

EFFECTS OF DIFFERENT COAGULANTS ON THE PHYSICO-CHEMICAL, MICROBIAL AND SENSORY PROPERTIES OF WARA, A NIGERIAN SOFT SOY-CHEESE

James^{1*}, S., Nwokocha², L., Tsebam¹, B.C., Amuga³, S.J., Ibrahim¹, A.B., and Audu¹, Y.

¹Department of Food Science and Technology, Federal University of Technology,
PMB 65, Minna, Niger State, Nigeria

²Department of Hospitality and Tourism Management, Delta State Polytechnic,
Ogwashi-Ukwu, Delta State, Nigeria

³Federal University Kashere, Gombe State, Nigeria

*Corresponding author's email: samaila.james@futminna.edu.ng

ABSTRACT

Nigerian soft cheese often referred to as wara was produced from soya bean milk using coagulants from lime, alum and steep water. Effects of these coagulants on the yield, proximate composition, functional properties, microbial loads and sensory properties were evaluated. The result revealed that different coagulants used had no significant ($p \geq 0.05$) effect on yield of the cheeses. The result for the proximate composition shows that, different coagulants used influenced all the proximate parameters except the protein content. Similarly, coagulants used significantly affected the water absorption capacity, oil absorption capacity, foam capacity and gelation capacity. However, the bulk density was not significantly ($p \geq 0.05$) affected. Steep water coagulated cheese had the highest total plate count. The result for sensory properties shows that, lime gave the best cheese in terms of general acceptability. Cheese produced from steep water as a coagulant gave the best proximate attributes, however, its high total plate count calls for concern.

Key words: local cheese, proximate, microbial, sensory and functional

INTRODUCTION

Meat analogues are food products that are made to have similar texture, color, taste and form as meat (Hurley and Liebman, 2006). They are considered to be meat substitutes or meat alternatives because, they provide good source of protein and can also be used as bulking agent to extend real meat products. Meat analogues usually contain flavor, spices and wheat gluten and are rich in fiber because they are made from plants (Klausner, 2002). *Wara* (a meat analogue) is an unripe cheese consumed in several parts of West Africa. It is often referred to as poor man's cheese and also known as 'tofu', 'beske', or 'wara' in Nigerian. It can serve as a snack which is easy to prepare and often eaten as a main meal.

Locally, meat analogue, *wara* is made from soybeans (*Glycine max*). Soybean is a leguminous vegetable belonging to the pea family. It is grown in the tropical, sub-tropical and temperate climates and it is known to have great nutritional value. Its importance ranges from milk production, oil processing, livestock feeds, industrial uses and human consumption (Iwe, 2003).

Soybeans have been recognized to be an ideal grain for meeting protein requirement for both man and animal. It is a source of good fat (unsaturated) unlike saturated fat from animal origin hence, good for heart health (Iwe, 2003; Samuel and George, 2009). Soybean is a cheap source of quality protein that balances the essential amino acids profile of cereal flour and it has a close protein content and amino acids to cow's milk (Belewu and Belewu, 2007). It is one of the best vegetarian food items as far as protein content is concerned. Whole soybean contains 40% protein, 30% carbohydrate, 20% lipid and 10% mineral (Iwe, 2003; Samuel and George, 2009). The medicinal nature of soybean is extremely essential in building body immune system. Soybean foods have significant protection against such health challenges as heart disease, diabetes, high blood pressure, stroke, menopause, ulcer, and cancer (Singh *et al.*, 1999; Fabiyi, 2006).

Wara processing involves the use of rudimentary equipment and largely a home art method. The process is not standardized and usually done under unhygienic conditions. In some communities, starter culture recovered from

previous production is kept for the next production; while others make use of lemon juice. These variations account for non-uniformity of the product in terms of nutrients, texture and acceptability. This work attempts to study the effects of different coagulants on the quality and sensory acceptability of the product.

MATERIALS AND METHODS

Sources of Materials

The soybeans, alum, lime and steep water were purchased from Kure Ultra-Modern Market, Minna, Nigeria.

Sample Preparation

Soybean seeds were cleaned to remove contaminants. The seeds were soaked in clean water (for 2 h.) to soften them, washed and then milled into a paste. The paste was diluted with water at ratio 1:4 and sieved through muslin cloth to extract the soy milk.

Production of Local Cheese

The milk obtained was boiled in a stainless steel pot for 10 min. after which the coagulant (lime, steep water or alum) was added to the boiling soy milk until the proteins clot at their isoelectric point. Coagulated cheese was then pressed in a muslin cloth to remove residual whey, cut into uniform cubes of 3 cm and deep-fried in a boiling vegetable oil to doneness (7-8 min). Local cheese from each coagulant was refrigerated (4°C) until analysis.

METHODS

Determination of Moisture Content

Moisture content of the samples was determined according to AOAC (1995). A porcelain crucible was washed and dried in a hot air oven for 30 min. at 105°C. It was then cooled in a desiccator for another 30 min. The crucible was then weighed and 2 g of the sample was poured into the crucible dish and recorded as W_1 and W_2 respectively. The crucible and the content were placed in an oven at 105°C for 3 h. It was then removed, cooled in the desiccator for 30 min and weighed recorded as W_3 .

$$\% \text{ Moisture Content} = \frac{W_3 - W_2}{W_2} \times 100$$

Determination of Fat

The fat content of the samples was determined according to AOAC (1995). A 2-g sample was carefully transferred into a thimble. The thimble was blocked with cotton wool and the extraction was carried out continuously for 8 h using petroleum ether (boiling point 60°C). The solvent was evaporated using water bath and the remaining extract was dried at 105°C for 60 min in an oven after which it was placed in desiccator to cool. The flask was weighed again and % fat calculated thus:

$$\text{Fat} = \frac{\text{Weight of extracted fat}}{\text{Weight of sample}} \times 100$$

Determination of Ash Content

The ash content of the samples was determined according to AOAC (1995). The weight of crucible dish was taken and 2 g of the sample was added to the crucible and placed in a muffle furnace rack and the temperature was set to 500°C for 16 h until there was complete ash. The ash in the crucible dish was removed and kept in desiccator to cool before weighing. Percentage ash was calculated as:

$$\% \text{ Ash} = \frac{\text{Total weight of extracted ash}}{\text{Weight of sample}} \times 100$$

Determination of Crude Fibre Content

The crude fibre of the samples was determined by the procedure outlined in AOAC (1990). Two g of the sample was weighed into 500 ml beaker and boiled in 200 ml HCl (10% V/V) for 30 min. The suspension was filtered and the residue was washed vigorously with distilled water until it was no longer acidic. It was then boiled in 200 ml 1.25 M NaOH for 30 min. filtered through Whatman filter paper (No. 1) and then washed with distilled water. The residue obtained was transferred into a pre weighed crucible in hot air oven for 30 min., then cooled in desiccator and reweighed.

$$\% \text{ Crude fibre} = \frac{W_2 - W_3}{W_1} \times 100$$

Determination of Protein Content

The protein content was determined by AOAC (1995). Half (0.5) g of sample was weighed into 500 ml Kjeldahl flask. One tablet of catalyst (Selenium) and 20 ml of 25% concentration of sulphuric acid (H_2SO_4) was added and the flask was fixed into Kjeldahl digestion plate. Digestion lasted for 6 h and the liquid was clear and free from brown or black coloration. The digested mixture was allowed to cool and made up to 100 ml in a conical flask. Two drops of indicator (2% methyl red) was added and placed under the collection spigot of the distillation apparatus. 10 ml of the digester was pipetted into stopper portion of the condenser and 10 ml of 40% sodium hydroxide solution was added the solution was allowed to distil for 15 min. or when the volume of ammonia collected in boric acid in the receiver flask was 50 ml and when the red solution had turned blue, the distillate was then titrated against 0.1 M hydrochloric acid (HCl) to a pinkish colour. The protein was calculated as:

$$\% \text{ Gramme Nitrogen} = \frac{T \times 0.014 \times M_{HCl} \times DF}{\text{Weight of sample}} \times 100$$

where T is titre value and DF is dilution factor. Multiplying the value of % Gramme Nitrogen by 6.25 gives % Crude Protein.

Determination of Carbohydrate Contents

Carbohydrate contents of the cheese samples were determined by difference:

$$\% \text{ Carbohydrate} = 100 - (\% \text{ moisture} + \% \text{ ash} + \% \text{ protein} + \% \text{ crude fibre} + \% \text{ fat})$$

Functional Properties

Bulk density

The procedure of Adebawale *et al.* (2005) was used. A specified quantity of the sample was put into a weighed 5ml measuring cylinder (W_1). For packed density, it was gently tapped to eliminate air spaces between the sample in the cylinder and the volume was noted as the volume of the sample used. The new mass of the sample and the cylinder was recorded (W_2). The bulk density was expressed as: $BD = W_2 - W_1$. For loose bulk density (LBD) space was not eliminated by trapping.

Water absorption capacity

Water absorption was determined using the method outlined by Adebawale *et al.* (2005). One gram of sample was weighed in pre-weighed 15 ml centrifuge tubes and 10 ml distilled water was added. After the mixture was wetted samples were allowed to stand at room temperature for 30 min. with occasional stirring. The mixture was centrifuged for 25 min. at 3000 rpm. The resultant supernatant was decanted and the centrifuge tube containing sediment was weighed.

Foaming properties

Foaming capacity was measured by mixing 1 g of sample with 50 ml of distilled water in a laboratory blender and transferred to 250 ml graduated cylinder. The foam formed was indicated as the foaming capacity (ml/100 ml sample). The final observation was made after 60 min. for determining the foam stability (Onwuka, 2005).

Emulsion properties

Emulsifying property was measured as described by Yatsumatsu *et al.* (1992) by mixing 0.5 g of the sample with 5 ml of distilled water followed by addition of 5ml of oil and vigorously shaken for 5 min. by magnetic stirrer. The resulting emulsion was centrifuged at 2000 g for 30 min. The volume of the emulsified layer divided by whole slurry and multiplied by 100 resulted in the emulsifying activity (ml/100ml of sample).

Gelatinization capacity

The gelation capacity was determined using the modified method by Kulkarin *et al.* (1991). A suspension of the sample was made by mixing it with distilled water at ratio 1:4 (W/V) in a 100 ml laboratory test tube. The suspension was heated for 1h in a boiling water bath followed by rapid cooling under running tap water. The test tube was

further cooled for 2 h at 40°C. The test tubes were then inverted one after the other. The least gelation capacity was taken as the concentration when the sample from the inverted test tube did not fall or slip.

Microbiological Analysis

The microbial load of the samples was determined according to AOAC (1990). Nutrient agar (28 g) was dissolved in one liter distilled water in a conical flask. The dissolved agar was autoclaved at 121°C for 15 min. One gram of the sample was weighed and dissolved in a test tube and serial dilution was carried out. One ml of the dilution was pour plated and incubated for 24 h at 37°C. The colonies were counted and recorded as colony forming unit/g.

Sensory Evaluation

The sensory evaluation of the cheese samples was carried out by 20 untrained panelists from the Department of Food Science and Nutrition, FUT, Minna. Hedonic scale was used with 1 representing 'extremely like' and 9 'extremely dislike'. The panelists were presented with the coded samples and were asked to judge the samples on the basis of appearance, aroma, taste, mouth feel and overall acceptability. The assessors were instructed on the basic taste panel procedures and to make their own individual judgment. They were equally instructed to take a sip of water and pause for a few seconds before tasting each sample and to re-taste if not sure of their decisions.

STATISTICAL ANALYSIS

Data obtained were analysed using analysis of variance (Steel and Torrie, 1980). Where differences between mean values existed, they were separated using the least significant difference test. Significance was accepted at 5% probability level.

RESULTS AND DISCUSSION

Proximate Composition of the Cheese

The isoelectric point (pH) of protein rich food is an important step in the production of cheese. The final quality of cheese can be affected by the type of coagulum used during the coagulation process. The results of the yield and proximate composition show that the samples significantly ($p \leq 0.05$) differed in all parameters measured except in yield and protein content (Table 1). There was no significant ($p \geq 0.05$) difference in the yield which implies that the various coagulants considered did not differ considerably in their coagulating ability. Moisture content of steep water coagulated cheese was significantly ($p \leq 0.05$) higher than lime and alum coagulated cheeses. These results contradict Omotosho *et al.* (2011) who reported that cheeses from cow milk had no significant moisture difference in all coagulants used. In such perishable food product, high moisture content is not preferred

Table 1: Yield and proximate composition of the cheeses

Yield (%)/ Proximate composition (%)	A	B	C
Yield	24.20±0.01	23.52±0.11	23.92±0.02
Moisture	25.22 ^a ±1.40	26.68 ^b ±0.34	28.43 ^a ±0.02
Ash	3.21 ^a ±0.02	2.63 ^b ±0.08	2.74 ^b ±0.10
Protein	40.00±1.21	41.30±0.00	41.40±0.41
Crude fibre	2.48 ^c ±0.10	2.94 ^b ±0.20	3.12 ^a ±0.10
Fat	38.85 ^a ±0.15	33.85 ^b ±0.15	33.85 ^b ±0.26
Carbohydrate	8.09 ^b ±1.43	13.01 ^a ±2.65	11.25 ^{ab} ±0.50

Values are means of triplicate readings. Values on the same row followed by the same superscript are not significantly different.

A - Lime coagulated cheese, B - Alum coagulated cheese, C - Steep water coagulated cheese

Table 2: Functional properties of the cheese samples

Functional properties (%)	A	B	C
Water absorption capacity	1.25 ^a ±0.05	1.17 ^b ±0.01	1.13 ^b ±0.20
Oil absorption capacity	1.31 ^a ±0.10	1.08 ^b ±0.30	1.35 ^a ±0.13
Emulsion capacity	41.46 ^b ±0.50	42.50 ^a ±0.10	39.47 ^c ±0.74
Foaming capacity	6.00 ^b ±1.00	8.00 ^a ±0.20	4.00 ^c ±0.30
Bulk density	0.59±0.14	0.58±0.02	0.59±0.01
Gelation capacity	Gel	Gel	Gel

Table 3: Microbial load of the cheese samples

Sample	Total Plate Count (CFU/g)
A	7.7 x 10 ⁴
B	8.4 x 10 ⁴
C	1.2 x 10 ⁵

A - Lime coagulated cheese, B - Alum coagulated cheese, C - Steep water coagulated cheese

Table 4: Sensory properties of the cheese

Sensory attribute	A	B	C
Texture	7.00 ^a ±0.97	6.30 ^{ab} ±1.34	6.15 ^b ±1.34
Flavour	6.05±1.73	5.85±2.00	5.70±1.75
Taste	6.30±1.59	5.65±1.87	5.90±1.99
Appearance	6.40±1.27	6.45±1.39	6.15±1.18
General acceptability	6.85±1.03	6.30±1.62	6.45±1.79

Values are means of triplicate readings. Values on the same row followed by the same superscript are not significantly different.

A - Lime coagulated cheese, B - Alum coagulated cheese, C - Steep water coagulated cheese

because it favours the growth and proliferation of microorganisms, thus reducing the shelf life. The moisture content (25.22 to 28.43%) obtained in this study is low compared with 38.10%, 37.45% and 49.07 to 49.71% for PM fruit liquor supplemented gouda cheese, CO fruit liquor supplemented gouda cheese and Mozzarella cheese, respectively (Andreatta *et al.*, 2009; Choi *et al.*, 2015). The ash content is a measure of mineral elements in a food. The ash content of lime coagulated cheese was significantly high ($p \leq 0.05$) than alum and steep water coagulated cheeses. This result agrees with

Choi *et al.* (2015) and Song *et al.* (1997) who reported a significant ($p < 0.05$) increase in the ash content of gouda cheese supplemented with PM fruit liquor over the control. The ash content in this study (2.63 to 3.21%) agrees with 2.57 to 3.95% for six Mexican cheeses (Caro *et al.*, 2014). Similar trend was observed in the crude fibre content. Lime coagulated cheese significantly ($p < 0.05$) had the highest fat content, while steep water and alum coagulated cheeses had the least values. The significantly ($p < 0.05$) higher fat content of lime coagulated cheese agrees with Choi *et al.* (2015) who reported an increase in the fat content of cheese supplemented with CO fruit liquor over the control. The result obtained in this study (38.85%) is high compared with 31.22 to 33.52% and 18.80 to 31.80% for gouda cheeses supplemented with two fruit liquors and six Mexican cheeses, respectively (Caro *et al.*, 2014; Choi *et al.*, 2015). The carbohydrate content ranged from 8.09 to 13.01% with alum having the highest value (13.01%). Similar result was reported by Choi *et al.* (2015) in gouda cheeses.

Functional Properties of the Cheese

Different coagulants significantly ($p \leq 0.05$) affected all the functional properties (WAC, FC, EC, GC) studied except the bulk density (Table 4). The water absorption capacity of lime coagulated cheese was significantly ($p \leq 0.05$) higher than alum and steep water coagulated cheeses. This could be attributed to the fact that lime juice did not interfere with the hydrophilic domains of the amino acids thereby giving the cheese high water binding capacity (James and Nwabueze, 2014; James *et al.*, 2016). The oil absorption capacity of steep water coagulated cheese was significantly ($p \leq 0.05$) higher than lime and alum coagulated cheeses. This implies that, possibly, steep water favoured the hydrophobic domains of the protein giving the cheese characteristics high oil absorption capacity. However, the emulsion capacity of alum coagulated cheese was significantly ($p \leq 0.05$) high than lime and steep water cheeses. This could be as a result of the presence of residual salt (alum) in the product which increased the solubility of active protein. Furthermore, foaming capacity and stability depend on the surface of active properties of protein involved (Iwe, 2000).

Microbial Load of the Cheese

Steep water coagulated cheese had the highest total microbial count followed by coagulated cheese from alum, the lime coagulated cheese had the lowest microbial count (Table 3). The high count observed in steep water coagulated cheese could be attributed to the fact that steep water itself contains a diverse microbial flora due to natural fermentation. By contrast, lime juice and alum are low in natural microbial load.

Sensory Attribute of the Cheese

There was no significant ($p \geq 0.05$) difference in the sensory parameters studied except in texture (Table 4). Lime coagulated cheese was significantly ($p \leq 0.05$) high in texture than alum and steep water coagulated cheeses.

CONCLUSION

Cheese coagulated from lime gave the highest yield. In proximate composition, the cheeses compared favourably however, lime coagulated cheese had the highest ash and fat content. The sensory attribute -mouth feel of cheese coagulated by steep water was low, with high microbial load.

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