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EFFECT OF IRRIGATION ON HEAVY METALS CONTENT OF WASTEWATER IRRIGATED FLUVISOLS ALONG RIVER TATSEWARKI, KANO, NIGERIA

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ABSTRACT

There is an urgent need to educate farmers on the dangers of the presence of heavy metals in soils as well as the quality of irrigation water especially if it comes from tanning industries for increased crop production. Accordingly, soil and irrigation wastewater study was conducted to assess the concentrations of heavy metals in wastewater irrigated Fluvisols and in the irrigation wastewater itself as well as the relationship between the two. Composite surface soil and water samples including controls were collected and analyzed using standard methods. Analytical results were compared with the controls, the EU and FAO standards. Results indicated that the concentrations of Cu (30.556 to 41.667 mgkg⁻¹), Cd (10.269 to 18.687 mgkg⁻¹), Zn (18.687 to 26.010 mgkg⁻¹), Cr (8.848 to 12.14 mgkg⁻¹) and Pb (3.953 to 4.787 mgkg⁻¹) in the soil were non-toxic across the river sections. Results also indicated that the irrigation water contained toxic levels of Cu (0.85 to 1.25 mg l⁻¹), Cd (0.36 to 0.72 mg l⁻¹) and Cr (0.47 to 0.67 mg l⁻¹) and is therefore considered unsafe for use. However, it did not contain toxic levels of Zn (0.58 to 1.06 mg l⁻¹) and Pb (0.22 to 0.25 mg l⁻¹) and it is considered safe for use accordingly. Results also revealed that the irrigation water was responsible for the accumulation of Cr (r = 0.019), Cu (r = 0.151), Cd (r = 0.190) and Pb (r = 0.202) in the irrigated soils but not Zn (r = - 0.214). It is recommended that periodic appraisal of the soils should be embarked upon so as to monitor the toxicity level of the soil to keep it within the present non-toxic level. It is also recommended that the wastewater should be remediated to minimize and control the heavy metal contamination through phytoremediation. These measures are expected to enhance increased and sustainable crop production in the study area.

Key words: Fluvisols, Heavy metals, Irrigation, Wastewater.

INTRODUCTION

The term heavy metal refers to any metallic chemical element that has a relatively high density and is toxic or poisonous even at low concentrations (Lentech, 2012). Heavy metals occur naturally and they come from many different sources like, mining industries, burning of fossil fuels like coal, burning garbage or tobacco, forest fires, textile and tanneries (C.R.D., 2004). Heavy metals are currently of much environmental concern being harmful to humans, animals and are susceptible to bio-accumulation in the food chain. Heavy metals are known to be non-biodegradable, and they persist for long duration in aquatic as well as terrestrial environment, which also served as a major contributor to heavy metals contamination in top soil (Kelly *et al.*, 1996). Heavy metals such as Pb, As, Hg, Cd and Al are among the most dangerous toxins in our ecosystem. They might be transported from soil to ground waters which may be used for irrigation purposes and may be taken up by plants including agricultural crops. Therefore the knowledge of metal – soils and plant interaction is very important for the safety of the environment. Heavy metals pollution of agricultural soils is one of the most land severe ecological problems faced worldwide (Shukry, 2001). The direct discharge of effluents from industries into bodies of water has

become a growing environmental problem (Akan *et al.*, 2007). Most of these waste waters are extremely complex mixtures containing organic and inorganic compounds (Fu *et al.*, 1994). Farmers are ignorant about the hidden toxicity of heavily polluted discharges and their subsequent negative impact as a result of using contaminated water for their crops (Nath *et al.*, 2005). High concentrations of heavy metals in irrigation water can result in death of crops, interfere with uptakes of other essential nutrients or for objectionable deposits on fruits and render edible portion of plants toxic to human and grazing animals, (Aikman, 1983). Presence of pollutants in water alters different physico-chemical parameters from their normal prescribed levels (Mishra and Pandey, 2005). Tanning industrial wastes rank among the most polluting of all industrial wastes and are a serious threat when they pollute stream and fresh water bodies (Javaid, 2000). Consequently, the tanning industry is a potentially pollution intensive industry. Some heavy metals contained in these effluents (either in free form in the effluents or adsorbed in the suspended solid) from the industries have been found to be carcinogenic while other chemicals equally present are poisonous depending on the dose and exposure duration (Kupechella and Hyland, 1986).

It has been reported that only about 20% of the large number of chemicals used in the tanning process is absorbed by leather, the rest is released as waste (Kupechella and Hyland, 1986).

A number of studies are conducted in order to draw the attention of farmers to heavy metals accumulation in soils and plants (Shallari *et al.*, 1998, Jamali *et al.*, 2009). Micó *et al.*, (2006) reported mean values (mg kg^{-1}) of heavy metals at the Segura River Valley, Alicante, Spain, as Cd, 0.38; Co, 7.9; Cr, 28.3; Cu, 21.6; Fe, 15,274; Mn, 320; Ni, 23.7; Pb, 19.6; and Zn, 57.8. These values followed the sequence: Fe > Mn > Zn > Cr > Ni > Cu > Pb > Co > Cd. Omar (2010) reported Zn and Cu range of 2.99 to 4.43 and 0.75 to 1.42 mg/kg , respectively for Fluvisols in southwestern Bauchi State, Nigeria. These levels were considered to be adequate for crop production. Fluvisols receive deposits of different materials including heavy metals with each flooding especially when they are irrigated with wastewater. Their evaluation for these metals is of utmost importance. Kobierski (2015) reported contamination of Fluvisols with heavy metals in the surface layer at Vistula River floodplain in the Chełmiński and Nadwiślański areas in Poland. Comprehensive reports on heavy metals contents of soils in Kano urban agricultural lands have been reported by Dawaki *et al.*, 2013; Mashi and Alhassan, 2007.

With about nine (9) tanning industries in Sharada Industrial Area (FME, 2001), where this study was carried out, coupled with the continuous use of the discharged tannery effluents into surrounding rivers from the industries, including the river under study, heavy metals are bound to be deposited in the soils which may affect crop production. This is further aggravated by the rapid expansion of Kano city leading to urban and peri-urban agriculture which most often necessitates the use of contaminated

wastewater for irrigation purposes. It is imperative to study the levels of these heavy metals in both soils and irrigation water along the *Tatsewarki River* so as to assess their level of contamination as well as the effect of the irrigation water on the soils for increased and sustainable crop production in the study area. The study is an attempt in this direction.

MATERIALS AND METHODS

Description of Study Area

The study was carried out on Fluvisols along *River Tatsewarki* of Kano, Nigeria. The study area is located on latitudes $11^{\circ} 53' 20''$ N to $11^{\circ} 56' 40''$ N and longitudes $8^{\circ} 30' 0''$ E to $8^{\circ} 33' 20''$ E (Figures 1 and 2). *River Tatsewarki* is one of the main drains in the southern part of Kano draining into river Challawa and conveys it to *River Challawa* (Bichi and Anyata, 1999). The area has an average elevation of 412 m above sea level, and the total length of this river has been estimated to be about 18.6 km; covering a longitudinal distance of about 12.3 km.

The study area is located within the savannah agro-ecological zone of West Africa. The climate is classified as tropical savannah. The area falls within the northern part of the Kano region with four distinct seasons namely; the dry and cool season, the dry and hot season, the wet and warm season (rainy season), and the dry and warm season as described by Olofin (1987). The mean annual rainfall is about 600 mm to 750 mm received between May and October in a normal year. The mean annual temperature ranges from 23°C to about 26°C in the coolest season and 27°C to 33°C in the hottest season.

The dominant land use in the area includes residential, commercial, industrial, urban and peri-urban agriculture. The major crops grown under rain fed agriculture include maize, sorghum and millet. Some areas are also densely populated with tree crops such as mango and guava.

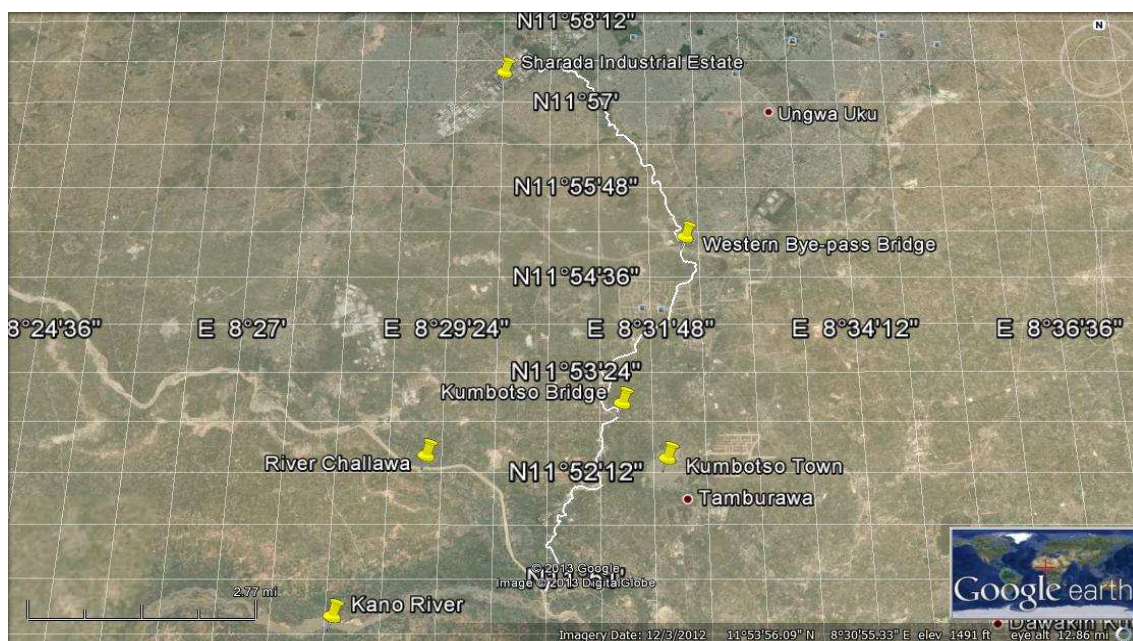


Fig. 1 Google satellite image of the study area outlining the effluent flow path.

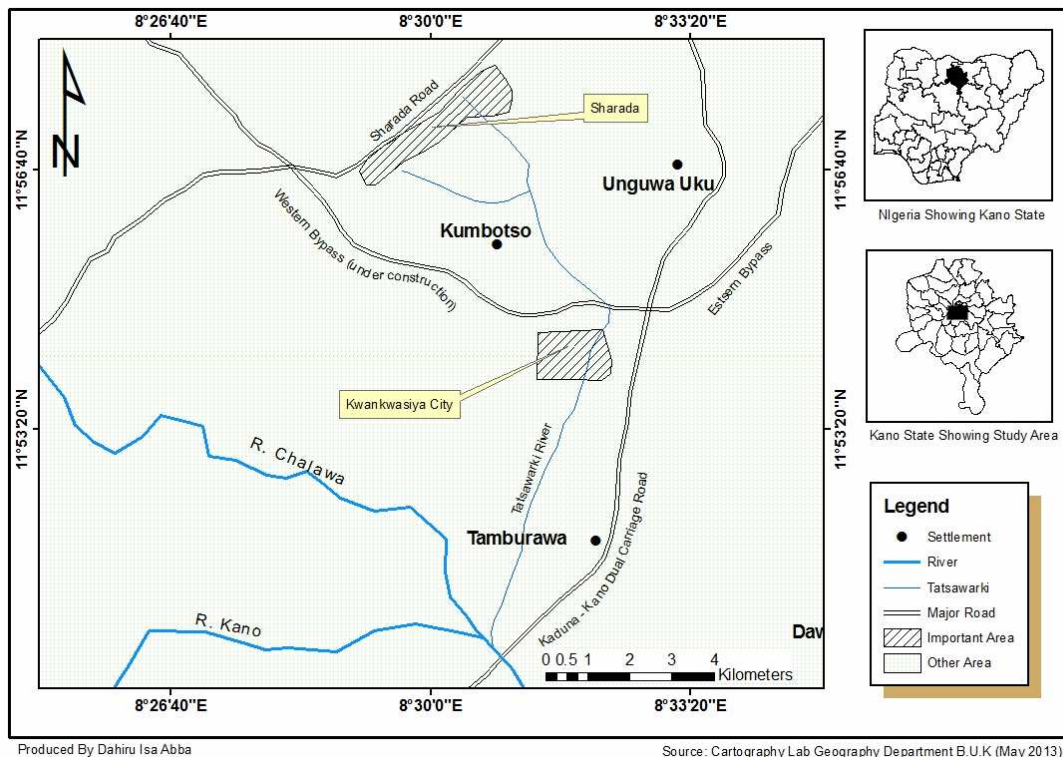


Fig. 2. A sketch map showing the location of Kano State and the study area.

Field Work

A pre-sampling reconnaissance survey of the study area was conducted to identify prospective soil sampling sites and to put logistics in place prior to soil sampling. The total length of the river has been estimated to be about 18.6 km covering a longitudinal distance of about 12.3 km. An area of 1.5 km × 0.5 km (75 ha) in the downstream portion of the river, precisely from where human settlement started, was selected for sampling. Settlements have overtaken most of the up – and mid- stream portions of the river from the source. The Google Earth Satellite Imagery (Figure 1) and Global Positioning System (GPS) were used to demarcate the extent of the area. A sketch map showing the location of Kano State and the study area is presented in Figure 2. The selection was based on the fact that settlements have overtaken most of the up- and mid-stream portions of the river from the

source. The selected sampling site of the river floodplain was divided into three sections namely; up-stream, mid-stream and down-stream. Composite surface soil samples were collected from 0 - 20 cm using soil auger for analysis. Similarly, composite soil samples were collected outside the floodplain to serve as the absolute control. Irrigation water samples were collected from each section of the river at the point of discharge into the irrigated farms. Similarly, control irrigation water samples were collected from tube wells outside the farms being irrigated with the wastewater (Fig. 3) in each section of the river. Collection of the water samples was done after the water was pumped for about 30 minutes to make sure that the water was from the aquifer. The water samples were collected in 2-litre plastic containers provided with a cap. The water samples were refrigerated at ≤ 4°C before analysis.

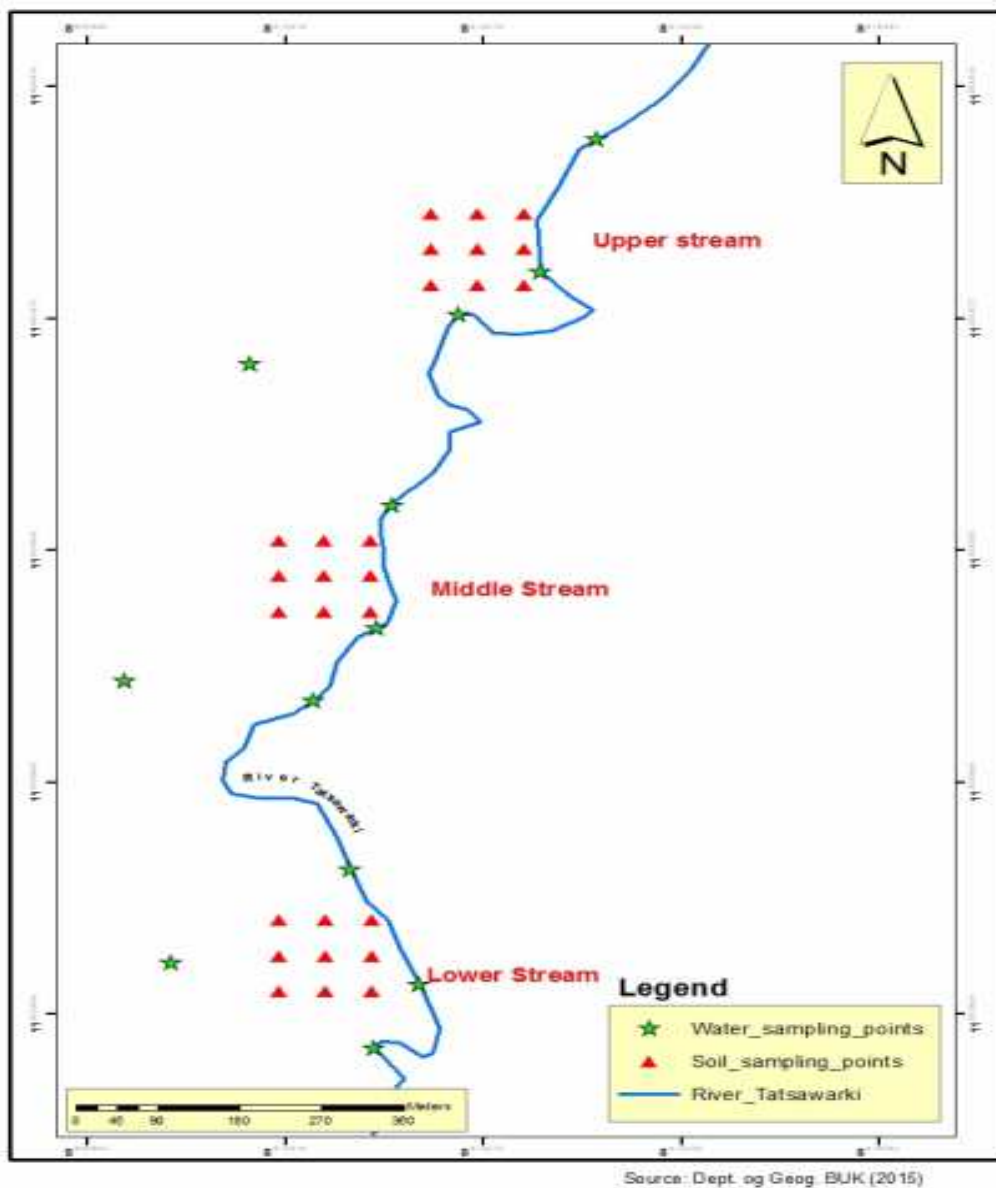


Fig.3. A sketch map showing soils and water sampling points

ANALYTICAL METHODS

In each soil sample, total concentrations of Cr, Pb, Cd, Cu, and Zn were determined using the procedure described by Sabudak *et al.*; (2007). The concentrations were read on Atomic Absorption Spectrophotometer (AAS) as described by Floyd and Hezekiah (1997). The irrigation water samples were also analyzed for Cr, Pb, Cd, Cu, and Zn and subsequently digested and the concentrations were also read on AAS (Floyd and Hezekiah, 1997).

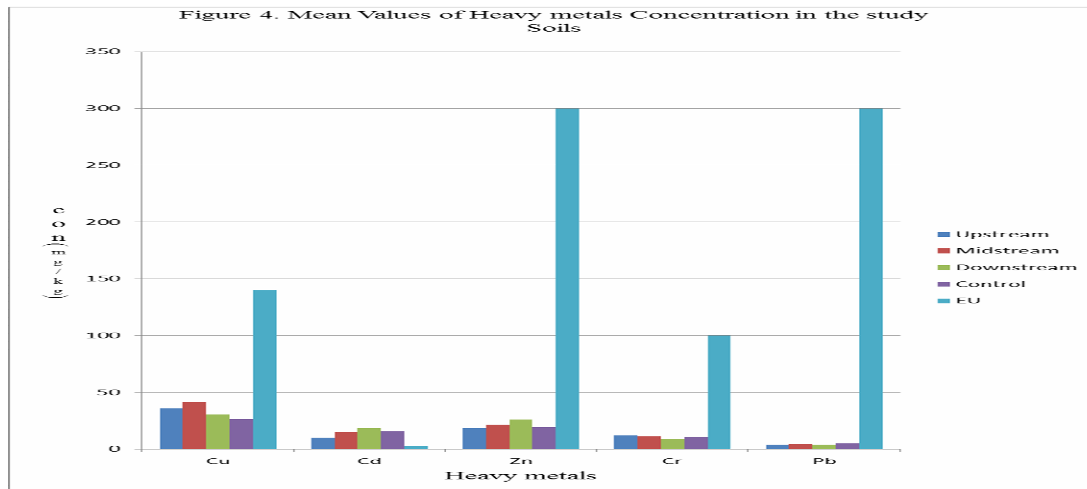
RESULTS AND DISCUSSION

The mean values of the heavy metals concentration in wastewater irrigated Fluvisols in sections of river Tasewarki in Kano is shown in Figure 4. Copper concentrations varied significantly ($P < 0.01$) among the sections. The distribution did not follow a particular pattern. The concentrations ranged from 30.556 (downstream) to 41.667 mg/kg (midstream). Mean values from all the sections were higher than the control (26.667 mg/kg). This clearly indicates that

Cu was deposited in the farms sampled for analysis particularly at the midstream. This agrees with the finding of Rajinder (2012) and Nguyen *et al.*, (2008) in which they reported higher concentration of Cu, Pb and zinc in wastewater irrigated soils as compared with control soils. This finding corroborates the report of Dawaki *et al.*, (2013) that concentrations of heavy metals are higher in soils receiving industrial and domestic sewages in soils of Kano urban agricultural lands. It is also in conformity with the findings of Mashi and Alhassan (2007) that soils in areas receiving industrial wastes recorded highest concentrations of heavy metals in fadama soils in Kano city, Nigeria. Dawaki *et al.* (2013), however reported very much lower ($4.95 - 5.99 \text{ mgkg}^{-1}$) concentrations of Cu in similar soils in Kano. This is because, the concentrations of these trace metal in wastewater is seem to be higher than the natural river water.

Copper is a low mobility metal in near-neutral soils. In more alkaline soils, soluble complexes of Cu^{2+} can form and increase the total copper solubility. Cadmium varied ($P < 0.05$) among the sections and ranged from 10.26 (upstream) to 18.687 mg/kg (downstream). It increased downslope (Figure 4). Generally there was no specific pattern of variation in

the Cd concentration between the soils and the control among the sections (Figure 4). This could be attributed to the chemical and physical nature of the soil. It has been reported by Sherene (2010) that addition of phosphatic fertilizers to the soil can lead to an increase in Cd concentration.

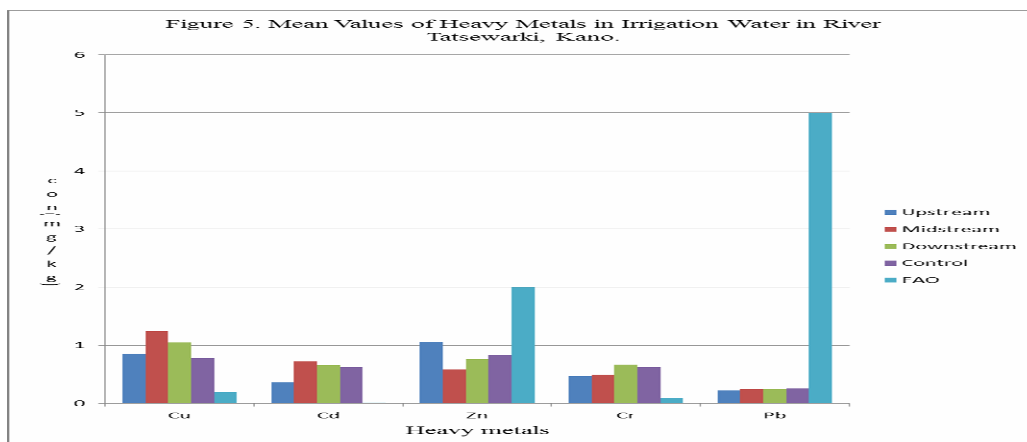


However, the concentration of Cd was only higher than the control (15.657 mg/kg) at the downstream section of the river. This is contrary to the findings of Mustapha and Umar (2013) who reported significantly higher amounts of cadmium at all wastewater irrigated areas as compared with the absolute control. The concentration of Zn showed significant ($P < 0.05$) among the sections (Figure 4) ranging from 18.687 (upstream) to 26.010 mg/kg (downstream) indicating higher accumulation of Zn downslope. The mean values at all the sections were higher than the control (19.687 mg/kg) except at the upstream even though the two mean values were statistically at par with each other. Once again this may be attributed to the chemical and physical nature of the soil.

There were no significant differences between Cr and Pb concentrations in all the sections. The concentrations of all the heavy metals studied were far less than the EU standard above which crop production may be impaired (Figure 4). This means

that even though the heavy metals studied generally accumulated in the soils, their concentrations did not reach toxic levels as to negatively affect crop production in the study area.

The mean values of heavy metals in the irrigation water are presented in Figure 5. Among the heavy metals evaluated, only Zn varied ($P < 0.05$) across the sections. Zinc concentration was higher at the upstream (1.06 mg/l). This is expected due to the fact that the upstream is very much closer to the source of the wastewater discharge from the tanning industry than the other sections of the river. The mean value (1.06 mg/l) was higher than the absolute control (0.83 mg/l). This indicates that the irrigation water was really contaminated with Zn. Concentration of Zn was observed to be higher (0.76 mg/l) at the downstream than at the midstream (0.58 mg/l). This may not be unconnected with downwards movement of water containing Zn. Generally, the concentration of Cr, Cu, Cd and Pb increased downslope.



The maximum concentration of heavy metals in irrigation water for it to be safe for use has been reported by FAO (1992) and this maximum limits for the heavy metals are presented in Figure 5. Concentrations of Cu, Cd and Cr across the sections were higher than the allowable FAO (1992) limits. This implies that the wastewater is not safe for use as irrigation since it can lead to deposition of Cu, Cd and Cr in the soils which negatively affects the soil quality as well as crop production in the study area. However, Zn and particularly Pb were within the allowable limits (Figure 5). This means that the irrigation water did not contain toxic concentrations of Zn and Pb and therefore considered safe for use as far as these metals are concerned.

The correlation coefficients (r) of heavy metals in the soils and in the irrigation water are presented in Table 1. The relationship between Cr, Cu, Cd and Pb in the irrigation water and in the irrigated soils was positive

Table 1. Pearson's correlation of Water Heavy metals with Soil Heavy Metal

	Cr _s	Cu _s	Zn _s	Cd _s	Pb _s
Cr _w	0.019	0.359	0.272	0.491	-0.354
Cu _w	-0.230	0.151	-0.133	-0.144	-0.208
Zn _w	0.445	-0.079	-0.214	0.034	0.315
Cd _w	-0.077	0.296	0.061	0.190	0.104
Pb _w	0.486	-0.006	-0.031	0.062	0.202

NB: w = water s = soil

CONCLUSION

The study shows that even though the heavy metals studied generally accumulated in the soils, their concentrations did not reach toxic levels. Results also indicated that the irrigation water contained toxic levels of Cu, Cd and Cr and is therefore considered unsafe for use as irrigation water as far as these heavy metals are concerned. However, it did not contain toxic levels of Zn and Pb. The irrigation water was responsible for the accumulation of all the heavy metals in the soil except Zn.

RECOMMENDATIONS

It is recommended that periodic appraisal of the soils should be embarked upon so as to monitor the toxicity level of the soil to keep it within the present

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but non – significant with r values of 0.019, 0.151, 0.190 and 0.202 respectively. This means that irrigation with the wastewater was responsible for the accumulation of Cr, Cu, Cd and Pb in the irrigated Fluvisols. This finding is in conformity with the reports of Sahu *et al.*, (2007), Nguyen *et al.*, (2008), Kibria, *et al.*, (2012). However, the irrigation water was not responsible for the accumulation of Zn in the soil (r = - 0.214). This is probably associated with low concentration and variability of Zn in the irrigation water as compared with the FAO (1992) maximum limits (Figure 5). It may also be due to other factors as reported by Haghghet and Asadi (2015) that the concentration of heavy metals in soils during irrigation with wastewater depends upon a number of factors such as the element's concentration in wastewater, the period of wastewater irrigation, soil structure, acidity as well as the percentage of soil organic materials.

non-toxic level where applicable. It is also recommended that the wastewater should be remediated to minimize and control the heavy metal contamination through phytoremediation. These measures are expected to enhance increased and sustainable crop production in the area studied.

CONTRIBUTION OF AUTHORS

The corresponding author is responsible for the scholarly write up of the paper while the co-author was involved in the collection and collation of data from the field.

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