Found: C, 68.79; H, 4.23; N, 9.30; O, 17.66%.

Diethyl 9a'H-spiro[indeno[1,2-b]quinoxaline-11,2'-pyrido[2,1-b][1,3]oxazine]-3',4'-dicarboxylate (5f). Yellow powder; m.p. = 198 °C; yield: 0.380 g (79%); IR (KBr) ($\nu_{\max}/\text{cm}^{-1}$) = 1695, 1734 (2C=O); NMR data for the major isomer (60%): ¹H NMR: δ 0.50 (3H, t, ³J = 7.1 Hz, Me), 1.44 (3H, t, ³J = 7.1 Hz, Me), 3.59 (2H, m, OCH₂), 4.48 (2H, m, OCH₂), 5.35 (1H, dd, ³J = 10 Hz, =CH), 5.54 (1H, dd, ³J = 9.9 Hz, =CH), 6.24 (1H, dd, ³J = 9.9 Hz, =CH), 6.51 (1H, d, ³J = 7.6 Hz, =CH), 7.03 (1H, dd, ³J = 3.2 Hz, CH), 7.50-8.16 (8H, m, =CH); ¹³C NMR: δ 13.15 (Me), 13.96 (Me), 60.23 (OCH₂), 62.78 (OCH₂), 79.44 (CH), 101.38 (C_{spiro}), 107.19, 116.47, 122.27, 123.81, 124.84, 125.11, 128.84, 129.16, 129.78, 130.23 (12=CH), 132.04, 138.53, 141.24, 142.69, 145.76, 147.56, 154.12, 161.91 (8C), 163.08, 163.35 (2C=O); NMR data for the minor isomer (40%): ¹H NMR: δ 0.50 (3H, t, ³J = 7.1 Hz, Me), 1.44 (3H, t, ³J = 7.1 Hz, Me), 3.60 (2H, m, OCH₂), 4.49 (2H, m, OCH₂), 5.35 (1H, dd, ³J = 10 Hz, =CH), 5.54 (1H, dd, ³J = 9.9 Hz, =CH), 6.24 (1H, dd, ³J = 9.9 Hz, =CH), 6.51 (1H, d, ³J = 7.6 Hz, =CH), 7.03 (1H, dd, ³J = 3.2 Hz, CH), 7.50-8.16 (8H, m, =CH); ¹³C NMR: δ 13.15 (Me), 13.96 (Me), 60.23 (OCH₂), 62.78 (OCH₂), 79.99 (CH), 101.38 (C_{spiro}), 107.19, 116.47, 122.27, 123.81, 124.84, 125.11, 128.84, 129.16, 129.78, 130.23 (12=CH), 132.04, 138.53, 141.24, 142.69, 145.76, 147.56, 154.12, 161.91 (8C), 163.08, 163.35 (2C=O). Anal. calcd. for C₂₈H₂₃N₃O₅ (481.51): C, 69.84; H, 4.81; N, 8.73; O, 16.61%. Found: C, 69.80; H, 4.83; N, 8.75; O, 16.60%.

CONCLUSIONS

In this work, one-pot, mild and efficient synthesis of various *spiro*-nitrogen heterocycle compounds based on the reaction of ninhydrin and 1,2-diaminobenzene with isoquinoline or quinoline or pyridine and dialkylacetylenedicarboxylate in CH₃CN at 50-60 °C is described. Using this approach, all *spiro*-nitrogen heterocycle compounds are obtained in high yields. The method offers several advantages including high diversity via various functional groups, operational simplicity and high yields.

ACKNOWLEDGEMENTS

We gratefully acknowledge the support of this work by Islamic Azad University, Science and Research Branch, Tehran, Iran.

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