

Orange-Fleshed Sweetpotato Flour: Characterization of Carotenoids and Conservation

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Abstract

Orange-fleshed sweetpotato (OFSP) contains carotenoids that are responsible for its nutritional value and beneficial effects on human health. Orange-fleshed sweetpotato tubers, with their high water content, are very highly perishable under tropical conditions and therefore have a short shelf life. The use of OFSP in local processing industries as a strategy that could stimulate the conservation, marketing and production of OFSP tubers is a promising alternative. The objective of this study was to identify the best conditions for the conservation of OFSP tubers by coupling TLC / MS (Thin Layer Chromatography / Mass Spectrometry) with HPLC (High Performance Liquid Chromatography) - MS (Mass Spectrometry). The molecular structures of carotenoids such as β -carotene, α -cryptoxanthin, β -cryptoxanthin, lutein, and zeaxanthin and possibly other β -carotene (α -carotene, ε -carotene, γ -carotene and lycopene) have been identified in extracts of OFSP. Total carotenoids retention rates were 84% under conditions of 27°C and 26% under conditions of 27°C and 26% relative humidity. Total carotenoids contents (TCC) and total antioxidant contents (TAC) were positively correlated ($R = 0.90$). Total carotenoids contents (TCC) were positively correlated ($R = 0.90$) with total antioxidant contents (TAC) ($P < 0.05$). These results show that under these conditions antioxidant micronutrients which are responsible for its nutritional value could be preserved up to 72% over 6 months of storage. The type of conservation appears to be an alternative in the conservation of OFSP.

Key words : TLC-MS, Folin-Ciocalteu Reagent, antioxidant, phenolics

Farine de Patate Douce à Chair Orange: Caractérisation des Caroténoïdes et Conservation

Résumé

La patate douce à chair orange (PDC) contient des caroténoïdes des points clés et leur nutritivité et de ses effets bénéfiques sur la santé humaine. Les tubercules de patate douce, avec leur teneur élevée en eau, sont très périssables dans les conditions tropicales et ont donc une durée de conservation limitée. L'utilisation de la PDC dans les industries de transformation locale comme stratégie qui pourrait stimuler la conservation, le marketing et la production de tubercules de PDC est une alternative prometteuse. L'objectif de cette étude était d'identifier les meilleures conditions de conservation de la PDC en couplant la chromatographie en couche mince / la spectrométrie de masse (TLC / MS) (Chromatographie en couche mince / Spectrométrie de masse) avec la chromatographie en phase liquide à haute performance (HPLC) - la spectrométrie de masse (HPLC - MS) (Chromatographie en phase liquide à haute performance - Spectrométrie de masse). Les structures moléculaires des caroténoïdes tels que le β -carotène, l' α -cryptoxanthine, le β -cryptoxanthine, le lutéine et le zéaxanthine et éventuellement d'autres β -caroténoïdes (α -carotène, ε -carotène, γ -carotène et leycopène) ont été identifiés dans des extraits de PDC. Les taux de rétention des caroténoïdes totaux ont été de 84% dans des conditions de 27°C et de 26% dans des conditions de 27°C et de 26% d'humidité relative. Les teneurs en caroténoïdes totaux (TCC) et les teneurs en antioxydants totaux (TAC) ont été positivement corrélées ($R = 0,90$). Les teneurs en caroténoïdes totaux (TCC) ont été positivement corrélées ($R = 0,90$) avec les teneurs en antioxydants totaux (TAC) ($P < 0,05$). Ces résultats montrent que dans ces conditions, les micronutriments antioxydants qui sont responsables de sa valeur nutritive pourraient être préservés jusqu'à 72% pendant 6 mois de stockage. Le type de conservation apparaît être une alternative dans la conservation de la PDC.

douce à chair orang eav eœur f ortteueur en eau, s ontrè f ortem epé ris s ablales conditions tropicales et ont donc une courte pé riode de conservation. Ces tubercules contiennent de grandes quantités de caroténoïdes, notamment du β -carotène, qui est une source importante de vitamine A. La transformation de la farine de patate douce en farine de patate douce blanche (OFSP) entraîne une perte de caroténoïdes et de polyphénols. Cette étude a évalué les pertes de caroténoïdes et de polyphénols pendant la transformation de la farine de patate douce en farine de patate douce blanche. Les résultats ont montré que les pertes de caroténoïdes et de polyphénols sont plus élevées dans les conditions de conservation à température ambiante qu'en réfrigérateur ou en congélateur. Ces résultats suggèrent que la conservation de la farine de patate douce blanche doit être optimisée pour minimiser les pertes de nutriments.

Mots clés: CCM-MS, Réactif de Folin-Ciocalteu, antioxydant, composés phénoliques

Introduction

Orange-fleshed sweetpotato (OFSP) contains many chemical substances which guarantee its nutritional value. It is a potential source of vitamin A (Mark et al.,2009; Haskell et al. , 2004), minerals (Fe, Zn, Mn), and other micronutrients such as polyphenols and carotenoids (Haskell et al. 2004). These micronutrients have biological properties of interest for humans, and pharmacological or nutritional properties. However, these phytomicronutrients are not always stable during processes of transformation of the fresh tubers.

Phytomicronutrients undergo various degradations due to the conditions of processing and preservation, thus, altering the nutritional value of the food which contains them. This work identifies molecules of carotenoids (by coupling TLC/MS with TLC -

MS interface CAMAG) extracted from Jewel variety flour and a determination of the levels of carotenoids, phenolics and total antioxidants. Then, losses of these phytomicronutrients were evaluated by spectrophotometry in relation to the time when this flour was kept at ambient (transparent and dark) either in the refrigerator or in the freezer.

Materials and Methods

Identification of the carotenoids found in the extract: The extracted solutions were sprayed using the semi-automatic Linomat 5. HPTLC-ESI-MS spectra were directly recorded using the TLC-MS interface (CAMAG). The mass spectrometer used was a Bruker microTOF-Qde in mode ESI positive.

Storage conditions: 200 g of flour of OFSP

was stored at ambient (laboratory) in transparent containers (SATC), Ambient (laboratory) in dark containers (SADC); stored in refrigerator (SR) and Freezer (SF). 10 mg of this flour were collected, extracted and dosed every two weeks.

Extraction of plant material: extractions were performed by maceration at 4°C with each of the following solvents systems: acetone-water-acetic acid (70: 29.5: 0.5 v/v) (Koala *et al.* 2013a) for assays of total antioxidants (TAC) and phenolics (TPC) contents and acetone-hexane (50: 50 v/v) (Kowalski *et al.* 2000, Koala *et al.* 2013b) for the determination of total carotenoids contents (TCC).

Determination of total antioxidants, phenolics and carotenoids content: these contents were determined spectrophotometrically using a microplate in quartz 96 wells (MP96, SAFAS spectrophotometer) and choosing standards. Thus, total carotenoids were assessed following the slightly modified method described by McMurry *et al.* 2008, Jun *et al.* 2011, and Koala *et al.* 2013. β-carotene, which absorbs at 450 nm was taken as standard. Total carotenoids content were obtained by reporting absorbance of extracts on a standard curve ($y = 25.56x + 0.016$; $R^2 = 0.999$) established using β-carotene. Total

carotenoids content were expressed in mg of β-carotene Equivalents (BCE) per gram of sample. All measurements were carried out in triplicate.

Using Ferric Reducing Antioxidant Power (FRAP) (Benzie *et al.* 1996; Pulido *et al.* 2000; Koala *et al.*, 2013), total antioxidants content which reacted with FRAP reagent to give intense blue color ($\lambda \approx 593$ nm) were determined. Results obtained from the calibration curve equation ($y = 28.67x + 0.066$; $R^2 = 0.999$), were expressed in mg of Trolox Equivalent (TE) per gram of sample.

Total phenolics content were determined using the Folin-Ciocalteu reagent (Nihal *et al.* 2007; Koala *et al.* 2013a). Using Gallic acid as standard, a calibration curve was established. Results determined from the calibration curve equation ($y = 46.41x + 0.063$, $R^2 = 0.998$), were expressed in mg of Gallic Acid Equivalents (GAE) per gram of sample. All measurements were carried out in triplicate.

Data analysis: the experiments were repeated three times and the results are expressed as mean ± SD calculated on the threshold of probability less than or equal to 95%. Analysis of variance (ANOVA) to determine the differences between formulations for their

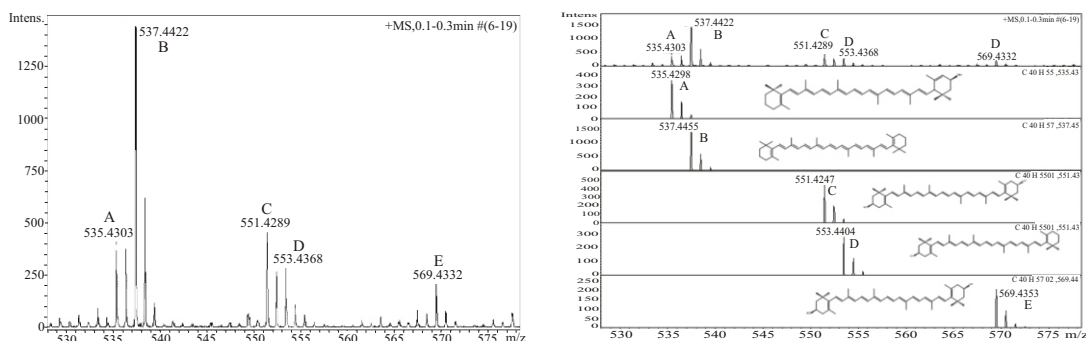


Figure 1: mass spectra of carotenoids

Table 1 : Structures of identified carotenoids






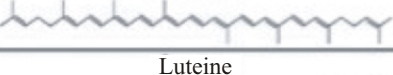
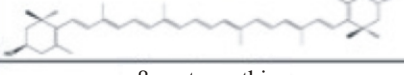
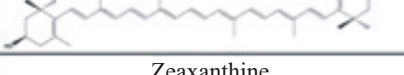
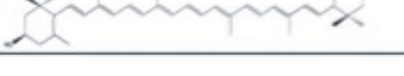
Molecular ion peaks	m/z [M+H]*	Formula	Pseudo molecular Ions	Structures of carotenoid suggested
A	553.4	C ₄₀ H ₅₆ O	535.4 [M-H-18]**	<p>α-crytoxanthine</p> 
B	537.4	C ₄₀ H ₅₆ O	-	<p>β-carotene</p> 
				<p>α-carotene</p> 
				<p>γ-carotene</p> 
				<p>ε-carotene</p> 
				<p>Lycopene</p> 
C	569.4	C ₄₀ H ₅₆ O ₂	551.4 [M-H-18]**	<p>Luteine</p> 
D	553.4	C ₄₀ H ₅₆ O	-	<p>β-crytoxanthin</p> 
E	569.4	C ₄₀ H ₅₆ O ₂	-	<p>Zeaxanthine</p> 

Table 2 : Total antioxidants, phenolics and carotenoids contents remaining after 14 weeks of conservation

Conditions of Storage	SATC	SADC	SR	SF
TCC (%)	27.39	26.10	55.71	84.13
TAC (%)	28.74	45.16	54.56	73.08
TPC (%)	71.42	73.04	75.03	75.48

total antioxidants, phenolics and carotenoids content were done using Genstat, edition 14.

Results and Discussions

Identification of carotenoid compounds from OFSP flour extract Jewel extract carotenoids isolated by the TLC, were identified on the basis of the combined information obtained from Liquid Chromatography and the mass spectrum. The analysis of the Spectra TLC-

MS/MS of the spot 1 ($R_f = 0.48$) with the solvents system hexane/ethyl acetate 96/4 v/v) eluent demonstrates several compounds of carotenoids. On the spectrum of Electrospray mass positive mode (+ ESI), we observed five quasi-molecular ions (A, B, C, D and E) were found at m/z 535,4 [M + H]⁺, m/z 537,4 [M + H]⁺, m/z 551.4 [M + H]⁺, m/z 553,4 [M + H]⁺ and m/z 569,4 [M + H]⁺, suggesting Atomic masses of 534 u 536 u, u 551, 552 u and 568 u, respectively (Figure 1). These Atomic masses are characteristic of the molecular weight of some carotenoids known in the literature. By analyzing mass spectra (Figure 1 and Table 1), it was clear that in addition to ordinary molecular ion peaks, there were the so-called very intense peaks pseudo-molecular ions. Their high intensity reflected their greater stability. These pseudo-molecular ions peak come from fragmentation of molecular ions whose structures are appropriate (De Rosso *et al.*, 2007a ;De Rosso *et al.*, 2007b; AF de Faria *et al.*, 2009).

Conclusion

This study has revealed increase in five molecules of carotenoids when the existent infant food flour is enriched with OFSP flour. Flour formulation containing 30% of OFSP lead to an optimum of total antioxidant content (0.469 mg TE / g), and total carotenoid contents (0.8968 mg BCE/g). This formulation with the best energy value (802.526 ± 5.14 kcal / 200g), can be recommended in combating vitamin A deficiency and protein-energy malnutrition in Burkina Faso.

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