REACTION OF PROPARGYLTRIMETHYLSILANE MAGNESIUM BROMIDE WITH ALDIMINES: SYNTHESIS OF 1-(ALKYLAMINO)-2-(TRIMETHYLSILYL)BUT-3-YNE

L. Mbaze Meva'a*; J.C. Ndom and E. Ngeufa Happi

University of Douala, Faculty of Science, Department of Chemistry, P.O. Box 24157, Douala, Cameroon

(Received February 22, 2001; revised May 10, 2002)

ABSTRACT. The reaction of propargylsilane magnesium bromide with aldimines provides in some cases, an isomeric mixture of 3 acetylenic amines 1, 2, 3 (1, 2, as major products). The reaction has great potential and has been exploited for an efficient synthesis of the pure amine 2.

KEY WORDS: Propargylsilane magnesium bromide, Aldimines, 1-(alkylamino)-2-(trimethylsilyl)but-3-yne, Acetylenic amines,

INTRODUCTION

Considerable attention has been given over the past few years to the use of propargyltrimethylsilane magnesium bromide as a synthetic reagent, such as: (a) in the study of the methylisosartotuate synthesis [1]; (b) in the reaction with carbonyl derivatives [2-4] for the preparation of 4-(trimethylsilyl)but-2-ynol, $Me_3SiCH_2-C\equiv C-CRR'(OH)$; (c) in the treatment with epoxide [5] in order to obtain 5-(trimethylsilyl)pent-3-ynol, $Me_3SiCH_2-C\equiv C-CH_2CH_2OH$; and (d) in the reaction with trimethylchlorosilane [6] to give 1,3-(bistrimethylsilyl)prop-1-yne, $Me_3SiCH_2-C\equiv C-SiMe_3$.

In continuation of the work in this field, we wanted to obtain a convenient preparation of 1. We obviously investigated the reaction of propargylsilane magnesium bromide with aldimines. The reaction is more complex as expected and gives a mixture of three products 1, 2, 3 after hydrolysis.

The amine 2 is the major product according to the experience. This paper describes the general method and the limitations for the preparation of this product.

RESULTS AND DISCUSSIONS

 $[\]hbox{*Corresponding author. E-mail: lmbazze@yahoo.fr}$

Reaction of propargylsilane magnesium bromide with aldimines

Propargylsilane magnesium bromide was treated with two aliphatic aldimines and four benzaldimines. The results are summarised in Table 1. According to these results, in some cases an isomeric mixture of $\mathbf{1}$, $\mathbf{2}$, and $\mathbf{3}$ is obtained. Only traces of $\mathbf{3}$ were observed with the aldimine $CH_2=NC(CH_3)_3$, $(R=H, R'=C(CH_3)_3)$.

Aldimine RCH=NR'		Condition*	Amine	Yield (%)
R	R'			
			1a	35
Н	Me ₃ C	(a)	2a	39
			3a	<2
Н	Me ₃ C	(b)	2a	53
iPr			1b	6
	Me	(a)	2b	11
		(c)	1b	12
iPr	nBu	(a)	2c	10
C ₆ H ₅	Me	(a)	2d	20
C ₆ H ₅	Et	(a)	2e	75
C_6H_5	nPr	(a)	2f	70**
C.H.	nRu	(a)	2α	54**

Table 1. Reaction of Me₃SiCH₂C≡C-MgBr/Et₂O with RCH=NR'.

*(a): mol propargyltrimethylsilane/mole ethylbromide magnesium/mol aldimine = 1/1.4/1.2. The reaction was carried out at room temperature for 16 hours. (b): mol propargyltrimethylsilane/mol ethylbromide magnesium/mol aldimine = 1/1.4/1.2. The reaction was carried out at room temperature for 16 hours. Then heated to reflux for 16 hours. (c): same as condition (a) but 1/1/1.** Traces of 4 RCH(C₂H₅)NHR', yield < 3%.

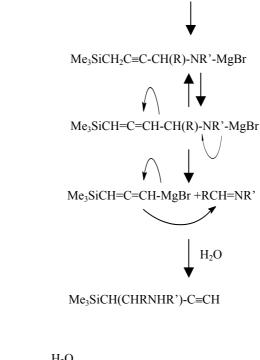
For the aliphatic aldimines the yields are relatively weak probably because of enamine formation [7]. The benzaldimines give pure 2 in good yields without heating. An excess of EtMgBr is usually necessary for better results (conditions (a) and (b)). The presence of the excess of EtMgBr seems to be important for the obtention of pure 2. Heating may accelerate the reaction (condition (b)). Formation of 2 can be explained by the isomerisation of the expected amine 1 into 2 under the influence of the excess of EtMgBr. This involves deprotonation of 1 with EtMgBr, then conversion of the metallated acetylenic amine into its allenic isomer. The metallated allenic amine rearrangement liberates the aldimine and an allenic magnesium. The reaction of the obtained allenic magnesium and the aldimine affords 2 (Scheme 1).

The aldimine also reacts with an excess of EtMgBr to give after hydrolysis small amounts of the amine $R(C_2H_5)$ CHNHR' 4 which have been isolated for heavy R' susbtituents (Scheme 2).

In order to confirm the proposed mechanism, we have prepared allenic amine Me₃SiCH=C=CH-CH₂NHC(CH₃)₃ **5** from reaction of **2a** with polyoxymethylene in acetonitrile in the presence of trifluoroacetic acid according to [8]. The treatment of **5** with excess of EtMgBr lead to pure **2a** (yield 24%).

In conclusion, we developed an efficient route for the preparation of pure propargylsilanes 2 with amine function; these compounds may be useful intermediates for the synthesis of new unsaturated amines via reduction reaction [9] and amino methylation [10-12].

 $Me_3SiCH_2C\equiv C-MgBr + RCH=NR'$



Scheme 1

RCH=NR' +
$$C_2H_5MgBr$$
 \longrightarrow $R(C_2H_5)CHNHR'$ 4
Scheme 2

EXPERIMENTAL

IR spectra were recorded on an IR 4220 Beckman, the frequencies are given in cm $^{-1}$. 1 H and 13 C NMR spectra were recorded on a JEOL JNM-EX 90 MHz spectrometer at 90 MHz and 22.5 MHz, respectively, in CDCl $_{3}$, unless otherwise stated, chemical shifts are given in δ units. Some of 1 H NMR spectra were obtained on a Perkin Elmer R 24A spectrometer at 60 MHz in CCl $_{4}$. Mass spectra were obtained on a GC 8000-MS FISONS TRIO 1000 at 70 eV.

Reaction of propargyltrimethylsilanemagnesium bromide with aldimines

Propargylsilane was prepared according to [13]. The aldimines were prepared according to [14]. In a three necked flask equipped with mechanical stirrer, propagyltrimethylsilane 0.048 mole (5 g) in 10 mL of anhydrous ethylether was added to 0.007 mole of ethylmagnesiumbromide. The reaction mixture was heated at 35 °C under nitrogen for 3 hours. A solution of 0.05 mole of aldimine in 10 mL of anhydrous ethylether was added dropwise at room temperature and the mixture was stirred at this temperature for 16 hours. The reaction mixture was treated cold 200 mL of saturated NH₄Cl solution. After separation of the layers, the aqueous phase was extracted with ethylether (3 x 25 mL). The resulting organic phase was treated with HCl 2 M (2 x 25 mL) and extracted with ethylether (3 x 25 mL). The organic phase was washed with water (3 x 25 mL). The resulting cooled aqueous layers were made basic by addition of NaOH 2 M

(75 mL) and extracted with ethylether (3 x 25 mL). The combined organic layers were washed with water, dried over potassium carbonate and concentrated. The residue is distilled to give the amine 2. For mixtures of products the concentrated organic phases was distilled bulb-to-bulb into a liquid nitrogen cooled receiver and separation is made by preparative GLC (30% SE 30 on chromosorb W columns)

$Me_3SiCH_2C \equiv C-CH(R)NHR'$ 1

Ia: (R = H, $R' = C(CH_3)_3$) 4-(Tertbutylamino)-1-(trimethylsilyl)but-2-yne. $n_D^{20} = 1.4545$, b.p. 64 °C/0.1 torr. IR (cm⁻¹): 3310 w (NH); 2220 w (C≡C); 1410 w (CN); 1360 m (C(CH₃)₃); 1250 s 840 s, 755 m (SiMe₃). ¹H NMR (CDCl₃) δ: 0.10 (s, 9H, SiMe₃); 0.85 (s, 1H, NH); 1.05 (s, 9H, CH₃); 1.40(t, J = 2.4Hz, 2H, CH₂); 3.25 (t, J = 2.4Hz, 2H, CH₂N). ¹³C NMR (CDCl₃) δ: −2.05 (SiMe₃); 6.75 (SiCH₂); 28.89 (CH₃); 50.25 (C); 51.95 (CH₂-N); 72.23 (SiCH₂C≡); 82.79 (≡CCH(R)). HRMS (EI): calculated for C₁₁H₂₃NSi: 197.16009, found: 197.1607.

Ib: (R = iPr, $R' = CH_3$) 1-(Methylamino)-1-(ipropyl)-4-(trimethylsilyl)but-2-yne. $n_D^{\ 20} = 1.4549$. IR (cm⁻¹): 3320 w (NH); 2195 w (C≡C); 1410 w (CN); 1250 s, 840 s, 755 m (SiMe₃).

¹H NMR (CDCl₃) δ: 0.10 (s, 9H, SiMe₃); 0.90 (d, J = 6.4 Hz, 6H, CH₃); 1.05 (s, 1H, NH); 1.40 (d, J = 2.3 Hz, 2H, CH₂); 1.45-1.90 (m, 1H, CH); 2.30 (s, 3H, NCH₃); 2.85 (dt, J = 5.2 Hz, J = 2.3 Hz, 1H, ≡C-CH).

¹³C NMR (CDCl₃) δ: −2.10 (SiMe₃); 6.70 (SiCH₂); 21.89 (CH₃); 22.01 (CH₃); 32.15 (CH); 37.01 (NCH₃); 70.10 (CHN); 72.55 (SiCH₂C≡); 82.75 (≡CCH(R)). HRMS (EI): calculated for $C_{11}H_{23}NSi$: 197.16009, found: 197.1609.

Me₃SiCH(CHRNHR')-C≡CH 2

2a: $(R = H, R' = C(CH_3)_3)$ 1-(Tertbutylamino)-2-(trimethylsilyl)but-3-yne. $n_D^{20} = 1.4549$, b.p. 80 °C/12 torr. IR (cm⁻¹): 3310 s, 2100 s (C \equiv CH); 1405 w (CN); 1360 s (C(CH₃)₃); 1245 s, 835 s, 750 m (SiMe₃). ¹H NMR (CDCl₃) δ : 0.04 (s, 9H, SiMe₃); 0.94 (s, 1H, NH); 1.03 (s, 9H, CH₃); 1.68-1.87 (m, 1H, CH); 1.94 (d, J = 2.8 Hz, 1H, \equiv CH); 2.54-2.61 (m, 2H, CH₂). ¹³C NMR (CDCl₃) δ : -3.08 (SiMe₃); 21.75 (CH); 28.94 (CH₃); 42.12 (CH₂); 50.13 (C); 69.27 (\equiv CH); 85.35 (C \equiv). HRMS (EI): calculated for C₁₁H₂₃NSi: 197.16009, found: 197.1611.

2b: $(R = iPr, R' = CH_3)$ 1-(Methylamino)-1-(isopropyl)-2-(trimethylsilyl)but-3-yne. IR (cm⁻¹): 3320 s, 2100 s (C=CH); 1405 w (CN); 1250 s, 840 s, 750 m (SiMe₃). ¹H NMR (CCl₄) δ : 0.10 (s, 9H, SiMe₃); 0.85 (d, J = 6.2 Hz, 3H); 0.87 (d, J = 6.3 Hz, 3H, CH₃); 1.35 (s, 1H, NH); 1.40-1.95 (m, 2H, CH); 1.95-2.45 (m, 5H, CHNCH₃, =CH). ¹³C NMR (CDCl₃) δ : -2.30 (SiMe₃); 18.12 and 19.38 (CH₃); 22.54 (Si-CH); 32.15 (CH); 37.15 (NCH₃); 61.05 (CHN); 70.75 (=CH); 84.05 (C=). HRMS (EI): calculated for C₁₁H₂₃NSi: 197.16009, found: 197.1604.

2c: (R = iPr, R' = nBu) 1-(Butylamino)-1-(isopropyl)-2-(trimethylsilyl)but-3-yne. IR (cm⁻¹): 3315 s; 2095 s (C≡CH); 1405 w (CN); 1250 s, 840 s, 750 m (SiMe₃). ¹H NMR (CDCl₃) δ: 0.10 (s, 9H, SiMe₃); 0.80-1.05 (m, 9H, CH₃); 1.20-1.50 (m, 6H, NH, CHCH₃, CH₂); 1.80-2.05 (m, 2H, CH, ≡CH); 2.25-2.65 (m, 3H, CHNCH₂). ¹³C NMR (CDCl₃) δ: −2.33 (SiMe₃); 14.02, 19.48 (CH₃); 20.52 (CH₂CH₃); 22.64 (CH); 32.25 (CHCH₃); 32.99 (CH₂); 48.06 (CH₂N); 61.10 (CHN); 70.50 (≡CH); 84.70 (C≡). HRMS (EI): calculated for C₁₄H₂₉NSi: 239.20707, found: 239.2076.

2d: $(R = C_6H_5, R' = CH_3)$ 1-(Methylamino)-2-(phenyl)-2-(trimethylsilyl)but-3-yne. b.p. 105 °C/0.1 torr. IR (cm⁻¹): 3320 s, 2105 s (C≡CH); 3035 m, 1605 m, 1495 s (C₆H₅); 1410 w

(CN); 1250 s, 850 s, 750 m (SiMe₃). ¹H NMR (CCl₄) δ : 0.10 (s, 9H, SiMe₃); 1.85 (s, 1H, NH); 2.00-2.20 (m, 2H, CH, \equiv CH); 2.25 (s, 3H, CH₃); 3.60 (d, J = 7.3 Hz, 1H, CHN); 7.25 (s, 5H, C₆H₅). ¹³C NMR (CDCl₃) δ : -2.40 (SiMe₃); 29.79 (Si-CH); 61.04 (CH₃); 62.28 (CH); 71.20 (\equiv CH); 84.21 (C \equiv); 127.29, 127.51, 128.12, 143.15 (C₆H₅). HRMS (EI): calculated for C₁₄H₂₁NSi: 231.1444, found: 231.1451.

2e: $(R = C_6H_5, R' = Et) \ 1$ -(Ethylamino)-1-(phenyl)-2-(trimethylsilyl)but-3-yne. $n_D^{20} = 1.5071$, b.p. 90 °C/0.05 torr. IR (cm⁻¹): 3320 s, 2105 s (C=CH); 3035 m, 1605 m, 1495 s (C_6H_5); 1410 w (CN); 1250 s, 850 s, 750 m (SiMe₃). ¹H NMR (CDCl₃) δ : -0.04 (s, 9H, SiMe₃); 1.05 (t, J = 7.1 Hz, 3H,CH₃); 1.96 (s, 1H, NH); 2.03-2.17 (m, 2H, CH, =CH); 2.40 (q, J = 7.1 Hz, 2H, CH₂); 3.73 (d, J = 7.7 Hz, 1H, CHN); 7.30 (s, 5H, C_6H_5). ¹³C NMR (CDCl₃) δ : -2.45 (SiMe₃); 15.12 (CH₃); 29.86 (CH); 41.44 (CH₂); 62.38 (CHCN); 71.27 (=CH); 84.31 (C=); 127.27, 127.51, 128.17, 143.14 (C_6H_5). HRMS (EI): calculated for $C_{15}H_{23}NSi$: 245.16009, found: 245.1607.

2*f*: ($R = C_6H_5$, R' = Pr) 1-(Propylamino)-1-(phenyl)-2-(trimethylsilyl)but-3-yne. $n_D^{20} = 1.5060$, b.p. 118 °C/1 torr. IR (cm⁻¹): 3310 s, 2100 s (C≡CH); 3025 m, 1600 m, 1490 s (C₆H₅); 1410 w (CN); 1250 s, 845 s, 760 m (SiMe₃). ¹H NMR (CDCl₃) δ : -0.02 (s, 9H, SiMe₃); 0.89 (t, J = 7.2 Hz, 3H, CH₃); 1.46 (m, 2H, CH₂); 2.00 (s, 1H, NH); 2.01-2.17 (m, 2H, CH, ≡CH); 2.35 (t, J = 6.9 Hz, 2H, CH₂N); 3.73 (d, J = 6.9 Hz, 1H, CHN); 7.31 (s, 5H, C₆H₅). ¹³C NMR (CDCl₃) δ : -2.42 (SiMe₃); 11.78 (CH₃); 23.09 (CH₂); 30.01 (CH); 49.13 (CH₂N); 62.41 (CHCN); 71.30 (≡CH); 84.28 (C≡); 127.21, 127.51, 128.14, 143.29 (C₆H₅). HRMS (EI): calculated for C₁₆H₂₅NSi: 259.17575, found: 259.1765.

2g: ($R = C_6H_5$, R' = nBu) 1-(Butylamino)-1-(phenyl)-2-(trimethylsilyl)but-3-yne. $n_D^{20} = 1.5031$, b.p. 108 °C/0.06 torr. IR (cm⁻¹): 3315 s, 2100 s (C≡CH); 3030 m, 1605 m, 1495 s (C₆H₅); 1410 w (CN); 1250 s, 850 s, 760 m (SiMe₃). ¹H NMR (CDCl₃) δ: -0.02 (s, 9H, SiMe₃); 0.88 (t, J = 6.4 Hz, 3H, CH₃); 1.15-1.55 (m, 4H, CH₂); 2.02 (s, 1H, NH); 2.03-2.16 (m, 2H, CH, ≡CH); 2.37 (t, J = 6.5 Hz, 2H, CH₂N); 3.72 (d, J 7.1 Hz, 1H,CHN); 7.31 (s, 5H, C₆H₅). ¹³C NMR (CDCl₃) δ: −2.42 (SiMe₃); 13.90 (CH₃); 20.40 (<u>C</u>H₂CH₃); 30.01 (CH); 32.19 (CH₂); 49.93 (CH₂N); 62.53 (CHN); 71.33 (≡CH); 84.31 (C≡); 127.24, 127.51, 128.17, 143.32 (C₆H₅). HRMS (EI): calculated for C₁₇H₂₇NSi: 273.19141, found: 273.1919.

$Me_3Si-C \equiv C-CH_2-CH(R)NHR'$ 3

3a: $(R = H, R' = C(CH_3)_3)$ 4-(tert-Butylamino)-1-(trimethylsilyl)but-1-yne. IR (cm⁻¹): 3310 w (NH); 2185 s (C≡C); 1405 w (CN); 1365 m (C(CH₃)₃); 1250 s, 840 s, 755 m (SiMe₃). ¹H NMR (CCl₄) δ: -0.03 (s, 9H, SiMe₃); 1.05 (s, 9H, CH₃); 1.10 (s, 1H, NH); 2.10-2.40 (m, 2H, CH₂); 2.50-2.80 (m, 2H, CH₂N). ¹³C NMR (CDCl₃) δ: -0.03 (SiMe₃); 6.71 (CH₂); 29.01 (CH₃); 50.21 (C); 51.12 (CH₂-N); 83.22 (SiC≡); 92.48 (≡C). HRMS (EI): calculated for C₁₁H₂₃NSi: 197.16009, found: 197.1618.

$R(C_2H_5)CHNHR'4$

4f: ($R = C_6H_5$, R' = Pr) N-(1-Phenylpropyl)propylamine. IR (cm⁻¹): 3320 w (NH); 3030 m, 1605 m, 1495 s (C_6H_5); 1405 w (CN). ¹H NMR (CCl₄) δ: 0.60-1.05 (m, 7H, CH₃, NH)); 1.10-1.80 (m, 4H, CH₂); 2.30 (t, J = 6.3 Hz, 2H, CH₂N); 3.35 (t, J = 6.4 Hz, 1H, CH); 7.10 (s, 5H, C_6H_5). **4g**: ($R = C_6H_5$, R' = nBu) N-(1-Phenylbutylpropyl)butylamine. IR (cm⁻¹): 3325 w (NH); 3030 m, 1605 m, 1495 s (C_6H_5); 1405 w (CN). ¹H NMR (CCl₄) δ: 0.55-1.05 (m, 7H, CH₃, NH); 1.08-

1.82 (m, 6H, CH_2); 2.32 (t, J = 6.2 Hz, 2H, CH_2N); 3.38 (t, J = 6.2 Hz, 1H, CH); 7.10 (s, 5H, C_6H_5).

 $Me_3SiCH=C=CH-CH_2-NHC(CH_3)_3$ 5. $n_D^{20}=1.5060$. IR (cm⁻¹): 3310 w (NH); 1935 s, 845 s, (CH=C=CH); 1405 w (CN); 1360 m (C(CH₃)₃; 1245 s,835 s,755 m (SiMe₃). ¹H NMR (CCl₄) δ: 0.130 (s, 9H, SiMe₃); 0.85 (s, 1H, NH); 1.05 (s, 9H, CH₃); 2.95-3.20 (m, 2H, CH₂); 4.70-4.95 (m, 2H, CH=). ¹³C NMR (CDCl₃) δ: -1.15 (SiMe₃); 15.25 (CH₃); 47.55 (CH₂); 50.51 (C); 82.62 (CH=); 83.43 (SiCH=); 209.45(=C=). HRMS (EI): calculated for $C_{11}H_{23}NSi$: 197.16009, found: 197.1615.

ACKNOWLEDGEMENT

We are grateful to Professor J. Pornet of the University of Poitiers (France) for determining the NMR data of the synthetic compounds. We thank the laboratory of organic synthesis of the University of Poitiers for financial support and TWA (Academy of Science) for reserarch grants RG/CHE/AF/AC.

REFERENCES

- 1. Xibin, L.; Xingxiang, X. Tretrahedron Lett. 2000, 41, 4641.
- 2. Pornet, J.; Randrianoelina, B.; Miginiac, L. Tretrahedron Lett. 1984, 25,651.
- 3. Pornet, J.; Miginiac, L.; Jaworski, K.; Randrianoelina, B. Organometallics 1985, 4, 333.
- 4. Pornet, J.; Damour, D.; Randrianoelina, B.; Miginiac, L. Tretrahedron 1986, 42, 2501.
- 5. Pornet, J.; Damour, D.; Miginiac, L. Tretrahedron 1986, 42, 2017.
- 6. Pornet, J.; Mesnard, D.; Miginiac, L. Tretrahedron Lett. 1982, 23, 4083.
- 7. Stork, G.; Dowd, S.R. J. Am. Chem. Soc. 1963, 85, 2178.
- 8. Mbaze Meva'a, L.; Pornet, J. Synth. Commum. 1996, 18, 3352.
- 9. Cram, D.J.; Allinger, N.L. J. Am. Chem. Soc. 1956, 78, 2518.
- Damour, D.; Pornet, J.; Randrianoelina, B.; Miginiac, L. J Organomet. Chem. 1990, 396, 289.
- 11. Damour, D.; Pornet, J.; Miginiac, L. Tetrahedron.Lett. 1987, 28, 4689.
- 12. Mbaze Meva'a, L.; Pornet, J. Synth. Commun. 1996, 18, 3351.
- 13. Pornet, J.; Jaworski, K.; N'kolani; Mesnard, D.; Miginiac, L. J. Organometallics. Chem. 1982, 2356, 177.
- 14. Tiolais, R. Bull. Soc. Chim. Fr. 1947, 708.